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2 **Seagrass rhizosphere microenvironment alters plant-** 3 **associated microbial community composition**

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18 Originality-Significance Statement:

19 Seagrasses actively select for a distinct microbial community composition at the
20 plant/sediment interface, which may positively affect the nutrient availability in the seagrass
21 rhizosphere owing to enhanced nitrogen fixation by sulphate-reducing, diazotrophic bacteria.
22 We provide evidence of a potential mutualistic beneficial relationship between seagrasses
23 and rhizospheric, heterotrophic diazotrophs based on a mutual exchange of essential
24 nutrients.

25

26

27 MAIN BODY: 4.593 words

28 **SUMMARY**

29 The seagrass rhizosphere harbors dynamic microenvironments, where plant-driven gradients
30 of O₂ and dissolved organic carbon form microhabitats that select for distinct microbial
31 communities. To examine how seagrass-mediated alterations of rhizosphere geochemistry
32 affect microbial communities at the microscale level, we applied 16S rRNA amplicon
33 sequencing of artificial sediments surrounding the meristematic tissues of the seagrass
34 *Zostera muelleri* together with microsensor measurements of the chemical conditions at the
35 basal leaf meristem (BLM). Radial O₂ loss (ROL) from the BLM led to ~300 µm thick oxic
36 microzones, wherein pronounced decreases in H₂S and pH occurred. Significantly higher
37 relative abundances of sulphate-reducing bacteria were observed around the meristematic
38 tissues compared to the bulk sediment, especially around the root apical meristems (RAM;
39 ~57% of sequences). Within oxic microniches, elevated abundances of sulphide-oxidizing
40 bacteria were observed compared to the bulk sediment and around the RAM. However,
41 sulphide oxidisers within the oxic microzone did not enhance sediment detoxification, as rates
42 of H₂S re-oxidation here were similar to those observed in a pre-sterilized root/rhizome
43 environment. Our results provide novel insights into how chemical and microbiological
44 processes in the seagrass rhizosphere modulate plant-microbe interactions potentially
45 affecting seagrass health.

46

47 *Keywords:* biogeochemistry, H₂S, microbes, rhizosphere, ROL, seagrass, sediment, sulphate-
48 reducing bacteria, sulphide-oxidizing bacteria, toxicity.

49

50

51 **Introduction**

52 Seagrass meadows are high-value ecosystems (Costanza et al. 1997) that provide numerous
53 ecosystem services to marine environments, such as facilitating nursery areas for many
54 juvenile fish and crustaceans (Harborne et al. 2006; Larkum et al. 2006), improving water
55 quality through increased sedimentation (Ward et al. 1984; Madsen et al. 2001), providing a
56 main food source for iconic marine animals such as sea turtles and dugongs, and a high ability
57 to sequester carbon in the sediment (Duarte et al. 2005; Fourqurean et al. 2012). Seagrass
58 ecosystems harbour a unique microbiome including populations of microbes attached to
59 seagrass leaves and within the rhizosphere, with microorganisms involved in the sulphur cycle
60 known to play particularly important functional roles (e.g. Devereux, 2005; Jensen et al. 2007;
61 Cúcio et al. 2016; Fahimipour et al. 2017). However, a quantitative understanding of the
62 importance of this microbiome on the fitness of seagrass plants is lacking (York et al. 2016).
63 Jensen et al. (2007) showed that sulphate-reducing bacteria (SRB) and sulphide-oxidizing
64 bacteria (SOB) are located in different microniches within the seagrass rhizosphere, with their
65 distributions largely defined by rhizospheric O₂ availability, and the authors suggested that
66 SOB may be beneficial to seagrasses by removing toxic H₂S from the rhizosphere. However,
67 this hypothesis was derived from bulk sample analysis and did not include measurements of
68 H₂S oxidation within oxic microniches harbouring SOB. To truly tease apart the relationships
69 between seagrasses and communities of SRB and SOB there is thus a need for combined
70 microbiological and biogeochemical characterization of the seagrass rhizosphere
71 microenvironment.

72 Seagrasses grow in largely anoxic, reduced marine sediments (Borum et al. 2005) that are
73 often enriched in the potent phytotoxin H₂S (Holmer et al. 2006; van der Heide et al. 2012;
74 Lamers et al. 2013). These anoxic and sulphidic conditions are a consequence of the
75 deposition of large amounts of organic matter in seagrass meadows driving microbial
76 remineralization processes, which are dominated by SRB (Jørgensen, 1982; Blaabjerg &
77 Finster, 1998; Blaabjerg et al. 1998; Nielsen et al. 2001). To accommodate growth in such
78 hostile environments, seagrasses release O₂ into their rhizosphere, *i.e.*, the small volume of
79 sediment directly influenced by root/rhizome secretions and associated microbes, from their
80 basal leaf meristem and root apical meristems. This so-called radial O₂ loss (ROL), leads to the
81 formation of localised oxic microniches around the seagrass roots/rhizome (Pedersen et al.

82 1998; Jensen et al. 2005; Brodersen et al. 2015a; Koren et al. 2015; Brodersen et al. 2016).
83 These plant-driven oxic microzones support local H₂S re-oxidation and thereby protect the
84 most vulnerable parts of the plants against the H₂S produced by SRB in the surrounding anoxic
85 sediments (Brodersen et al. 2015a). While oxic microniches form mainly around growing root
86 tissues, mature regions of the roots possess physical barriers to ROL, which are typically
87 composed of Casparian-band like structures that may function as an analogous barrier to H₂S
88 intrusion (Barnabas, 1996; Colmer, 2003). Seagrass rhizomes and roots also release significant
89 amounts of labile dissolved organic carbon (DOC), especially around the root-cap (Moriarty
90 et al. 1986; Pollard & Moriarty, 1991), which can stimulate microbial activity including
91 sulphate reduction in the rhizosphere (Isaksen & Finster, 1996; Blaabjerg et al. 1998; Hansen
92 et al. 2000; Nielsen et al. 2001). Seagrasses thus substantially alter the biogeochemical
93 conditions within their rhizosphere, resulting in a dynamic mosaic of chemical microgradients
94 that influence both the distribution and activity of the plant-associated microbial community
95 (Devereux, 2005; Jensen et al. 2007; Brodersen et al. 2015a; Brodersen et al. 2016; Cúcio et
96 al. 2016).

