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1 **Techno-economic assessment of fertiliser drawn forward osmosis process**
2 **for greenwall plants from urban wastewater**

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18 **Highlights**

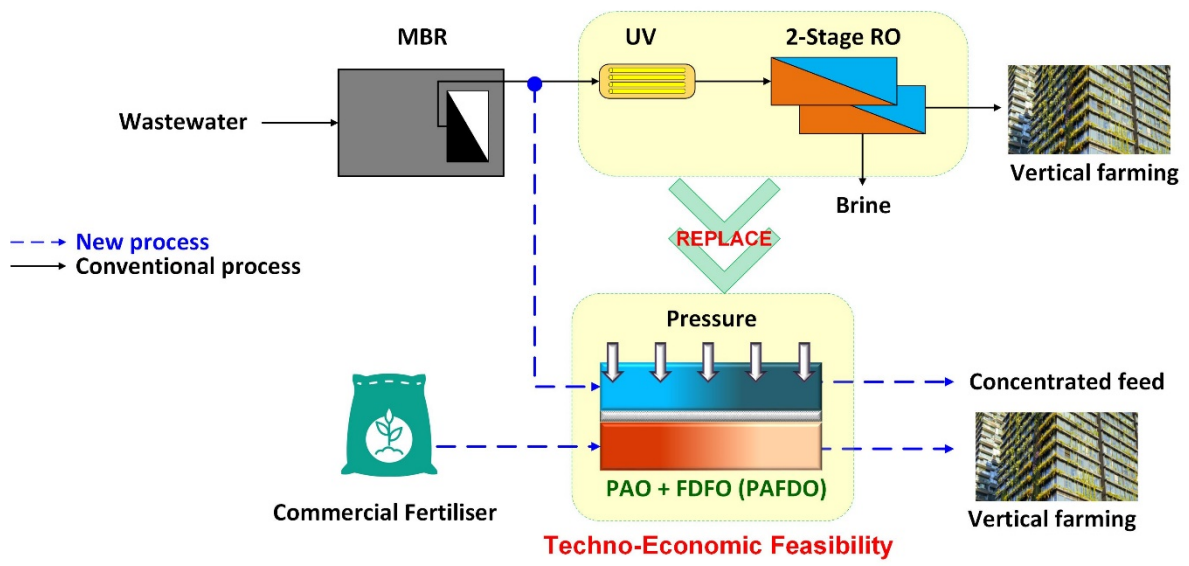
- 19 • Fertiliser drawn forward osmosis (FDFO) for urban wastewater reuse was investigated
- 20 • Applying additional pressure in the FDFO was considered as an alternative
- 21 • Pressure applied at lower than a 60-fold fertiliser dilution was recommended
- 22 • Water flux of $10 \text{ Lm}^{-2}\text{h}^{-1}$ was required to make the FDFO economically feasible
- 23 • Pressure-assisted FDFO could be competitive with the existing water reuse facility

24

25

26 Graphical abstract

Conventional water reuse facility vs. Pressure-assisted fertiliser drawn forward osmosis process



27

28 **Abstract**

29

30 Pressure-assisted osmosis (PAO) has been suggested to integrate with fertiliser driven forward
31 osmosis (FDFO) to improve the overall efficiency of simultaneous wastewater reuse and
32 fertiliser osmotic dilution. This study aims to demonstrate the techno-economic feasibility of
33 pressure-assisted fertiliser driven forward osmosis (PAFDO) hybrid system compared to the
34 existing ultraviolet and reverse osmosis (UV-RO) process. The results showed that coupling
35 FDFO with PAO (i.e. PAFDO) could help fulfill the water quality required for greenwall
36 fertigation. An economic analysis on capital and operational costs for the PAFDO showed that
37 the PAO mode application at a lower FDFO dilution stage could significantly reduce the costs.
38 However, when considering the different applied pressures in PAO (i.e. 2, 4, and 6 bar), the
39 increase in the total water cost was not significant. This indicates that the dilution stage for
40 applying PAO is more sensitive to the total water cost of the PAFDO than the applied pressure.
41 A coupling of higher average water flux (>10 L/m²h) and lower draw solution (DS) dilution
42 factor (DF <60) is recommended. Therefore, this could make the PAFDO system economically
43 viable compared to the benchmark for the UV-RO disinfection system.

