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Performance, microbial community and fluorescent characteristic of microbial products in a solid-phase denitrification biofilm reactor for WWTP effluent treatment

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ABSTRACT

Microbial products, i.e. extracellular polymeric substance (EPS) and soluble microbial product (SMP), have a significant correlation with microbial activity of biologically based systems. In present study, the spectral characteristics of two kinds of microbial products were comprehensively evaluated in a solid-phase denitrification biofilm reactor for WWTP effluent treatment by using poly (butylene succinate) (PBS) as carbon source. After the achievement of PBS-biofilm, nitrate and total nitrogen removal efficiencies were high of $97.39 \pm 1.24\%$ and $96.38 \pm 1.1\%$, respectively. The contents of protein and polysaccharide were changed different degrees in both LB-EPS and TB-EPS. Excitation-emission matrix (EEM) implied that protein-like substances played a significant role in the formation of PBS-biofilm. High-throughput sequencing result implied that the proportion of denitrifying bacteria, including *Simplicispira*, *Dechloromonas*, *Diaphorobacter*, *Desulfovibrio*, increased to 9.2%, 7.4%, 4.8% and 3.6% in PBS-biofilm system, respectively. According to EEM-PARAFAC, two components were identified from SMP samples, including protein-like substances for component 1 and humic-like and fulvic acid-like substances for component 2, respectively. Moreover, the fluorescent scores of two components expressed significant different trends to reaction time. Gas chromatography-mass spectrometer (GC-MS) implied that some new organic matters were produced in the effluent of SP-DBR due to biopolymer degradation and denitrification processes. The results could provide a new insight about the formation and stability of solid-phase denitrification PBS-biofilm via the point of microbial products.

1. Introduction

With the rapid development of global economy, nitrogen-containing wastewaters are increasingly produced from various industries and discharged into the environment (Cyplik et al., 2012). Till now, biological nitrification and denitrification process have been widely applied in wastewater treatment plant (WWTP) for nitrogen removal from municipal and industrial wastewaters. However, WWTP effluent may still contain a relatively high concentration of inorganic nitrogen such as nitrate and nitrite. It is well accepted that nitrate has a negative impact on the eutrophication of water environment as well as hazards to human health (Nuhoglu et al., 2002; Robinsonlora and Brennan, 2009). Moreover, due to the increasingly strict wastewater discharge standard, advanced treatment of WWTP effluents is becoming an inevitable process for total nitrogen removal.

Recently, many methods have been applied to eliminate nitrate

from WWTP effluent, including ion-exchange, reverse osmosis, electrodialysis and biological denitrification etc. Among all above methods, biological denitrification is generally applied to advanced wastewater treatment because of its advantages, such as high efficiency, cost-effective and environment-friendly (Foglar et al., 2005). In this process, organic carbon and nitrate are utilized as electron donor and acceptor by denitrifying bacteria, respectively. However, secondary effluent has limited carbon source for biological denitrification to achieve high removal efficiency of nitrate. To date, additional organic carbon source, such as ethanol, glucose, acetate, and methanol etc., is commonly applied to denitrification (Gómez et al., 2000). Nevertheless, the above soluble organic carbon sources can bring a lot of problems, such as complex operation, high-cost, and secondary pollution of effluent quality (Wu et al., 2013). In contrast, solid carbon source, also acting as biofilm carrier themselves, has been wildly exploited and developed for nitrate removal that could avoid the risks of overdosing or insufficient carbon dosing. Solid carbon sources are commonly divided into two types: synthetic polymers and natural materials (Wang and Wang, 2009). Compared with natural materials (e.g. wheat straw, cotton and waste newspaper), synthetic biodegradable polymers including Poly (butanediol succinate) (PBS), polyhydroxyalkanoates (PHAs), polycaprolactone (PCL), and poly-3-hydroxybutyric acid (PHB) as carbon sources could resolve color and excessive DOC problems in effluent (Boley et al., 2000; Zhu et al., 2015).

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Microbial products, i.e. extracellular polymeric substance (EPS) and soluble microbial product (SMP), are confirmed to have a significant influence on microbial activity of biologically based systems under dynamic conditions (NI and YU, 2012). As a wide range of complex organic compounds, EPS are located at or outside the cell surface, whereas SMP are the soluble cellular polymeric components dissolved in aqueous solution. Moreover, EPS are often described as dynamic double-layer-like structure, inner and outer, which composed of tightly bound EPS (TB-EPS) and loosely bound EPS (LB-EPS), respectively. It has been confirmed that EPS play an important role in mediating microbial adhesion onto bio-carrier surface and therefore maintain the mechanical stability of biofilm. In contrast, SMP are generally considered as one kind of organic compounds that are released from substrate microbial metabolism (usually with biomass growth and biomass decay) (Krasner et al., 2009; Laspidou and Rittmann, 2002). Since there are co-existed two processes in the solid-phase denitrification system, i.e. biopolymer degradation and denitrification, the production of both EPS and SMP may be more complex than the conventional denitrification process. Hence, it is essential to in-depth explore and identify the structure changes of two kinds of microbial products in order to better understand the formation and stability of solid-phase denitrification PBS-biofilm.

Therefore, the main objective of present study was to demonstrate the feasibility of solid-phase denitrification biofilm reactor (SP-DBR) achievement and long-term performance by using PBS as the carbon source and biofilm carrier for WWTP effluent treatment. To achieve this purpose, the changes of LB-EPS and TB-EPS were qualitatively and quantitatively analyzed before and after PBS-biofilm formation by using

chemical and spectroscopic approaches. Meanwhile, a combination of excitation-emission matrix (EEM), Parallel factor analysis (PARAFAC), and Gas Chromatography-Mass Spectrometer (GC-MS) were applied to SMP production evaluation.

2. Methods and materials

2.1. Experiment set-up

The experiment was conducted in a perspex solid-phase denitrification biofilm reactor (SP-DBR) with the inner diameter and height of 12 and 25 cm, respectively, which had a working volume of 3.4 L (schematic diagram was shown in [Supplementary material](#)). The SP-DBR was filled with 40% PBS granules as solid carbon source and biofilm carrier for biological denitrification. The diameter and height of each PBS carrier were 3 and 5 mm, respectively. Influent wastewater was stored in a water tank (60 L) and flowed into the reactor through a water pump. The volumetric exchange ratio of reactor was 50%, and as a result, hydraulic retention time (HRT) was 24 h.