97

98 Seagrass-driven modulation of the rhizosphere microhabitat has been suggested to initiate
99 mutualistic relationships between seagrasses and diazotrophic SRB based on reciprocal
100 nutrient exchange, whereby the seagrass host provides a carbon source to the diazotrophs
101 that in return fix dinitrogen (Hansen et al. 2000; Welsh et al. 2000; Nielsen et al. 2001).
102 Sulphide-oxidizing bacteria within oxic microenvironments in the rhizosphere have also been
103 proposed to play an important role in sediment detoxification (Jensen et al. 2007; Cúcio et al.
104 2016; Fahimipour et al. 2017). Bacterial-mediated H₂S oxidation is 10⁴-10⁵ times faster than
105 spontaneous chemical oxidation (Jørgensen & Revsbech, 1983; Nelson et al. 1986) and given
106 ample O₂ supply, SOB may thus efficiently remove toxic H₂S, facilitating seagrass colonization
107 of highly sulfidic sediments. However, technical constraints in examining the distribution and
108 activity of bacteria at the appropriate microscale resolution and at specific plant-driven
109 chemical micro-hotspots have prevented confirmation of a beneficial role of SOB within the
110 seagrass rhizosphere.

111 Here we present a detailed description of the microbial diversity surrounding the
112 meristematic tissues of *Z. muelleri* in combination with measurements of plant-modulated

113 chemical micro-habitats in the seagrass rhizosphere, with the aim of elucidating the potential
114 importance of SOB in detoxifying sediments for seagrasses, relative to plant-derived
115 spontaneous chemical re-oxidation with O₂.
116

117 **Results and discussion**

118 The seagrass rhizosphere is dominated by bacteria involved in the sediment sulphur cycle
119 (Jensen et al. 2007; Cúcio et al. 2016), but little is known about whether they alleviate or
120 aggravate the exposure of below-ground seagrass biomass to phytotoxic H₂S. This lack of
121 knowledge is particularly significant for the vital meristematic regions of the plants that lack
122 barriers to ROL and are thereby most likely to experience H₂S intrusion (Colmer, 2003).
123 However, this would only occur when ROL is insufficient to re-oxidize sulphide diffusing
124 towards the below-ground tissue surface, as seen during severe events of water-column
125 hypoxia in darkness (Brodersen et al., 2015a). Overall, the ROL-driven oxic microshields are
126 understood as providing better protection of the below-ground tissues as they prevent
127 phytotoxic sulphide from reaching the tissue surface (Koren et al. 2015; Brodersen et al.
128 2015a; Brodersen et al. 2017). Instead, the relatively impermeable mature parts of the below-
129 ground tissues likely improve long-distance, internal transport of O₂ from leaves to distal root-
130 tips (Colmer, 2003; Pedersen et al. 2004; Borum et al. 2006). Here, we combined detailed
131 microscale measurements of the geochemical conditions and dynamics in the seagrass
132 rhizosphere with analysis of the microbial community composition to determine, whether
133 seagrasses benefit from rhizospheric bacterial-mediated sulphide oxidation. We focused on
134 the small volume of sediment surrounding the meristematic regions of the plant, as this part
135 of the seagrass rhizosphere is most significantly affected by the seagrass host and exhibits
136 strong chemical microgradients of O₂ and DOC availability (Moriarty et al. 1986; Pedersen et
137 al. 1998; Jensen et al. 2007; Brodersen et al. 2015a). We were able to detect changes in the
138 microbial community composition as a response to the activity of the seagrass host, without
139 disturbing the below-ground biogeochemical micro-gradients and habitats during
140 microsensor measurements, and determined that SOB are playing an important role in H₂S
141 oxidation in the seagrass rhizosphere.

142

143 *Dynamics of the below-ground chemical microenvironment*

144 Radial O₂ loss (ROL) from the basal leaf meristem led to a ~300 µm thick oxic microzone,
145 protecting seagrasses from intrusion of H₂S through chemical oxidation at the oxic/anoxic
146 interface (Fig. 1; Fig. S1 and S2; Table 1). In light, the O₂ efflux amounted to ~745 nmol O₂ cm⁻¹

147 $^2 \text{ h}^{-1}$ compared to $\sim 322 \text{ nmol O}_2 \text{ cm}^{-2} \text{ h}^{-1}$ in darkness in the microbe-enriched treatment (Table
148 1). In the pre-sterilized environment, ROL also increased more than 2-fold in the light,
149 resulting in an O_2 efflux of $\sim 891 \text{ nmol O}_2 \text{ cm}^{-2} \text{ h}^{-1}$, compared to $\sim 378 \text{ nmol O}_2 \text{ cm}^{-2} \text{ h}^{-1}$ in
150 darkness (Table 1). The enhanced ROL from the below-ground tissue during light stimulation
151 of the leaf canopy resulted in a slightly enhanced thickness of the rhizospheric oxic microzone
152 by $\sim 50 \mu\text{m}$ in both treatments (Fig. 1).

153 At the plant-driven rhizospheric oxic/anoxic interface (*i.e.*, the dotted line on Fig. 1 $\sim 300 \mu\text{m}$
154 from the tissue surface), we found a rapid decrease in sediment H_2S concentrations from ~ 100
155 $\mu\text{mol L}^{-1}$ at $\sim 1 \text{ mm}$ away from the below-ground tissue surface to $0 \mu\text{mol L}^{-1}$ at the oxic/anoxic
156 interface or slightly within the oxic microzone (Fig. 1; Fig. S1 and S2); thus showing complete
157 detoxification of sedimentary H_2S before reaching the tissue surface. The ratio between the
158 O_2 and total sulphide fluxes, *i.e.*, the amount of O_2 molecules released from the below-ground
159 tissue compared to the total amount of sulphide (H_2S , HS^- and S^{2-}) oxidized at the basal leaf
160 meristem in the artificial sediment was 0.6 in darkness and 1.7 in light for both treatments
161 (Table 1). This is indicative of incomplete H_2S oxidation to elemental sulphur (S^0) in the dark
162 (the stoichiometry of sulphide oxidation to S^0 is: $0.5\text{O}_2 \rightarrow \text{S}^0$; Nielsen et al. 2006) and complete
163 H_2S re-oxidation to sulphuric acid (H_2SO_4) in the light (stoichiometry of 2:1). Alternatively, the
164 higher O_2 flux in the light could indicate higher microbial O_2 consumption rates and/or
165 chemical oxidation of other reduced compounds, such as Fe^{2+} , within the rhizosphere. Or that
166 some of the rhizosphere sulphide ($\text{S}_{\text{tot}}^{2-}$) precipitated as FeS during darkness, which would
167 also explain the lower degree of co-existence between O_2 and H_2S in the seagrass rhizosphere
168 in darkness as compared to in light (seen on Fig. 1). This is in accordance with a recent study
169 (Brodersen et al. 2017) that demonstrated Fe^{3+} reduction, and thus Fe^{2+} mobilisation, in the
170 seagrass rhizosphere predominantly during night-time (Brodersen et al. 2017). Moreover, at
171 the seagrass-driven rhizospheric oxic/anoxic interface we also observed a rapid decrease in
172 rhizosphere pH by ~ 2 pH units (Fig. 1; Fig. S1 and S2), most likely as a result of protons
173 generated from the chemical reaction between O_2 and H_2S (Nielsen et al. 2006; Brodersen et
174 al. 2015a; 2016). This resulted in acidification of the rhizospheric oxic microniches reaching
175 pH 4-5 (Fig. 1).