44

45 **Keywords**

46 Forward osmosis; Fertigation; Pressure assisted osmosis; Wastewater reuse; Techno-economic
47 assessment.

48

49 **1 Introduction**

50

51 The world population is projected to cross 9 billion by 2050 (Diaz et al. 2017). The rapid
52 population growth coupled with climate change and urbanization have placed an increasing
53 demand for limited potable water resources throughout the world. As the agricultural sector
54 accounts for around 70 % of the world freshwater consumption (Wisser et al. 2008), food
55 production may therefore soon be hindered by water availability. To guarantee food and water
56 security, robust and sustainable methods to supply clean water are increasingly needed while
57 mitigating the impact on the environment (Zhang et al. 2017, Zhang et al. 2018).

58

59 To date, reverse osmosis (RO) process has worldwide attention in both wastewater reclamation
60 and desalination mainly due to the development of good performance membranes and its lower
61 environmental impact compared to the thermal technologies (Al-Obaidi et al. 2017, Bunani et
62 al. 2015). A recent study investigated the performances of two different types of RO
63 membranes in removal of various dissolved species in secondary effluent stream and showed
64 that the quality of the RO permeates is suitable for agricultural irrigation (Bunani et al. 2015).
65 In addition, the use of reclaimed water produced from an integrating RO system consisting of
66 RO-ultraviolet (UV) or UV-RO was demonstrated for irrigation or non-potable applications
67 (Kargari and Mohammadi 2015, Ordóñez et al. 2011, Von Gottberg 2005). Although such RO
68 integrated system can produce high-quality water for reuse, this still leads to high operational
69 costs as the pressurised system requires more pumping costs and cleaning operations (Chekli
70 et al. 2016).

71

72 One of the most promising technologies is fertiliser drawn forward osmosis (FDFO) process,
73 which has recently gained global attention. In the concept of the FDFO process, when a highly

74 concentrated fertiliser solution (i.e. draw solution, DS) and a low saline water (i.e. feed solution,
75 FS) are separated by a selectively permeable membrane, this allows passage of fresh water
76 from the FS to the DS by osmotic concentration differential. The diluted fertiliser, containing
77 fertiliser nutrients, can thus be directly used for irrigation of crops. The concentration of the
78 fertiliser after dilution in the FDFO process must be acceptable for direct application and this
79 has however been found to be challenging (Phuntsho et al. 2013 a). The final diluted fertiliser
80 produced from the FDFO process is limited by the feed stream concentration (i.e. osmotic
81 pressure) based on the osmotic equilibrium between the feed and draw streams (Phuntsho et al.
82 2014). When feed water solution with a high content of salt is used for such application, the
83 final product at osmotic pressure equilibrium could have much higher concentration of
84 nutrients than allowable levels for irrigation. Reclaimed water with the relatively low salinity
85 can be good candidates for enhancing water flux (i.e. dilution effect).

86

87 FDFO can be used as a stand-alone process or coupled with a post-treatment process such as
88 RO and nanofiltration (NF) for draw solution recovery and water purification. In the latter case,
89 the post-treatment process provides further purification of the product water. For example,
90 Phuntsho et al. (Phuntsho et al. 2013 a) demonstrated that NF as post-treatment was found to
91 be more effective in reducing the nutrient concentrations in the final product. Including this,
92 different approaches were proposed in our previous investigation to mitigate the nutrient
93 concentrations in the final product fertiliser such as mixing with fresh water at the final stage
94 (Phuntsho et al. 2012), hybrid FO process for treating wastewater treatment (Phuntsho et al.
95 2012) and applying additional pressure on the feed side during the FO operation (Sahebi et al.
96 2015).