Seed sludge was obtained from a lab-scale SBR, which has an effective volume of 18 L that running for over 1 year. The initial mixed liquor suspended solids (MLSS) concentration in SP-DBR was about 3.0 g/L. The SP-DBR was sequentially operated at a cycle of 720 min, including 5 min influent, 580 min anoxic phase, 10 min settling, 5 min effluent and 120 min idle. The system was automatically operated through a time controller. The influent wastewater for SP-DBR was obtained from the effluent of the secondary tank after a lab-scale A/O reactor. The main compositions of the influent SP-DBR were as follows (mg/L): $\text{NH}_4^+ \text{eN}$, 3.5 ± 1.1 ; $\text{NO}_2^- \text{-N}$, 2.0 ± 0.3 ; $\text{NO}_3^- \text{-N}$, 25.9 ± 1.7 and COD, 30 ± 2.0 .

2.2. EPS, SMP extraction and fluorescent spectra

Two types of EPS, including LB-EPS and TB-EPS, were extracted by using a heat method from seed sludge and biofilm as described previously literature (Yin et al., 2015). The supernatant was collected from reactor by centrifugation at 4000 rpm, and subsequently filtrated at 0.45- μm membrane filter to represent SMP. 3D-EEM was obtained by subsequent scanning excitation from 200 to 400 nm at 10 nm increments, and the emission wavelength from 280 to 550 nm with a slit-width of 10 nm. Synchronous fluorescence spectra of SMP samples were determined by ranging the excitation wavelengths from 250 to 550 nm with offset ($\Delta\lambda$) of 60 nm. All samples were scanned at a speed of 1200 nm/min. PARAFAC decomposed N-way arrays into N loading matrices was applied for the further analysis of SMP fluorescence data (Yu et al., 2010).

2.3. Microbial community analysis

Total genome DNA from two samples was extracted using CTAB/SDS method. DNA concentration and purity was monitored on 1% agarose gels. According to the concentration, DNA was diluted to 1 ng/ μL using sterile water. A pair of universal 16S rRNA gene primers, included 515F (GTGCCAGCMGCCGCGTAA) and 806R (GGACTACHVGGGTWCTAAT) were used to amplify V4 region of the 16S rRNA gene (Caporaso and Gordon, 2011). The reverse primer contained a 6-bp error-correcting barcode that was unique to each sample. The barcode was permuted for each sample and allowed the identification of individual samples in a mixture in a single Illumina MiSeq sequencing run. DNA was amplified in triplicate for each sample following the protocol described previously. The library quality was assessed on the Qubit@ 2.0 Fluorometer (Thermo Scientific) and Agilent Bioanalyzer 2100 system.

2.4. Analytical methods

$\text{NH}_4^+ \text{eN}$, $\text{NO}_2^- \text{-N}$, $\text{NO}_3^- \text{-N}$, TN concentrations in this experiment were determined according to their standard methods (Association and Washington, 1995). Polysaccharide content was determined by using anthrone-sulfuric acid method with glucose as the standard, and protein content was determined by using modified Lowry method with bovine serum albumin (BSA) as the standard. 3D-EEM spectra were characterized by using a Luminescence spectrometer (LS-55, Perkin-Elmer Co., USA). Types of organic composition of SMP sample was detected by using GC-MS (Shimadzu QP-2010). Fourier transform infrared spectroscopy (FTIR) analysis by using a Perkin-Elmer FTIR spectrometer (United States). Before FTIR analysis, EPS samples were freeze-dried at -60°C .

3. Results and discussion

3.1. Achievement and performance of SP-DBR

After successive 180 days' operation, seed sludge was successfully attached onto the PBS surface and formed biofilm. Fig. 1 shows the long-term performance of nitrogen removal in the SP-DBR during the whole operation. It was found that the influent $\text{NO}_3^- \text{-N}$ was average at 25.87 ± 1.74 mg/L, whereas $\text{NH}_4^+ \text{eN}$ and $\text{NO}_2^- \text{-N}$ were 1.44 ± 0.22 and 0.25 ± 0.21 mg/L, respectively. As shown in Fig. 1A-C, the $\text{NO}_3^- \text{-N}$ concentration in the effluent decreased to 1.22 mg/L and $\text{NO}_3^- \text{-N}$ removal efficiency increased to 94.47% in start-up period (days 1-5), while the effluent $\text{NO}_2^- \text{-N}$ and $\text{NH}_4^+ \text{eN}$ remained at low levels (less than 1.0 mg/L). As the formation of PBS-biofilm, the nitrate removal efficiency further increased from $95.86 \pm 2.91\%$ (days 6-44) to $97.34 \pm 1.23\%$ (days 45-180). This result may be attributed to small craters and holes that developed by microbial corrosion, which pro-