176

177 *Microbial diversity in the seagrass rhizosphere*

178 The microbial community composition in the artificial sediment was similar to that reported
179 from other seagrass-vegetated natural sediments (e.g. Jensen et al. 2007; Cúcio et al. 2016);
180 with many members of the rhizospheric microbial community affiliated with the sulphur cycle
181 (Cúcio et al. 2016; Ettinger et al. 2017; Fahimipour et al. 2017). Compared to the bulk
182 sediment, we observed higher mean relative abundance of SRB taxa including OTUs matching
183 *Desulfovibrio* sp., especially around the root apical meristems, where members of the SRB
184 class *Clostridia* (Devereux, 2005; Sallam & Steinbüchel, 2009) dominated, with ~57% of
185 sequences affiliated with this bacterial class ($t(5)_{\text{RAM-BS}}=4.015$, $p=0.01$; $t(4)_{\text{RAM-NC}}=16.944$,
186 $p<0.001$; Fig. 2). This is notable given that many members of this class are also known to be
187 diazotrophs, i.e., N_2 fixing SRB. It has previously been argued that diazotrophic bacteria (e.g.
188 *Lachnospiraceae* & *Desulfovibrio* sp.) live in a mutualistic relationship around the below-
189 ground biomass of seagrasses (e.g. Welsh, 2000), where the plant provides DOC and
190 diazotrophic SRB reciprocally provide fixed nitrogen to the seagrass. Nielsen *et al.* (2001), for
191 example, showed higher rates of N_2 -fixation and sulphate reduction around and on below-
192 ground tissues of seagrasses as compared to those in the sediment, and that root/rhizome
193 associated SRB fix more dinitrogen than needed for their own growth. In addition, the
194 occurrence of SRB within the rhizosphere may also lead to increased phosphorus
195 solubilisation owing to reduction of insoluble Fe^{3+} oxyhydroxides that can promote the
196 release of previously adsorbed phosphate to the pore-water (Pagès et al. 2011, 2012;
197 Brodersen et al. 2017).

198 Within the plant-derived oxic microzone in the microbe-enriched treatment, sulphide-
199 oxidizing *Epsilonproteobacteria*, including *Arcobacter* sp. and *Sulfurimonas* sp., were detected
200 (Fig. 2). *Epsilonproteobacteria* constituted ~34% of the community at the basal leaf meristem
201 compared to other designated bacterial classes (including *Deltaproteobacteria*, *Clostridia* and
202 *Bacteroidia*), with a relative increase of ~22% as compared to around the root apical
203 meristems (Fig. 2). However, the higher abundance of SOB within plant-derived oxic
204 microniches did not seem to enhance sulphide detoxification, as we observed similar H_2S re-
205 oxidation rates within the pre-sterilized and microbe-enriched treatments (Fig. 1-2; Table 1);
206 which was also shown in calculated rates of total sulphide oxidation (Fig. S3 and S4; Table 1).

207 Spontaneous chemical H₂S re-oxidation via ROL thus seemed of similar magnitude as
208 biological H₂S re-oxidation (Fig. 1; Table 1). However, the mere presence of high populations
209 of SOB indicate that they are playing an important role in H₂S oxidation in the seagrass
210 rhizosphere.

211 Within the seagrass-derived rhizospheric low-pH microniches (corresponding to the oxic
212 microzones; Fig. 1), we observed a slightly lower mean relative abundance of Bacteroidia
213 (~13%) as compared to the root apical meristems and the bulk sediment (>23%) albeit not
214 significant for the porewater enriched sediment ($t(4)_{\text{BLM-RAM}}=-1.141$, $p>0.05$; $t(5)_{\text{BLM-BS}}=-0.461$,
215 $p>0.05$; $t(4)_{\text{BLM-NC}}=-7.894$, $p=0.001$; Fig. 2). The growth rate of *Bacteroidetes* is pH dependent
216 (Thomas et al. 2011), and the potentially impeded growth of these potential plant pathogens
217 (Fig. 2) as a result of reductions in rhizosphere pH driven by ROL from the below-ground
218 tissues (Fig. 1) deserves further attention.

219 Sulphate reduction thus seemed to be the microbial metabolism in the seagrass rhizosphere
220 that was most strongly affected by the activity of the seagrass host, where locally enhanced
221 DOC, especially at the root-tips of seagrasses (Moriarty et al. 1986; Pollard & Moriarty, 1991),
222 could be responsible for the microbial community composition shift (Fig. 2; Fig. S5). The
223 higher relative abundance of the SRB class *Clostridia* at the root apical meristems is supportive
224 of a potential mutualistic relationship between seagrasses and heterotrophic, diazotrophic
225 bacteria based on reciprocal nutrient exchange.

226

227 *Methodology*

228 Our novel experimental approach allowed the combined use of molecular and microsensor
229 techniques, without disturbing the below-ground biogeochemical microenvironment during
230 detailed O₂, H₂S and pH microsensor measurements. Detailed information about the chemical
231 microgradients around the below-ground tissues provided by the microsensor measurements
232 enabled us to sample sediment within only the plant-affected regions of the seagrass
233 rhizosphere. While many modern 'omics' studies are often carried out "blindfolded" by
234 relying on bulk sample analysis, the combination of sampling with microenvironmental

235 analyses, such as described in this study, may alleviate such limitations and guide more
236 hypothesis-driven approaches.