97

98 The pressurization of the FS of the FDFO process can offer a range of potential benefits over
99 the limitations of the stand-alone FDFO system such as low water flux and high reverse salt
100 flux. The pressure assisted FDFO system can thus take an advantage of the synergetic effect of
101 the driving force to improve the permeate flux and thus further dilution of the DS. Chekli et al.
102 (Chekli et al. 2017) recently reported that integrating FDFO with pressure assisted osmosis
103 (PAO) could provide an insight into an opportunity for a cost-effective FDFO process and also
104 assessed the applicability of the FDFO process to yield an irrigation solution for a hydroponic
105 grow system, which is a widely applied technique for growing plants in the water/fertiliser
106 solution. However, no study has directly evaluated the techno-economic feasibility of FDFO
107 in the reuse of real wastewater effluent to a desirable quality for greenwall plants growth (i.e.
108 vertical farming).

109

110 This work examined the techno-economic feasibility of the FDFO process for irrigation to
111 greenwall plants. This includes short and long-term operations of the FDFO process with real
112 urban wastewater of different qualities (i.e. primary and secondary effluents) as a feed solution
113 candidate and commercial fertiliser as a draw solution to demonstrate its technical feasibility.
114 The effect of a hydraulic pressure on the FS (i.e. pressure assisted osmosis, PAO) was also
115 evaluated as an alternative way of reducing the final diluted fertiliser concentration. In addition,
116 the economic performance of the pressure assisted FDFO (PAFDO) process was delineated to
117 provide a better understanding of the applicability of this technology and its implications to
118 make it economically feasible.

119

120 **2 Materials and methods**

121 **2.1 Commercial fertiliser draw solution and real waste feed water**

122

123 The commercial fertiliser diamond blue (denoted as DB) used in this study as DS was obtained
124 from Campbells Fertilisers Australasia. The fertiliser DS was prepared to obtain 175 g/L as
125 total dissolved solids (TDS, pre-filtered with 0.45 μm filters), corresponding to an osmotic
126 pressure of 92.48 bar. The osmotic pressure of DB fertiliser was calculated using the
127 thermodynamic modelling software OLI Stream Analyser (OLI Systems Inc., USA). Table 1
128 shows the characteristics of the DB. The bench-scale experiments were conducted using
129 deionized water (DI water) as FS to elucidate the performance of the DB as DS in the FDFO
130 process.

131

132 The wastewater streams used in this study as FS were collected from the Central Park
133 Wastewater Treatment Plant (CPWTP) in Sydney, New South Wales, Australia. The CPWTP
134 consists of a screen mesh, a membrane bioreactor (MBR) followed by an ultraviolet (UV)
135 disinfection unit, RO system and chlorine contact, before finally being stored in the treated
136 water storage tank for reuse. Three types of wastewater streams with different qualities were
137 evaluated for their performances as a FS candidate in the FDFO process: raw wastewater, MBR
138 supernatant and MBR effluent. The characteristics of the wastewater streams are presented in
139 Table 2.

140

141

142 **Table 1.** Characteristics of the commercial fertiliser diamond blue.

Parameters	Diamond Blue Fertiliser
Electrical conductivity (EC, $\mu\text{S}/\text{cm}$)	150.4
Total dissolved solids (TDS, g/L)	175.0
pH	3.91
Total organic carbon (TOC, mg/L)	1,102.5
Osmotic pressure (bar)	92.48
Total Nitrogen (TN, mg/L)	72,000
NO_3^- -N (mg/L)	22,700
NH_4^+ -N (mg/L)	16,900
Total Phosphorus (TP, mg/L)	7,300
SO_4^{2-} (mg/L)	17,000
K^+ (mg/L)	26,440
Na^+ ((mg/L)	3,090
Mg^{2+} (mg/L)	3,860
Ca^{2+} (mg/L)	470

143

144 **Table 2.** Central Park Wastewater characteristics used in this study as FS.