vided greater adhesion area and anoxic environment for denitrification

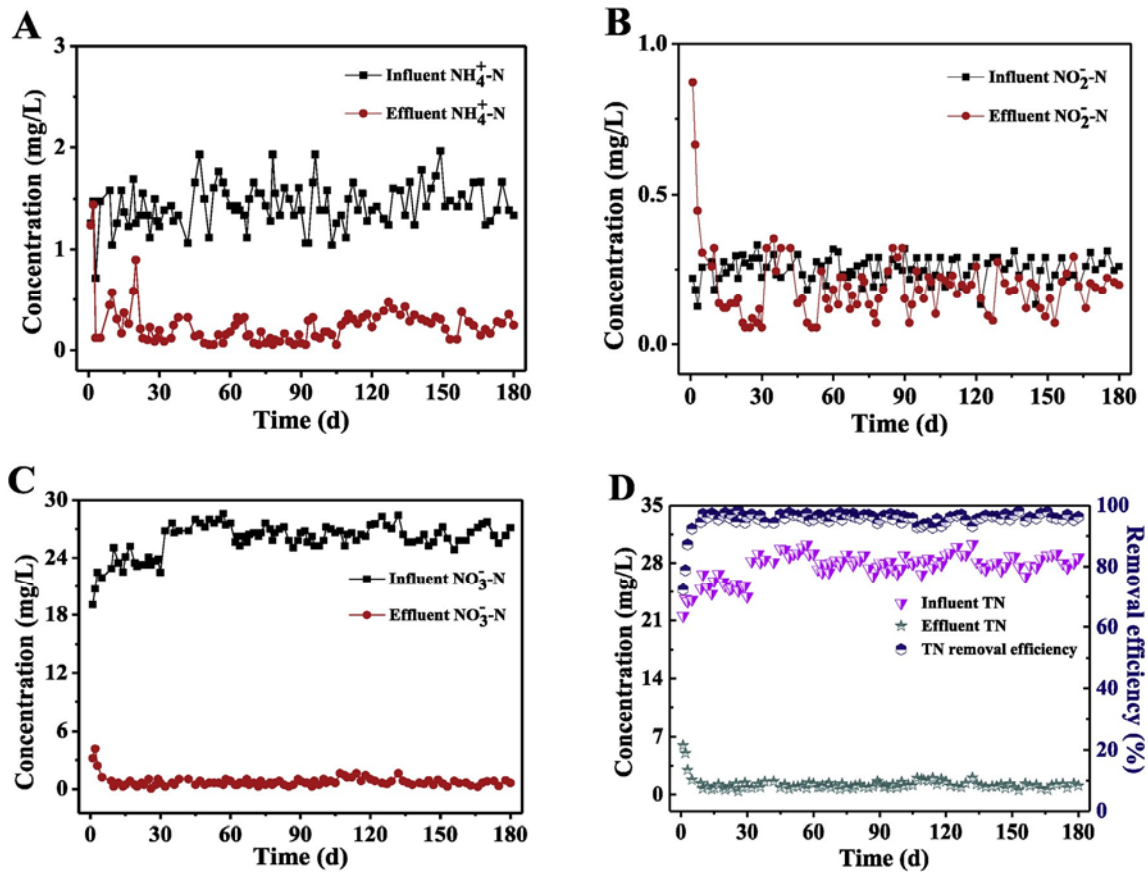


Fig. 1. Long-term performance of nitrogen removal in the SP-DBR during the whole operation: Influent and effluent concentrations of $\text{NH}_4^+\text{-N}$ (A); $\text{NO}_2^-\text{-N}$ (B); $\text{NO}_3^-\text{-N}$ (C) and TN removal efficiencies (D).

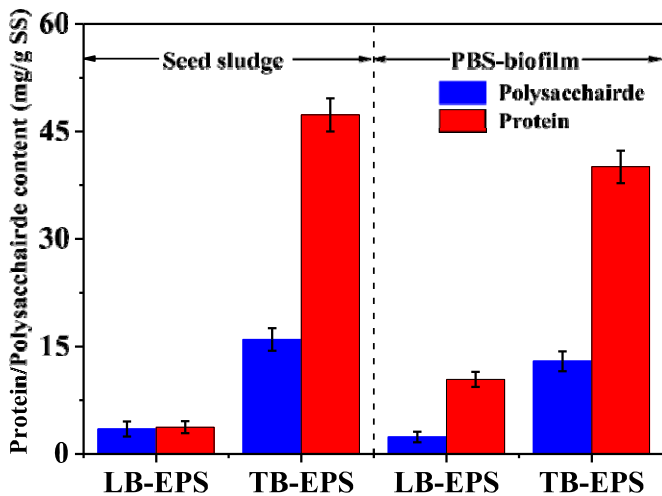


Fig. 2. The changed contents of LB-EPS and TB-EPS during the SP-DBR achievement process.

process, as similarly reported by Wu et al. (2013). Meanwhile, the TN removal efficiency was stabilized at approximately $96.38 \pm 1.1\%$ (Fig. 1D).

The nitrogen transformation process in a typical cycle during stable operation phase was shown in Supplementary material. After the influent pumped into the reactor (0–5 min), $\text{NO}_3^-\text{-N}$ concentration was rapidly reduced from 23.5 to 12.0 mg/L due to the dilution effect. Thereafter, $\text{NO}_3^-\text{-N}$ concentration decreased linearly from 11.2 to 0.6 mg/L with the final removal efficiency of 97.5% in the following eight hours. In contrast, $\text{NO}_2^-\text{-N}$ concentration generally increased to

the maximum peak of 2.6 mg/L at 7 h, and subsequently decreased to 0.6 mg/L at 10 h. Compared with acetate, methanol and glucose as carbon and electron sources, nitrite accumulation was much lower in solid-phase denitrification system (Ge et al., 2012). It is well reported that nitrite accumulation during denitrification process could be also affected by many parameters, such as carbon source species, carbon/nitrogen (C/N) ratio, and pH value (Gómez et al., 2000; Glass and Silverstein, 1998).

In order to provide the evidence for PBS biodegradation and biofilm growth, SEM observations of PBS surface before and after utilization were conducted during 180 days' operation was shown in Supplementary material. It was found that the surface of raw PBS was relatively smooth, whereas it became rough with many hollows and holes caused by microbial corrosion. As a result, a large surface area and anoxic environment of PBS was provided for microorganism attachment and biological denitrification (Wu et al., 2013). Meanwhile, two different morphologies (rhabditiform and filament) of microorganism were observed in holes and hollows. Filamentous organisms played significant role on the connection between microorganism and PBS-surface that bring about the formation and stability of biofilm. SEM images also demonstrated that PBS was comparatively durable as carbon source for denitrification.

3.2. Characteristic of EPS in biofilm formation

Fig. 2 shows the contents of LB-EPS and TB-EPS from seed sludge and PBS-biofilm during the SP-DBR achievement process. It was observed that protein and polysaccharide contents in TB-EPS were much higher than those in LB-EPS, indicating that TB-EPS was the main distribution and constituent in the double-layered of EPS. Although LB-EPS accounts for a small proportion in total EPS, it plays an important