237

238 Some of the limitations of using an artificial sediment matrix mainly relate to it being a
239 simplification of complex natural sediments, where other elements and especially solid-phase
240 species (such as Fe^{3+} oxyhydroxides) also play important roles for reactions affecting
241 distributions of O_2 , sulphide and pH in the seagrass rhizosphere (Brodersen et al. 2017).
242 Insoluble Fe^{3+} oxyhydroxides are e.g. likely to accumulate in the seagrass rhizosphere during
243 day-time as a result of chemical and biological oxidation of dissolved Fe^{2+} and precipitated
244 FeS (Brodersen et al. 2017). At night-time, on the other hand, precipitated Fe^{3+} oxyhydroxides
245 are reduced via sulphide, which re-generates dissolved Fe^{2+} leading to consumption of
246 sulphide through precipitation of FeS (Brodersen et al. 2017). Hence, sulphide and O_2
247 consumption processes are more complex in natural sediments as compared to in artificial
248 sediment. Different processes can also be temporally separated, where H_2S consumption at
249 night is in part a result of ROL-driven Fe^{3+} formation during the day, and O_2 consumption in
250 the day is partially due to re-oxidation of FeS, that precipitated in the seagrass rhizosphere
251 during darkness, to regenerate insoluble Fe^{3+} oxyhydroxides (Brodersen et al. 2017).

252

253 *Conclusions*

254 The novel combination of measuring microscale patterns in microbial diversity and sediment
255 chemical characteristics within an artificial sediment matrix showed that seagrass-mediated
256 alterations of rhizosphere geochemistry result in pronounced shifts of the rhizosphere
257 microbial community composition. ROL from the basal leaf meristem resulted in a marked
258 decrease in sediment pH and H_2S concentrations at the plant-driven oxic/anoxic interface,
259 and within the plant-derived oxic microzones, H_2S was completely re-oxidized, protecting the
260 most vulnerable part of the plant against phytotoxic H_2S intrusion. We observed significantly
261 elevated abundances of SRB in the seagrass rhizosphere, and presence of SOB within oxic
262 microniches. Our high-resolution characterization of rhizosphere chemistry and microbial
263 communities indicate that SOB can play an important role for H_2S oxidation in the seagrass
264 rhizosphere, especially within seagrass-generated oxic microniches.

265

266

267 **Experimental procedures**

268 *Seagrass specimens and sediment sampling*

269 Specimens of *Zostera muelleri* subsp. *capricorni* (Asch.) S.W.L. Jacobs and marine sediment
270 were collected from shallow waters (<2 m depth) in a dense seagrass meadow in Narrabeen
271 Lagoon, NSW, Australia (-33.72°S, 151.29°E). Seagrass sediment was collected with open-
272 barrel, push cores (PVC pipe, 25 cm length, 7 cm internal diameter) as described previously
273 (e.g. Wesley, 2009). After sampling, the sediment and seagrass samples were transported to
274 a greenhouse facility at University of Technology Sydney, where they were kept under natural
275 sunlight in large, aerated and temperature-controlled seawater reservoirs (temperature of
276 ~22°C; salinity of ~34) before further treatments. Prior to experiments, selected seagrass
277 specimens were gently uprooted and washed free of any adhering sediment particles.

278

279 *Experimental setup*

280 To enable the examination of microscale gradients in chemical and microbiological properties
281 near to and at the seagrass below-ground tissue surface, we employed a novel artificial
282 sediment system consisting of a custom-made split flow-chamber, wherein the investigated
283 seagrass specimens were grown in a transparent, artificial sediment matrix (Brodersen et al.
284 2014). Selected seagrass specimens (one plant at a time) were maintained in the flow-
285 chamber with the leaf canopy positioned in the aerated, free-flowing seawater compartment
286 and the below-ground biomass embedded in a reduced, deoxygenated agar matrix in the
287 adjoining “sediment” compartment. A detailed description of the casting procedure and
288 chemical characteristics of the artificial sediment is given below.

289 A water pump submerged into an aerated and temperature-controlled seawater bath
290 (temperature of ~22°C; Salinity of ~34) provided a constant flow (~0.5 cm s⁻¹) of aerated
291 seawater to the water compartment of the flow chamber. Within the sediment compartment,
292 a ~3 cm-thick anoxic and HEPES buffered (10 mM) water layer, residing above the artificial
293 sediment matrix, functioned as a liquid-phase diffusional barrier to O₂ invasion into the
294 artificial sediment (Brodersen et al. 2014). The anoxic water layer was constantly flushed with
295 humidified N₂ throughout the seagrass cultivation period. Below the artificial sediment
296 matrix, pieces of gauze, pre-soaked in an acidic (pH 4) and anoxic 1 mM Na₂S solution, were
297 deployed to ensure a continuous supply of H₂S to the overlaying artificial sediment matrix.

298 Finally, the sediment compartment was covered with aluminum foil to avoid incoming stray
299 light, retain N₂ and thus limit O₂ intrusion into the anoxic water layer.

300

301 Light was provided as a 12h:12h light/dark cycle with a fiber-optic tungsten halogen lamp (KL-
302 2500; Schott GmbH, Mainz, Germany) connected to a timer and equipped with a collimating
303 lens to restrict the illumination to the leaf canopy only. The incident photon scalar irradiance
304 (PAR, 400-700 nm) at the leaf canopy during cultivation was ~150 μmol photons m⁻² s⁻¹. Scalar
305 photon irradiance was measured with a spherical photon irradiance sensor (Walz GmbH,
306 Effeltrich, Germany) connected to a calibrated photon irradiance meter (LI-250A, LiCor,
307 Lincoln, NE, USA).

308

309 *Seagrass cultivation period*

310 Seagrass specimens were generally allowed 2 weeks to recover from sampling and to
311 acclimatize to experimental conditions (temperature and salinity) before cultured in the
312 reduced, transparent artificial sediment. Each experiment ran for a total of 8 days, where (i)
313 pulse amplitude modulated (PAM) measurements were performed on the 4th and the second
314 last day (7th) of the experiment, (ii) microsensors measurements on the 5th to the 7th day of the
315 experiment, and (iii) sediment samples for 16S rRNA amplicon sequencing were taken on the
316 last day (8th) of the experiment, just after the microsensors and PAM measurements. To avoid
317 effects of seasonal changes (such as temperature and nutrient availability) on the sediment
318 microbial community composition, three plant replicates were chosen per treatment (giving
319 a total of 48 experimental days). This is a minor limitation of the methodology, *i.e.*, when
320 combining microsensors profiling with molecular techniques, as determining the chemical
321 microenvironment around below-ground tissues in the casted artificial sediment in high
322 spatio-temporal resolution is a time-consuming process.

323

324 *Artificial sediment matrix in the sediment compartment*

325 To enable identification of potential mutual beneficial relationships between seagrasses and
326 SOB, two treatments were applied, whereby the artificial sediment matrix was either (i)
327 sterilized, including the below-ground biomass surface (negative control), or (ii) enriched with
328 native pore water microbes. Each procedure is explained further in the following paragraphs.