Parameters	Raw wastewater*	MBR supernatant*	MBR effluent
EC ($\mu\text{S}/\text{cm}$)	1299.0	820.0	759.0
pH	7.90	7.38	7.50
TDS (mg/L)	646.0	357.0	336.0
Turbidity (NTU)	63.6	1.30	0.39
Osmotic pressure (bar)	0.307	0.226	0.194
NO_2^- (mg/L)	0.19	0.08	0.10
NO_3^- -N (mg/L)	0.20	2.90	3.30
NH_4^+ -N (mg/L)	65.6	1.30	1.90
TP (mg/L)	22.4	7.0	3.0
SO_4^{2-} (mg/L)	38.0	38.0	45.1
K^+ (mg/L)	25.4	20.45	18.24
Na^+ ((mg/L)	142.4	122.5	113.0
Mg^{2+} (mg/L)	9.25	8.0	5.56
Cl^- (mg/L)	72.0	67.0	21.54

145 * Pre-treated only by sedimentation to collect the supernatant.

146

147 2.2 Forward osmosis experimental procedure

148

149 A bench-scale crossflow FO experimental process is shown schematically in Fig. 1. Low-
 150 pressure variable speed gear pumps (Cole Palmer, USA) were installed to circulate the feed
 151 and draw streams. Each pump was connected to a membrane cell (2.6 cm width, 7.7 cm length,
 152 and 0.3 cm depth). A thin film composite (TFC) FO membrane supplied by Toray Industry Inc,

153 which is made of a polyamide active layer deposited on a polysulfone support layer, was used
154 for all FO experiments. The intrinsic properties of the TFC membranes are pure water
155 permeability (A) of $2.47 \text{ Lm}^{-2}\text{h}^{-1}\text{bar}^{-1}$ and NaCl rejection (B) of 96%. The temperatures of the
156 feed and draw solutions were maintained at $25\pm 0.5 \text{ }^\circ\text{C}$ using a heater/chiller system. The cross-
157 flow velocity for the feed and draw streams was circulated at 10.68 cm/s in counter-current
158 configuration. The water flux was determined by changing the weight of the draw tank
159 collected on the digital weighing scale connected to a computer for the data recorder. During
160 all FO experimental work, both FS and DS were recirculated back to their respective reservoirs.
161 The initial volume of FS and DS solutions was 1 L for each short-term experiment that lasted
162 for 5 hr. The long-term operations were conducted for 5 days with the initial volume of FS and
163 DS of 5 L and 200 mL, respectively.

164

165 The reverse salt flux (RSF) was investigated either by measuring the electrical conductivity
166 (EC) using a multi-meter (Hach, Germany) or by analyzing the major anions and cations using
167 inductively coupled plasma mass spectrometry (ICP-MS) and microwave plasma-atomic
168 emission spectroscopy (MP-AES). Nutrient concentrations were also evaluated with a
169 spectrophotometer (Spectroquant NOVA 60; Merck, Germany). A total organic content
170 analyser (TOC analyser, Analytikjena, Jena, Germany) was used to measure the TOC of the
171 FS and DS. The turbidity of wastewater was measured with a 2100P Portable Turbidimeter
172 (Hach, USA).