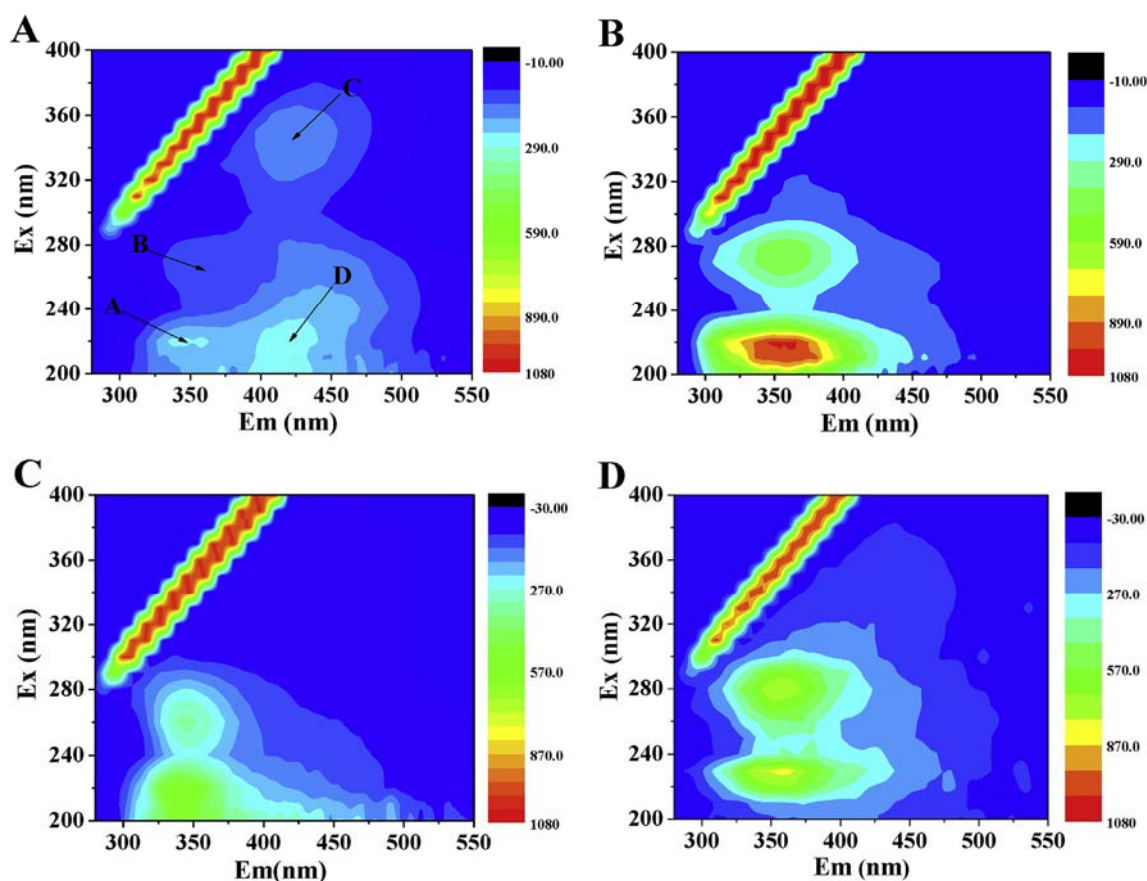


Fig. 3. Excitation emission matrix (EEM) spectra of LB-EPS and TB-EPS extracted from seed sludge and biofilm: (A) LB-EPS and (B) TB-EPS of seed sludge; (C) LB-EPS and (D) TB-EPS of biofilm.

role in the membrane fouling, sedimentation and flocculation etc. (Li and Yang, 2007; Wang et al., 2009). After the achievement of PBS-biofilm, protein and polysaccharide contents were reduced by 17.2% and 19.0% of TB-EPS, respectively (Fig. 2). It has been reported that protein and polysaccharide were represented as hydrophobic and hydrophilic substances in previous research, respectively (Bos et al., 1999). In present study, protein/polysaccharide ratio was lower in TB-EPS of biofilm than that of seed sludge, suggesting that the relative hydrophobicity increased after the formation of PBS-biofilm, which may be attributed to high-MW and cross-linked structure of polysaccharides exhibited some influences on cell aggregation (Seviour et al., 2009).

Fig. 3 shows the EEM spectra of LB-EPS and TB-EPS extracted from seed sludge and biofilm. As shown in Fig. 3A, three main peaks (A, C and D) were identified in LB-EPS of seed sludge at excitation/emission wavelengths (Ex/Em) of 220/350, 340/420 and 225/430 nm, which were assigned to aromatic tryptophan protein-like substances, humic-like substances and fulvic acid-like substances, respectively (Meng et al., 2011). In contrast, only two peaks (Peak A and Peak B) were identified at Ex/Em of 220/340 nm and 265–270/340–350 nm in LB-EPS of biofilm (Fig. 3C), corresponding to the presence of aromatic and tryptophan protein-like substances. Peak A expressed higher fluorescent intensity in biofilm than that of seed sludge, suggesting that protein in LB-EPS was more sensitive to anoxic environment. It was found from Fig. 3B and D, two major peaks (aromatic protein-like and tryptophan protein-like substances) in TB-EPS were observed at Ex/Em wavelengths of 225–230/350–370 nm and 270–285/352.5–360 nm regardless of seed sludge or biofilm. For biofilm, there were only aromatic protein-like and tryptophan protein-like substances in two types of EPS, implying that protein-like substances played a significant role in the

formation of PBS-biofilm. The significantly changed EPS composition may be in connection with the microorganisms by transferring from aerobic to anoxic environments, as similarly reported by Hong et al. (2007).

The FTIR spectra of LB-EPS and TB-EPS extracted from seed sludge and PBS-biofilm were shown in Supplementary material. It was observed that the position and number of FTIR peaks in two types of EPS of seed sludge and biofilm were no obvious difference, indicating that the types of chemical groups were similar. Specifically, the broad band around 3200–3600 cm^{-1} was related to the stretching vibration of both O–H (of the polysaccharides) and amino groups (of the proteins) (Sun et al., 2009), which was more abundant in TB-EPS than that in LB-EPS regardless of seed sludge or PBS-biofilm. This result suggested that the TB-EPS had higher abundances of proteins and polysaccharides compared with LB-EPS (Basuvaraj et al., 2015). Meanwhile, TB-EPS of seed sludge exhibited higher peak intensities associated proteins than PBS-biofilm, which was in agreement with the chemical and fluorescent data. The band at approximately 1641–1651 cm^{-1} region was corresponded to the functional groups C=O stretching vibrations of aromatic proteins and C=O vibrations of amides, which existed in all of EPS samples and favored bio-flocculation (Badireddy et al., 2010). The bands near 1384–1429 cm^{-1} were related to symmetric stretching of eCOO⁻ groups associated with amino acids and deprotonated carboxylic acid groups, which indicated the acidic nature of EPS components (Badireddy et al., 2010). A sharp peak around 1112 cm^{-1} was correlated with the C–O–C and C–O–H stretching vibrations of polysaccharides and aromatics, indicating that polysaccharides were an important component of EPS (Wei et al., 2017). Some bands at “fingerprint zone” (< 1000 cm^{-1}) might be assigned to phosphate group, which was one of the functional groups of nucleic acids. Compared with

LB-EPS, a new weak absorption band appeared at 2933–2964 cm^{-1} of TB-EPS, which were corresponding with the asymmetrical C–H stretching vibration of the aliphatic CH_2 -group (Liang et al., 2010). Furthermore, the bands of LB-EPS at 3448, 1651 and 1384 cm^{-1} in seed sludge were shifted to 3411, 1643 and 1411 cm^{-1} in biofilm, indicating that the chemical structure of EPS had changed after the formation of PBS-biofilm. FTIR data illustrated that EPS was a complex mixture, which was composed of proteins, polysaccharide, humic substances, amino acids and nucleic acids.