329

330 i) Artificial sediment with pore water microbes

331 The transparent, reduced artificial sediment consisted of a ~0.7% (w/v) deoxygenated
332 agar/seawater solution, buffered with an anoxic solution of HEPES buffer (final concentration
333 of 10 mM; pH ~7), and amended with Na₂S (final H₂S concentration of 500 μM; at pH 7) and
334 pore-water microbes (~50% pore-water in the final 0.7% w/v solution). During casting of the
335 artificial sediment with pore-water microbes, the pore-water was homogenously
336 incorporated into the pre-heated agar/seawater solution (~1.4% w/v) shortly before the
337 artificial sediment matrix was poured into the sediment compartment of the split flow
338 chamber at a matrix temperature of ~38°C. Thereafter, the artificial sediment with microbes
339 was rapidly cooled down to room temperature in the sediment compartment embedding the
340 below-ground tissue of the investigated seagrass specimen (covering the rhizome with
341 artificial sediment to a total depth of ~0.5 cm). The applied pore water was extracted from
342 sediment from the sampling site by means of (i) mild ultrasonication (30 s) in a 50 mL Falcon
343 tube to dissociate microorganisms from the sediment grain surfaces and sediment aggregates
344 (Ramsay, 1984; Lindahl & Bakken, 1995), (ii) centrifugation (2 x 3500g for 5 min at 20°C), and
345 (iii) filtration of supernatants (continuously flushed with N₂ to avoid oxygenation; Millipore®,
346 Polycarbonate membrane filters, 10 μm, USA) to exclude the remaining fine sediment
347 particles.

348

349 ii) Pre-sterilized below-ground environment (negative control)

350 The below-ground biomass of the investigated seagrasses was surface-sterilized by
351 submerging sediment-free roots and rhizomes in a saline, anoxic ~1.05% (w/v) hypochlorite
352 solution for 30 s (Blaabjerg & Finster, 1998) followed by 3 x 1 min rinses in anoxic, filter-
353 sterilized (0.2 μm) seawater. Prior to casting the sterilized artificial sediment, all added
354 solutions and seawater were filter-sterilized (0.2 μm) and the agar solution was heated to
355 120°C in an oven for 30 min. The sterilized artificial sediment matrix consisted of a ~0.7%
356 (w/v) deoxygenated agar/seawater solution, buffered with sterilized, anoxic HEPES (final
357 concentration of 10 mM: pH 7) and amended with Na₂S to a final H₂S concentration of 500
358 μM (at pH 7); resulting in similar chemical properties to the artificial sediment with added
359 pore water microbes as described above. The transparent, artificial sediment matrix applied

360 here permitted the precise and combined application of microsensor measurements and
361 molecular characterisation of microbial communities within specific microzones of interest,
362 *i.e.*, around the basal leaf meristem, root apical meristems and within the bulk sediment.

363

364 *Specimen characteristics and performance*

365 Seagrass specimens with a similar above- to below-ground biomass ratio were selected for
366 this study, to ensure comparable below-ground tissue oxidation capabilities of the
367 investigated specimens (e.g. Frederiksen et al. 2006; Frederiksen & Glud, 2006) (Table S1).
368 The photosynthetic performance of the investigated seagrasses during cultivation was
369 determined as the maximum PSII quantum yield in dark-adapted samples and the effective
370 PSII quantum yield in illuminated samples by means of pulse amplitude modulated (PAM)
371 variable chlorophyll fluorometry (Beer et al. 1998; PocketPAM, equipped with an optical fiber;
372 Gademann Messtechnik GmbH, Germany) (Table S1) to confirm that the seagrasses were
373 generally healthy and photosynthetically active under the experimental conditions (n = 3-7).
374 Following the experiments, the Dry Weight (DW) biomass ratio of the above- to below-ground
375 tissues was obtained after drying each seagrass specimen in an oven at 60°C until a constant
376 weight was reached.

377

378 *Microsensor measurements and flux calculations*

379 We used microsensors to determine the chemical conditions and dynamics at the
380 plant/sediment interface. Clark-type O₂ microsensors (OX-50, Unisense A/S, Aarhus,
381 Denmark; Revsbech, 1989) were used to measure the radial O₂ loss (ROL) from the below-
382 ground tissue of *Zostera muelleri*. The O₂ microsensors were linearly calibrated from signal
383 readings in 100% air saturated seawater and anoxic seawater (obtained by flushing with N₂
384 and adding the O₂ scavenger sodium sulphite) at experimental temperature and salinity. To
385 avoid drifting calibrations during measurements, the O₂ microsensors were pre-contaminated
386 with H₂S before calibrations (Brodersen et al. 2015b). Clark-type H₂S microsensors (H₂S-50,
387 Unisense A/S, Aarhus, Denmark; Jeroschewski et al. 1996; Kühl et al. 1998) were used to
388 measure the H₂S concentration at and around the below-ground tissue of *Zostera muelleri*.
389 The H₂S microsensors were linearly calibrated in acidic (pH 4), anoxic Na₂S solutions of defined
390 H₂S concentrations (0, 50 and 100 µM) at experimental temperature and salinity. pH

391 measurements were performed by means of pH glass microelectrodes (pH-50, Unisense A/S,
392 Aarhus, Denmark; Kühl & Revsbech, 2001) that were used in combination with a reference
393 electrode (REF-RM, Unisense A/S, Aarhus, Denmark) submerged in the split flow chamber to
394 allow the pH microelectrode to develop an electric potential relative to the reference
395 electrode. The pH microelectrodes were linearly calibrated from signal readings in pH buffers
396 (pH 5, 8 and 9) at experimental temperature and salinity.

397

398 Microsensors were mounted on a motorized micromanipulator (MM33-2 & MC-232,
399 Unisense A/S, Aarhus, Denmark) and connected to a microsensor multimeter (Unisense A/S,
400 Aarhus, Denmark) that was interfaced with a PC running dedicated microsensor positioning
401 and data acquisition software (SensorTrace PRO, Unisense A/S, Aarhus, Denmark). The
402 microsensors were carefully positioned at the surface of the basal leaf meristem (defined as
403 0 mm distance from the below-ground tissue on the figures) by manually operating the
404 micromanipulator, while observing the tip of the microsensor relative to the surface of the
405 below-ground tissue through a submerged hand-held lens (described in Brodersen et al. 2014)
406 with a stereo microscope mounted on an articulating arm (SM-6TZ, Amscope, Irvine, CA, USA).
407 All microprofiles were measured in distance increments of 50 μm . Plants were allowed to
408 acclimatize to the experimental conditions for ~ 72 h before microsensor measurements
409 commenced to ensure steady state geochemical conditions. During microsensor profiling, an
410 additional source of N_2 was immersed into the anoxic seawater layer of the sediment
411 compartment (described above) to avoid O_2 intrusion into the layer and loss of H_2S from the
412 artificial sediment due to oxidation when removing the covering aluminium foil (Brodersen et
413 al. 2014; 2015a,b). Three-to-five microsensor measurements were performed in the artificial
414 sediment at the basal leaf meristem and averaged to produce one replicate microprofile for
415 each of the 2-3 investigated seagrass specimens (*i.e.*, $n = 3-5$, technical replicates; $n = 2-3$,
416 biological replicates; which gives a total of 6-15 microsensor profile replicates) in each
417 treatment. Note, that one plant replicate was excluded from the sterilized environment as we
418 could not convincingly determine the below-ground tissue surface during measurements
419 (further described in the supplementary information; Fig. S2).