173

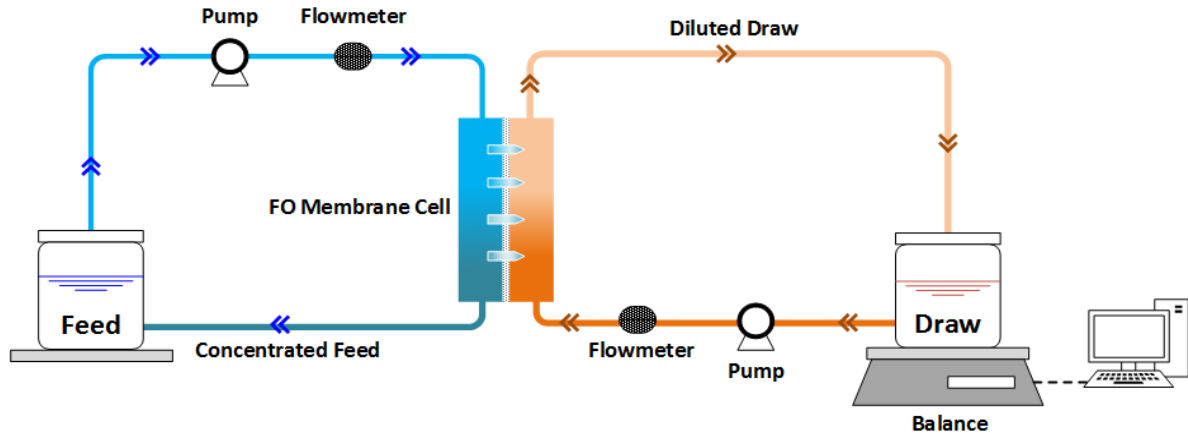


Fig. 1. A bench-scale FO experimental set-up and flow direction.

174

2.3 Determination of water flux, reverse salt flux and specific reverse salt flux

175

176

177 Water flux (J_w , $\text{Lm}^{-2}\text{h}^{-1}$, LMH) across the membrane was calculated automatically every 5

178 minute from the increase in DS weight recorded by a digital balance, on which the DS tank

179 was placed, and connected to PC for real-time data collection.

180

$$181 \quad J_w = \Delta W_D / (S_m * \Delta t) \quad (1)$$

182

183 where ΔW_D is the weight change of the draw, S_m the effective membrane surface (m^2) and Δt

184 the time interval (hr).

185

186 The dilution factor DF (i.e. how many times the DS is diluted) of the DS is determined by the

187 following equation:

188

$$189 \quad DF = (V_{D,i} + \Delta V_D) / V_{D,i} \quad (2)$$

190

191 where $V_{D,i}$ (L) is the initial DS volume and ΔV_D (L) is the increase in draw volume over time.

192

193 RSF represents the amount of draw solutes that pass across the membrane to the feed side in a
194 unit membrane area and in a unit operating time. RSF J_s ($\text{gm}^{-2}\text{h}^{-1}$) was calculated by monitoring
195 the increase in the electrical conductivity of the FS using DI water as feed with a conductivity
196 and pH meter (HACH, Germany) connected to the computer for data logging.

197

$$198 \quad J_s = (F * \Delta(EC_F * V_F)) / (S_m * \Delta t) \quad (3)$$

199

200 where F is a conductivity calibration factor for the conversion between conductivity and
201 concentration and $\Delta(EC_F * V_F)$ the feed conductivity differential per feed volume change
202 ($\mu\text{S} * \text{L}/\text{cm}$). It should be noted here that ‘ F ’ is valid for DI water as feed water. When the feed
203 water contains various components, the detailed characteristics of the feed water need to be
204 conducted. The amount of fertiliser lost per volume of water that permeates the membrane is
205 represented by the specific reverse solute flux (SRSF) (g/L) calculated dividing J_s by J_w .

206

207 **2.4 Economic analysis**

208

209 The scope of this study was to conduct an economic feasibility study that compares an existing
210 UV-RO disinfection system with a FDFO process in terms of the capital and operational
211 expenses in Australian dollar (i.e. CAPEX and OPEX in AUD) by considering different FDFO
212 process configurations. Several assumptions have been considered as follow:

- 213 ■ The CAPEX cost includes the cost of FO and RO modules in the FDFO and UV-RO
214 systems respectively. Other components such as pipeline, pumps and valves are not
215 included in this study due to its minor contribution to the total cost (Zhou et al. 2014). The

216 FO element cost was assumed to be the same as the RO element cost of \$700 with a lifetime
 217 of 3 years for RO and 7 years for FO. It has to be noted here that the lifetime of the FO
 218 element was assumed longer than that of the RO element since fouling propensity and
 219 cleaning frequency of RO are more significant than FO due to the use of high pressure (Lee
 220 et al. 2010). A unit cost of \$1,000 for one pressure vessel was assumed for both FO and
 221 RO processes.