3.3. Microbial community structure

High-throughput sequencing analysis of 16S rRNA amplicon technology was introduced to evaluate the variation and characteristics of microbial community during biofilm cultivation. Fig. 4A shows the different species of bacteria communities of seed sludge and biofilm at the phylum level. It was found that the vast majority of sequences in two sludge samples belonged to 10 phyla: *Proteobacteria*, *Bacteroidetes*, *Firmicutes*, *Acidobacteria*, *Planctomycetes*, *Spirochaetes*, *Chloroflexi*, *Verucomicrobia*, *Actinobacteria* and *Latescibacteria*. There was a little difference of proportion in phylum of two samples. To be specific, the proportion of *Proteobacteria*, *Firmicutes* and *Chloroflexi* increased from 59.10%, 4.30% and 1.26%–70.83%, 6.04% and 1.80%, while *Bacteroidetes* decreased from 9.35% to 5.18% with PBS-biofilm formation, indicating the microbial communities structure has changed caused by feed environment. Miura et al. (2007) reported that *Chloroflexi* was a special phylum responsible for SMP degradation such as carbohydrates and cellular materials. It is well known that the main function of anaerobic bacteria is destroy large molecules matter, such as proteins, cellulose and polysaccharides, which is an essential step for hydrolyzation and utilization of the solid carbon source.

As shown in Fig. 4B, the relative abundance of *Betaproteobacteria* and *Flavobacteriia* at the class level decreased from 47.40% to 3.9%–34.06% and 0.6%, while *Deltaproteobacteria*, *Gammaproteobacteria*, *Negativicutes* observably increased from 2.27%, 5.11% and 0.13%–16.31%, 10.09% and 4.51%, respectively. In this study, within the *Betaproteobacteria*, *Comamonadaceae* and *Rhodocyclaceae* families became more abundant in biofilm sample (increased from 4.27% to 3.43%–17.94% and 10.24%, data not shown). Hiraishi and Khan reported that most of polyhydroxyalkanoate-degrading denitrifying bacteria have been proved belong to the family *Comamonadaceae* within *Betaproteobacteria* (Hiraishi and Khan, 2003). Miao et al. (2016) elaborated that *Comamonadaceae* was a dominating member participating in denitrification. Tang et al. (2017) reported that *Rhodocyclaceae* increased from 10.7% to 14.8% when fermentation liquid of food waste (FLFW) as an external carbon source added to anoxic/oxic-membrane bioreactor (A/O-MBR).

Fig. 4C shows the microbial community structure at genus levels with different sludge samples. The proportion of four genera, including *Simplicispira*, *Dechloromonas*, *Diaphorobacter*, *Desulfovibrio*, have an obvious increased with PBS-biofilm formation, which are identified to denitrifying bacteria. Khan et al. (2007) founded that most of the isolates (90%) were assigned to genera of the *Betaproteobacteria*, especially those of the family *Comamonadaceae* (i.e., *Acidovorax*, *Brachymonas*, *Comamonas*, *Diaphorobacter* and *Simplicispira*). Meanwhile, *Diaphorobacter* was discovered that it possibly transfers electron sequentially in the denitrification system from nitrate to dinitrogen formation reported by Chakravarthy et al. (2011). Besides, Coates et al. (2001) reported that two *Dechloromonas* strains, RCB and JJ, can completely mineralize various mono-aromatic compounds with nitrate as the electron acceptor in the absence of O_2 . Moreover, small proportion of denitrifying

bacteria was observed in genera level including *Thauera*, *Pseudomonas*, *Denitromonas*, which are the main functional microorganisms related with denitrification (Babatsouli et al., 2015).

3.4. SMP production from SP-DBR system

3.4.1. Fluorescence spectra of SMP in a typical cycle

EEM spectra has been successfully utilized to recognize the components of SMP owing to its particular advantages including strong selectivity, high sensitivity and easy operation (Yamashita and Jaffé, 2008). Fig. 5 shows 3D-EEM spectra of SMP samples in a typical cycle of SP-DBR (from 5 to 600 min). It was observed that Peak A, C and D were identified at Ex/Em wavelengths of 280/350, 240–260/430–450 and 340/440 nm, which were assigned to tryptophan protein-like substances, fulvic acid-like and humic-like substances, respectively (Sheng and Yu, 2006). Peak B was firstly observed at Ex/Em of 230/350 nm at the 240 min, which was regarded as aromatic protein-like substrates (Chan et al., 2009).

Generally, the release of SMP from solid-phase denitrification system was related the co-existed two processes: biopolymer degradation and denitrification. It was found that all above intensities of four fluorescence peaks expressed different trends with reaction time, implying that the species of SMP were changed in the influence of two-coexisted processes. The fluorescence intensity of Peak A increased from 248.95 to 391.60 a.u. at the first hour, which may be attributed to the increased microbial metabolism from denitrification process in the presence of solid substrate source (Thompson, 1976). At subsequent period, it was generally reduced to 142.77 a.u. till the end of cycle (600 min), suggesting that the released protein-like substances may be gradually utilized as carbon source for denitrification process. Synchronous fluorescence results were consistent with 3D-EEM (detailed results in Supplementary material). Meanwhile, the intensity of Peak C had a slight change from 384.75 to 312.90 a.u., suggesting that fulvic acid-like substance were reduced through the denitrification process. The formation of SMP in solid-phase biological denitrification biofilm reactor was significant different with soluble organic carbon sources reported by Xie et al. (2010), who found that SMP concentration increased steadily as the utilization of external substrate (e.g. NaAC).

3.4.2. PARAFAC analysis

PARAFAC combined with 3D-EEM is utilized to decompose the fluorescence spectra into independent fluorescent phenomena (Bro, 1997). Fig. 6A shows the fluorescence components of SMP samples identified by PARAFAC based on EEM spectra. It was found that two Fluorescence components were derived from PARAFAC model. For component 1, two main peaks were observed at Ex/Em of 275/370 nm and 230/370 nm, indicating the presence of tryptophan protein-like substances and aromatic protein-like substances, respectively. Component 2 also had two obvious peaks that occurring around Ex/Em of 350/440 and 250/440 nm, which were identified as humic-like and fulvic acid-like substances. Similar two kinds of components from PARAFAC have been also identified from SMP released from aerobic granular sludge under the stress of 2, 4-Dichlorophenol (Wei et al., 2016).