420

421 **(i) Flux calculations**

422 The radial O₂ loss (ROL) from the below-ground tissue (nmol O₂ cm⁻² h⁻¹) was calculated via a
 423 cylindrical version of Fick's first law of diffusion (Steen-Knudsen, 2002) assuming a
 424 homogenous, cylinder-shaped O₂ flux from the surface:

$$425 \quad J(r)_{BLM} = \varphi D_0 (C_1 - C_2) / r \ln\left(\frac{r_1}{r_2}\right)$$

426 where φ is the porosity of the artificial sediment (here assumed to be similar to seawater); D_0
 427 is the molecular diffusion coefficient of O₂ in seawater at experimental temperature and
 428 salinity; r is the radius of the basal leaf meristem; and C_1 and C_2 are the O₂ concentrations
 429 measured at the radial distances r_1 and r_2 from the tissue surface, respectively. The H₂S
 430 oxidation rates in the immediate rhizosphere (nmol H₂S cm⁻² h⁻¹) were calculated in a similar
 431 manner by correcting D_0 to the molecular diffusion coefficient for H₂S at experimental
 432 temperature and salinity (factor 0.7573; tabulated values are accessible on
 433 www.unisense.com).

434

435 The following equations were used to calculate the total sulphide concentration
 436 microprofiles and fluxes from the measured H₂S concentrations and pH microprofiles
 437 (equations are available at www.unisense.com):

$$438 \quad pK_1 = -98.08 + \frac{5765.4}{T} + 15.04555 \times \text{LN}(T) + (-0.157 \times (S^{0.5})) + 0.0135 \times S$$

439 where S is the salinity and T is the temperature in Kelvin.

$$440 \quad \text{total sulphide } [S_{tot}^{2-}] = [H_2S] \times \left(1 + \frac{K_1}{[H_3O^+]}\right)$$

441 where $[H_3O^+] = [H^+] = 10^{-pH}$ and $K_1 = 10^{-pK_1}$; for pH < 9 (Jeroschewski et al. 1996).

442

443 *Sediment sampling for DNA extractions*

444 Artificial sediment samples were obtained from selected regions of interest, i.e., around the
 445 basal leaf meristem (BLM; at the root/shoot junction), around the root apical meristem (RAM;
 446 at the root-tip), and from the bulk sediment (BS) using a sterilized surgical knife and spatula
 447 (Fig. S6). Samples around the below-ground tissues were carefully collected at a radial
 448 distance of up to ~1 mm from the tissue surface (final volume of ~100 μ L). Only one sediment
 449 sample was acquired from each of the meristem areas (i.e., the BLM and the RAM) of the
 450 three investigated seagrass plants cultured in artificial sediment enriched with native
 451 porewater microbes, as we were only interested in the small volume of sediment affected by

452 the plant ($n = 3$). Whereas several samples (or large sediment volumes) were obtained from
453 the bulk artificial sediment area ($n = 3-4$); taken from both the porewater enriched and pre-
454 sterilized (here used as a microbial negative control) treatments. After sampling, the sediment
455 samples were stored in 2 mL Eppendorf tubes in a -80°C freezer until further analysis. Prior to
456 DNA extraction, four rounds of washing were performed in order to remove the agarose. The
457 artificial sediment samples were first liquefied in a dry bath at $\sim 50^{\circ}\text{C}$ and were then
458 subsequently diluted via centrifugation with 1 mL of 3x PBS (2x 7500g at room temperature
459 for 10 min, followed by 2x 4000g at 40°C for 10 min; all after re-heating the sample/PBS
460 mixture to $\sim 45^{\circ}\text{C}$). This additional cleaning step was implemented to separate bacterial cells
461 from the agarose medium.

462

463 *DNA extraction and PCR sequencing*

464 A modified phenol:chloroform DNA extraction protocol was employed to extract microbial
465 DNA from the artificial sediment matrix. We added 600 μL lysis buffer (TE buffer pH 8, 0.5%
466 SDS, 0.1 mg mL^{-1}) to the pellets prior to incubation at 37°C for 1 h. Then 100 μL of 5 M NaCl,
467 and 80 μL of 10% CTAB were added and the mixture was incubated at 65°C for 10 min. Lysates
468 were transferred to sterile tubes and DNA was extracted following standard
469 phenol:chloroform procedures (Zhou et al. 1996). The obtained DNA was air-dried,
470 resuspended in 20 μL of dH_2O and stored at -20°C until further analysis. DNA quantity and
471 purity was evaluated using a Nanodrop-1000 Spectrophotometer (NanoDrop 1000; Thermo
472 Scientific, USA).

473

474 **(i) PCR amplification and sequencing**

475 To track shifts in the overall composition of the bacterial community, 16S rRNA amplicon
476 sequencing was performed. Amplicons of variable regions V1-V3 of the 16S rRNA gene,
477 generated using the primers 27F (5'-AGAGTTTGATCMTGGCTCAG-3') (Weinbauer et al. 2002)
478 and 519R (5'-GWATTACCGCGGCKGCTG-3') (Lane, 1991; Turner et al., 1999), were sequenced
479 on the Illumina MiSeq platform (Molecular Research LP; Shallowater, TX, USA) following the
480 manufacturer's guidelines. Bacterial 16S rRNA gene sequences were analysed using the QIIME
481 pipeline (Caporaso et al., 2010; Kuczynski et al., 2012). Briefly, paired-end DNA sequences
482 were joined, *de novo* Operational Taxonomic Units (OTUs) were defined at 97% of sequences,

483 and identity was assigned against the Greengenes database (version 13/8/2013) using BLAST
484 (Altschul et al., 1990). Chimeric sequences were detected using ChimeraSlayer (Haas et al.,
485 2011) and filtered out from the dataset. Chloroplasts and mitochondrial reads were removed
486 before downstream analysis. Sequences were then rarefied to the same depth (7265
487 sequences per sample) to remove the effect of sampling effort upon analysis. Raw data files
488 in FASTQ format were deposited in the NCBI Sequence Read Archive (SRA) with the study
489 accession number SRP073850 under Bioproject number PRJNA315465.