222 ■ The annualized CAPEX cost (\$/yr, CAPEX_a) was determined at an interest rate of 6% (i.e.
 223 i) and a plant availability of 0.95 for a 20-year plant lifetime (i.e. n) (Kim et al. 2018). The
 224 CAPEX_a cost in \$/yr is therefore calculated based on the following equation:

$$CAPEX_a = Total\ CAPEX\ cost \frac{i(1+i)^n}{(1+i)^n - 1}$$

225
 226 ■ The annual OPEX cost (\$/yr) comprises the energy consumption, membrane maintenance,
 227 and chemical consumption costs. The annual energy cost was estimated at an electricity
 228 cost of 0.29 \$/kWh (Kim et al. 2017).

229 ■ Based on the real capacity of the Central Wastewater Treatment Plant, all configurations
 230 were set at 400 m³/day of product water.

231 ■ In the case of PAO application, the applied pressure on the feed side was assumed to be 2
 232 bar because higher operating pressure could result in additional costs and more severe
 233 fouling (Blandin et al. 2015, Kook et al. 2018). However, for a sensitivity analysis, the
 234 applied pressure was assumed to be varied from 2 to 6 bar.

235 ■ The total cost (\$/yr) is the sum of CAPEX_a and OPEX costs. The total product cost (\$/m³)
 236 can be therefore calculated from the following equation:

$$Total\ product\ cost\ (\$/m^3) = \frac{Total\ cost\ \left(\frac{\$}{yr}\right) \times Plant\ lifetime\ (yr)}{Plant\ capacity\ \left(\frac{m^3}{day}\right) \times 365 \times Plant\ availability}$$

237

238

239 **2.5 Process description**

240 **2.5.1 Conventional recycled water plant**

241

242 In Central Park WTP, the primary water (i.e. after screen process and biological processes) is
243 first filtered through two membrane tanks and both contain one GE 500 membrane cassette
244 with a surface area of 1099.8 m² followed by UV disinfection unit and RO. Two UV
245 disinfection units provide disinfection of the filtered water. The setpoint of UV dosage rate in
246 the system is 250 J/m². The RO system operates as a two stage-one pass process, which
247 comprises six RO vessels containing four spiral wound membrane elements (BW30-400, Dow
248 Filmtech Chemicals, USA) in each pressure vessel. The RO system is designed based on a
249 design recovery of 80%, to produce 400 m³/day of permeate. In the first stage, two sets of two
250 RO vessels operate in parallel while two pressure vessels operate in series in the second stage.
251 A booster pump between the two stages is required to compensate for osmotic pressure increase.
252 Based on the current plant design, the RO system simulation and cost analysis to produce 400
253 m³/day were conducted using WAVE simulation software (Water Application Value Engine,
254 Dow Filmtech Chemicals, USA). A schematic diagram of the conventional WTP and flow
255 directions is presented in Fig. 2. Calculations of the CAPEX and OPEX for UV-RO system are
256 based on the percentage contribution to the total annual cost adapted from the literature
257 (Holloway et al. 2016) and the results achieved from the WAVE analysis.

258

259 **2.5.2 FDFO system**

260

261 A full-scale FDFO system was simulated using lab-scale FDFO experimental results, equations
262 developed by Deshmukh et al. (Deshmukh et al. 2015) and a mass balance relationship in the

263 process. The feed water for the FDFO process was considered an MBR effluent with an osmotic
264 pressure of 0.2 bar as shown in Table 2, determined by the results obtained from the lab-scale
265 FDFO experiments. As mentioned above, the draw solution was commercial fertiliser with
266 TDS of 175 g/L and an osmotic pressure of 92.48 bar (Table 1). The FO system was designed
267 to produce 400 m³/day of the product permeate. Membrane fouling was indirectly considered
268 through physical cleaning and membrane replacement intervals. A schematic diagram of a
269 hybrid PAO and FDFO process and flow directions is presented in Fig. 2.