Fig. 6C shows the fluorescence intensity scores of two PARAFAC-derived components in SMP samples at different reaction time. It was found that the scores of two components expressed significant different trends during the denitrification process. The fluorescence intensity score of Component 1 increased from 0.40 to 0.42 in the initial 60 min, suggesting the active microbial metabolism with utilization of solid substrate source for denitrification process under feast condition,

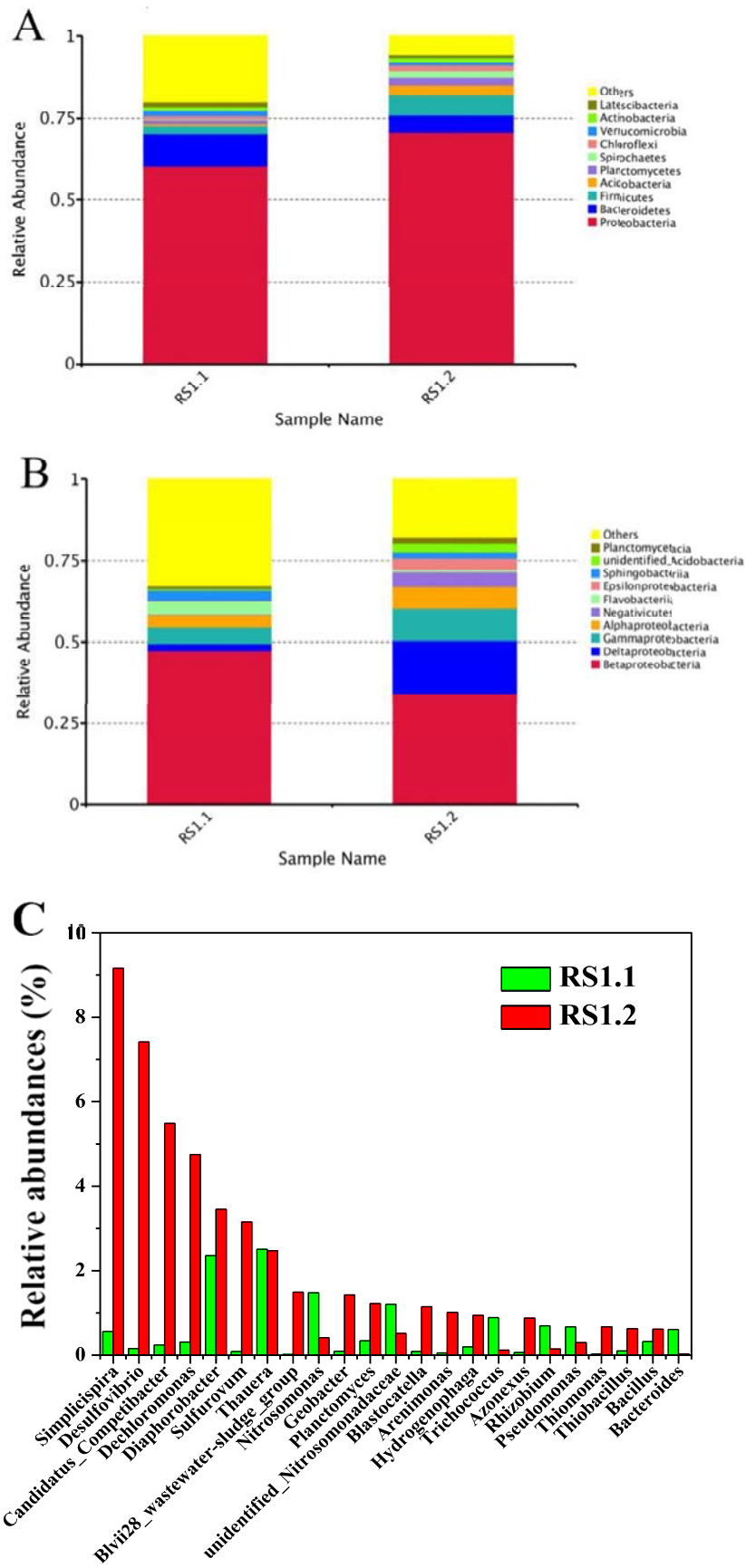


Fig. 4. Microbial community structure at phylum levels (A), class levels (B) and genus levels (C) of two sludge samples (Seed sludge sample, RS1.1; Biofilm sample, RS1.2).