490

491 *Statistical analysis*

492 Data were tested for equal variance prior to statistical analysis. Student's *t*-tests were used
493 to compare relative microbial abundances between the different regions of interest (i.e. RAM,
494 BLM and BS).

495

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503

504 **SUPPLEMENTARY INFORMATION**

505 Supplementary Information accompanies the paper on the *Environmental Microbiology*
506 website ([http://onlinelibrary.wiley.com/journal/10.1111/\(ISSN\)1462-2920](http://onlinelibrary.wiley.com/journal/10.1111/(ISSN)1462-2920)).

507 **Figure S1.** Chemical microenvironment as measured with microsensors at the surface of the
508 basal leaf meristem of *Zostera muelleri* maintained in (i) a pre-sterilized environment and (ii)
509 with added native pore water microbes – plant 2.

510 **Figure S2.** Chemical microenvironment as measured with microsensors at the surface of the
511 basal leaf meristem of *Zostera muelleri* maintained in reduced, artificial sediment with added
512 native pore water microbes – plant 3.

513 **Figure S3.** Total sulphide concentration microprofiles at the surface of the basal leaf meristem
514 of *Zostera muelleri* maintained in (i) a pre-sterilized environment and (ii) with added native
515 pore water microbes – plant 1.

516 **Figure S4.** Total sulphide concentration microprofiles at the surface of the basal leaf meristem
517 of *Zostera muelleri* maintained in (i) a pre-sterilized environment and (ii) with added native
518 pore water microbes – plant 2.

519 **Figure S5.** Principal component analysis plot of the rhizosphere microbial community
520 composition, illustrating the separation of the microbial consortia within selected
521 rhizospheric regions of interest.

522 **Figure S6.** Conceptual diagram visualizing sampling areas (i.e. region of interests) within the
523 reduced, artificial sediment.

524 **Table S1.** Photosynthetic parameters as measured by variable chlorophyll fluorescence and
525 measures of the above-ground:below-ground biomass ratios.

526

527

528 **CONFLICT OF INTEREST STATEMENT**

529 The authors declare that the research was conducted in the absence of any commercial or
530 financial relationships that could be construed as a potential conflict of interest.

531

532 **FIGURE LEGENDS**

533 **Figure 1.** The below-ground chemical microenvironment at the basal leaf meristem, i.e., the
534 meristematic region of the rhizome of the seagrass *Zostera muelleri*. (a) and (b) represent
535 microsensor measurements in an artificial sediment matrix with added pore water microbes.
536 (c) and (d) represent microsensor measurements in a pre-sterilized environment, i.e.,
537 sterilized artificial sediment matrix and below-ground tissue surface. (a) and (c) show
538 measurements in darkness. (b) and (d) show measurements in light (photon irradiance of
539 $\sim 150 \mu\text{mol photons m}^{-2} \text{ s}^{-1}$). Black line and symbols show the O_2 concentration; Red line and
540 symbols show the H_2S concentration; Blue line and symbols show pH. The dotted lines
541 indicate the thickness of the plant-derived oxic microzone, and $X = 0$ indicates the surface of
542 the basal leaf meristem. Symbols with error bars represent means \pm S.D ($n = 3\text{-}4$ technical
543 replicates; biological replication of the below-ground chemical microenvironment dynamics
544 is shown in the Supplementary Results; Fig. S1 and S2).

545 **Figure 2.** Microbial diversity in the rhizosphere of the seagrass *Zostera muelleri* determined
546 via 16S rRNA amplicon sequencing. The phylogenetic tree denotes the spatial separation of
547 the microbial consortia as determined via beta diversity analysis by Jackknife comparison of
548 the weighted sequences data. The heat-map shows the abundance of the respective bacterial
549 class/genus within the selected regions of interest, where (o) and (f) denote order and family
550 classification, respectively. The heat-map includes taxonomic groups within each sample that
551 represent $>1\%$ of the total sequences, which cumulatively represents $>85\%$ of the total
552 sequenced data. Diagrams (in %) show the mean relative abundance of designated bacterial
553 classes present within the selected regions of interest of the artificial sediment matrix. All
554 data originate from reduced, artificial sediment with added native pore water microbes
555 (except from data given for the negative control, which originates from a pre-sterilized
556 environment as described in the Experimental procedures section). $n = 2\text{-}3$.

557

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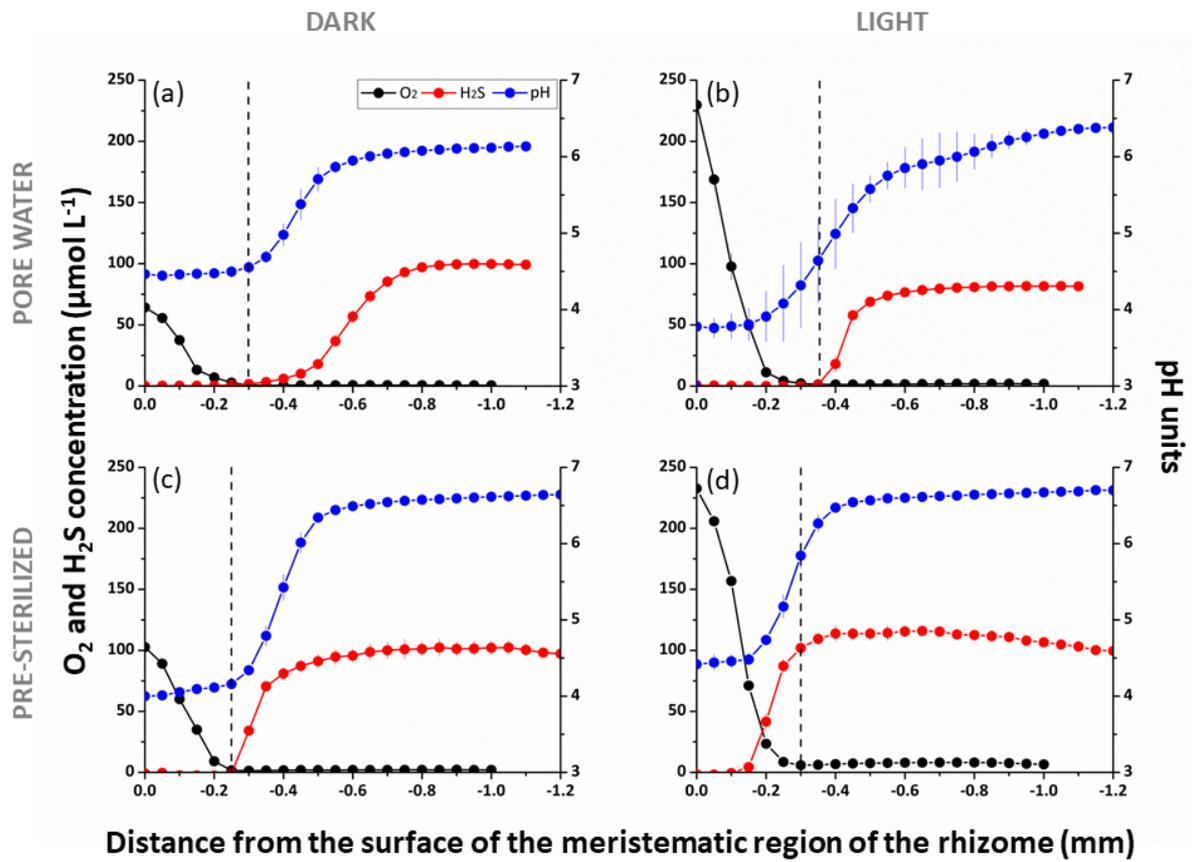
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738 **Figures**

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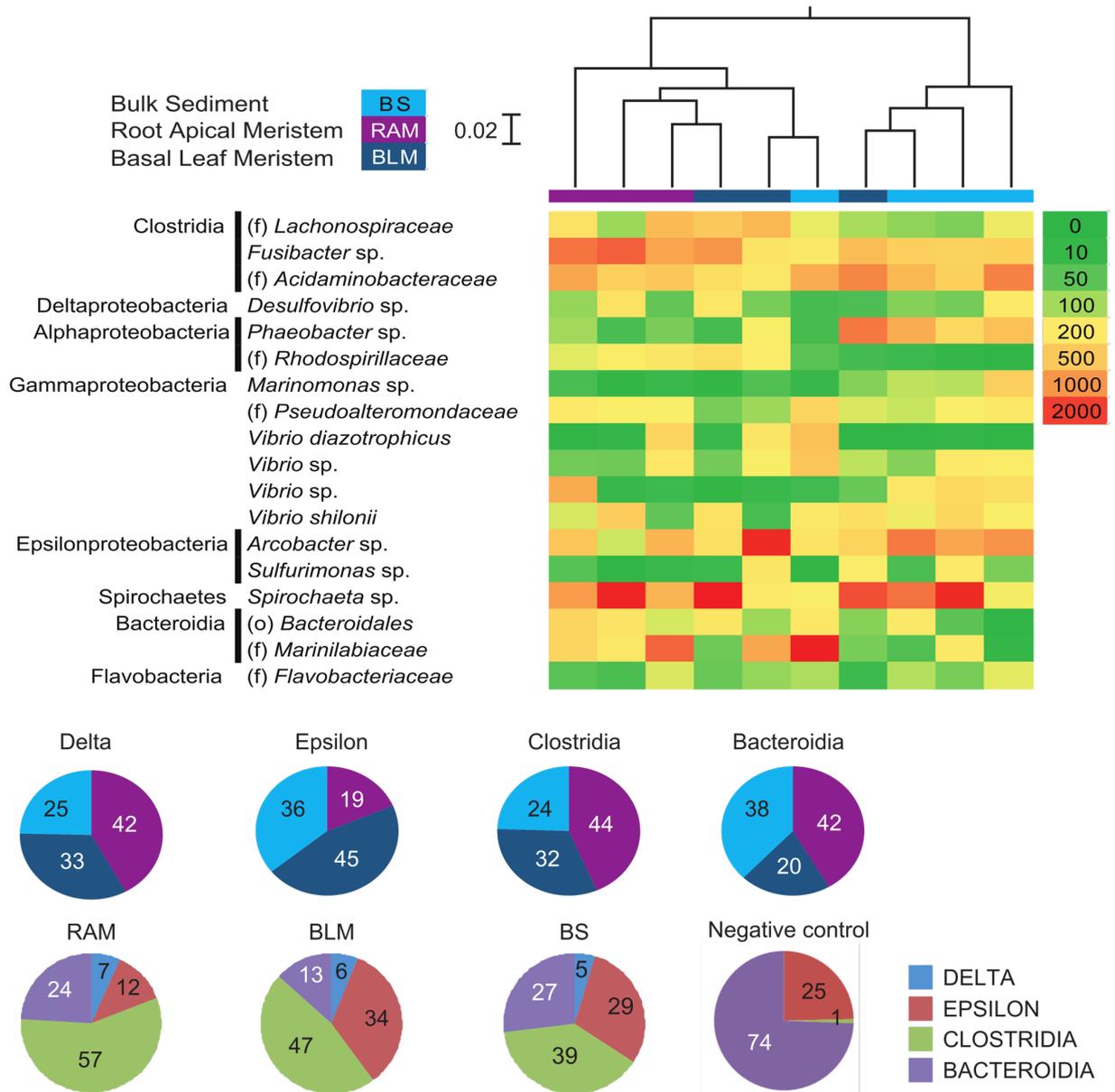
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Fig. 1

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Fig. 2

749 **Tables**

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751 **Table 1.** Radial O₂ loss (ROL), plant-derived H₂S re-oxidation/sediment detoxification and ΔpH in the
 752 immediate rhizosphere of *Z. muelleri*.

	ROL nmol O ₂ cm ⁻² h ⁻¹	H ₂ S re-oxidation nmol H ₂ S cm ⁻² h ⁻¹	S _{tot} ²⁻ oxidation nmol S _{tot} ²⁻ cm ⁻² h ⁻¹	ΔpH pH units
<i>pore water</i>				
dark	-322 ± 39	334 ± 1	555 ± 164 ^a	2.1 ± 0.4 ^a
light	-745 ± 118	418 ± 4	440 ± 6	2.4 ± 0.2
<i>pre-sterilized</i>				
dark	-378 ± 3	636 ± 136 ^a	651 ± 146 ^a	2.5 ± 0.1
light	-891 ± 52	508 ± 33	533 ± 22	2.5 ± 0.2

753 n = 2-3, biological replication. Values are mean ± S.E.M. *Pore water* indicate artificial sediment matrix
 754 with added native pore water microbes. *Pre-sterilized* indicate a sterilized below-ground environment,
 755 i.e., sterilized sediment and below-ground biomass. S_{tot}²⁻ = total sulphide. ^aNote relative high standard
 756 error of the mean (S.E.M.).

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