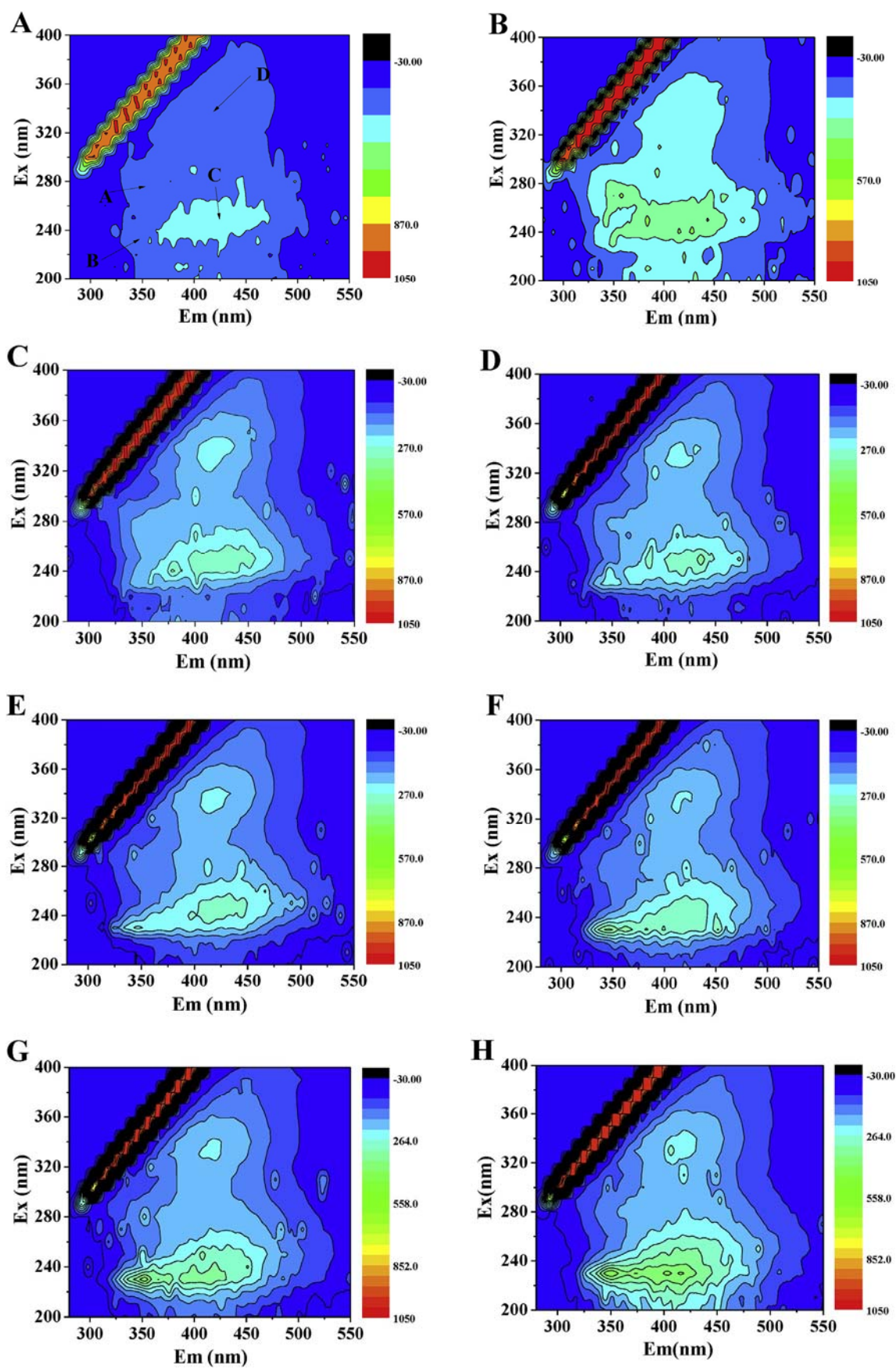


Fig. 5. Fluorescence spectra of SMP samples in a typical cycle of SP-DBR: (A) 5 min; (B) 60 min; (C) 120 min; (D) 240 min; (E) 360 min; (F) 480 min; (G) 540 min; (H) 600 min.

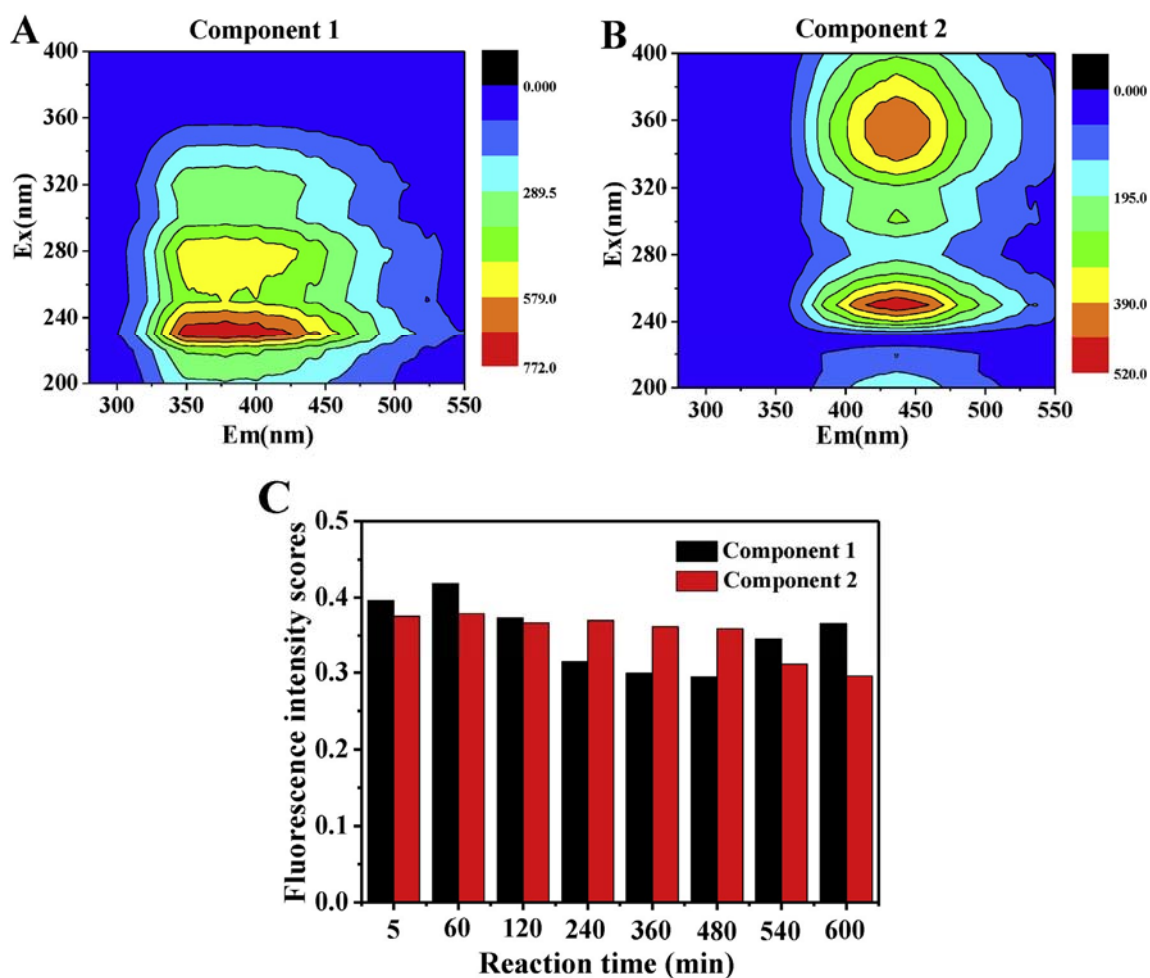


Fig. 6. Fluorescence components of SMP samples identified by PARAFAC based on EEM spectra: (A) Component 1; (B) Component 2; (C) Fluorescence intensity scores of two PARAFAC-derived components.

corresponding to previous literature reported by Yang et al. (2017). Afterwards, the score of Component 1 was generally reduced to 0.29 at 480 min, implying that the released protein-like substances may be utilized as carbon source for denitrification process. After the substrate was completely consumed, the increase in SMP concentration may be because the SMP produced from microbial decay (Xie et al., 2010). Correspondingly, the fluorescence intensity score in Component 2 was reduced from 0.38 at 5 min to 0.30 at 600 min. Moreover, it was well reported that the humic-like substances were mainly substrate-utilization associated, while the fulvic acid-like substances were non growth-associated (Ni et al., 2010).

3.4.3. GC-MS analysis

GC-MS method was further applied as qualitative and quantitative analysis the type of main organic matters in the influent and effluent of SP-DBR (Bai et al., 2017). As shown in Table 1, there were about 16 classes organic matters identified from influent of SP-DBR, including alcohols, ketones, esters, and alkanes etc. The N-containing compounds was mainly constituted by pyridine, 2-methyl and pyrrole molecules, which may be attributed to proteins in the organic fraction of wastewater (Gonzalez-Vila et al., 2009). After solid-phase biological denitrification, the number of the main organic species decreased to 7 (Table 2). It can be seen that some organic matters, such as Succinaldehyde, 1, 3-Propanediol, 2, 1-Benzisoxazole was not detected in the effluent. Moreover, some new organic substances, such as 1,2-Benzenedicarboxylic acid dibutyl ester, 1,3-Benzodioxol-2-one, which were related with aromatic and poly-aromatic substances (Pognani

et al., 2011). This result may be due to the decomposition of quondam organic matters by microbial activities in biological denitrification and biopolymer degradation processes. In previous studies, Pognani et al. (2011) reported that 16 classes organic matters identified in a full-scale composting for treatment sewage sludge and biowaste, which including fatty acids, cycle-alkanes, alkanes, cycle-alkenes, alkenes, alcohols, ketones, aldehydes, halogens, nitrogen compounds and aromatic molecules etc. Dignac et al. (2005) investigated that fatty acids were abundant composition of organic matter in various fresh and composted wastes.

4. Conclusions

In summary, WWTP effluent was successfully treated in a SP-DBR by using PBS as carbon source and biocarrier. High NO_3^- -N and TN removal efficiencies achieved during stable operation phase. The contents and compositions of both LB-EPS and TB-EPS were significant changed after PBS-biofilm formation. Microbial analysis suggested that *Comamonadaceae* and *Rhodocyclaceae* within the *Betaproteobacteria* were dominant families. Two components were identified from SMP in PBS-biofilm system, including protein-like substances for Component 1, humic-like and fulvic acid-like substances for Component 2. Moreover, the number of organic matter species in two SMPs were detected by GC-MS reduced from 16 to 7.

Table 1
Main organic matters in the influent of DBR detected by GC-MS.

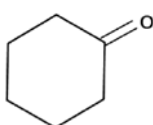
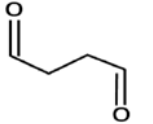

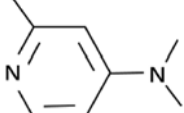
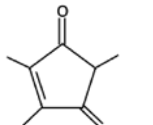
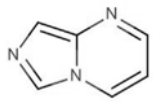
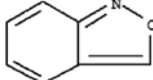
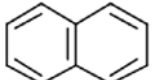
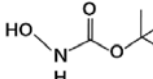
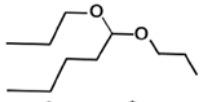
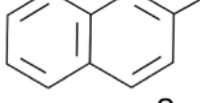
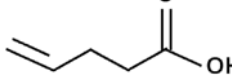
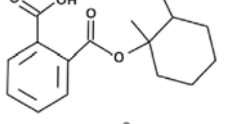
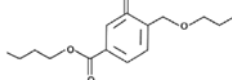
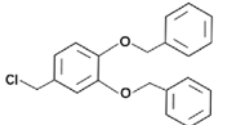

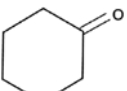
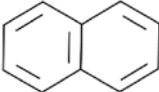
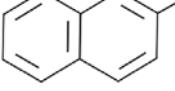
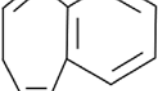
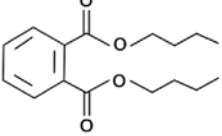
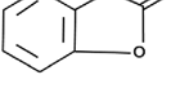

Full name	Chemical formula	#CAS	Chemical structure
Cyclohexanone	C ₆ H ₁₀ O	108-94-1	
Succindialdehyde	C ₄ H ₆ O ₂	638-37-9	
1,3-Propanediol	C ₃ H ₈ O ₂	504-63-2	
4-Pyridinamine,N,N,2-trimethyl-	C ₈ H ₁₂ N ₂	37941-24-5	
2,3,5-Trimethyl-4-methylene-2-cyclopenten-1-one	C ₉ H ₁₂ O	29765-85-3	
Imidazo[1,5-a]pyrimidine	C ₆ H ₅ N ₃	274-67-9	
2,1-Benzisoxazole	C ₇ H ₅ NO	271-58-9	
Naphthalene	C ₁₀ H ₈	91-20-3	
Tert-Butyl N-Hydroxycarbamate	C ₅ H ₁₁ NO ₃	36016-38-3	
Pentanal dipropyl acetal	C ₁₁ H ₂₄ O ₂	13112-64-6	
2-Methyl naphthalene	C ₁₁ H ₁₀	91-57-6	
4-Pentenoic acid	C ₅ H ₈ O ₂	591-80-0	
(Dimethylcyclohexyl) hydrogen phthalate	C ₁₆ H ₂₆ O ₄	1322-94-7	
1,4-Benzenedicarboxylic acid 1,4-dibutyl ester	C ₁₆ H ₂₂ O ₄	1962-75-0	
Toluene,3,4-bis(benzyloxy)-a-chloro- (7Cl)	C ₂₁ H ₁₉ ClO ₂	1699-59-8	
9-Octadecenamide	C ₁₈ H ₃₅ NO	3322-62-1	

Table 2

Main organic matters in the effluent of DBR detected by GC-MS.

Full name	Chemical formula	#CAS	Chemical structure
Cyclohexanone	C ₆ H ₁₀ O	108-94-1	
Naphthalene	C ₁₀ H ₈	91-20-3	
2-Methyl naphthalene	C ₁₁ H ₁₀	91-57-6	
7H-Benzocycloheptene	C ₁₁ H ₁₀	264-09-5	
1,2-Benzenedicarboxylic acid dibutyl ester	C ₁₆ H ₂₂ O ₄	84-74-2	
1,3-Benzodioxol-2-one	C ₇ H ₄ O ₃	2171-74-6	
Erucylamide	C ₂₂ H ₄₃ NO	112-84-5	

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Appendix A. Supplementary data

Supplementary data related to this article can be found at <https://doi.org/10.1016/j.jenvman.2018.09.002>.

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