270

271 The reliability of economic impact assessment is highly dependent on the selected background
272 data due to a large number of input parameters including permeation flux, total membrane area
273 required and membrane element and pressure vessel costs. (Kim et al. 2018). Among the
274 various input parameters, the average permeation fluxes play a crucial role in economic
275 feasibility of FO (Blandin et al. 2015, Kim et al. 2017). In addition, operating in PAO mode
276 causes the advanced driving force thus resulting in a significant saving in total water product
277 cost even with increased energy requirements (Blandin et al. 2015, Chekli et al. 2017, Sahebi
278 et al. 2015). However, it is important to determine a certain dilution stage to apply PAO mode
279 in the FDFO process because this can influence the economic viability of the PAFDO system.
280 Therefore, a sensitivity analysis was carried out based on two different approaches: (i) PAO
281 mode in different DS dilution stages in the FDFO process and (ii) PAO mode with three
282 different applied pressures (2, 4, and 6 bar). The results were finally compared with the
283 conventional wastewater reuse plant to evaluate its economic feasibility.

284

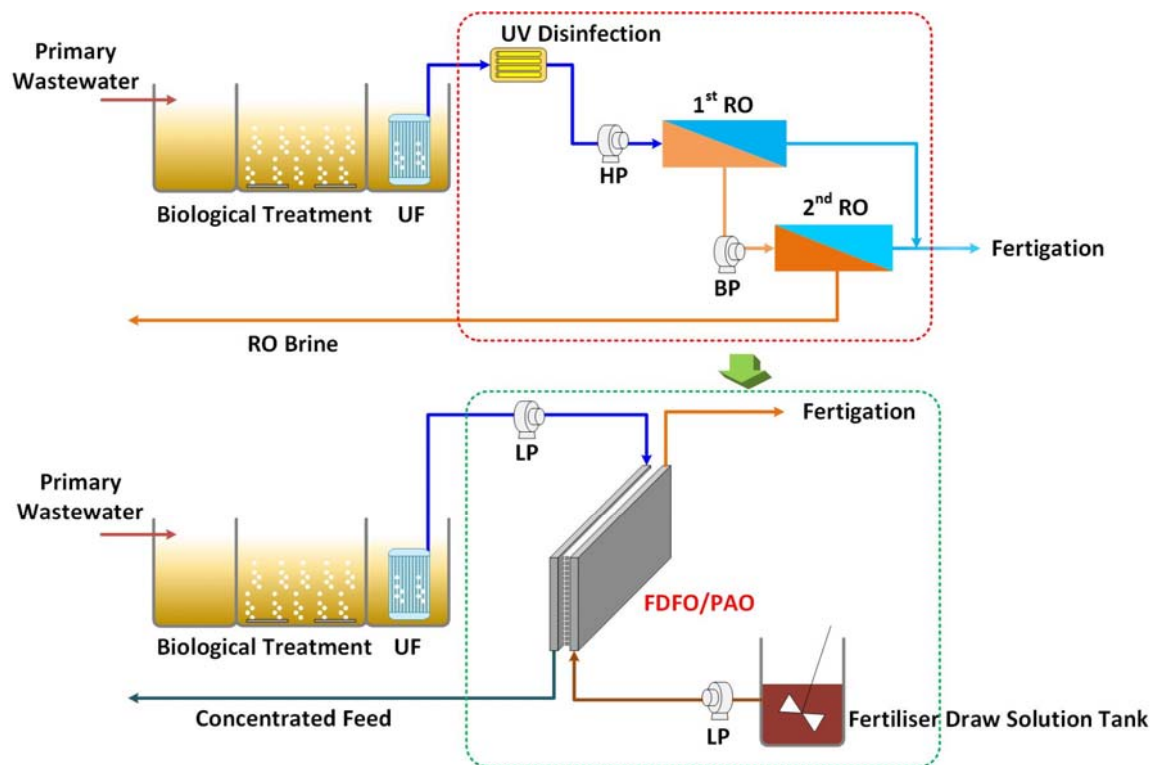


Fig. 2. Schematic diagram of MBR-UV-RO (Central Park Wastewater Treatment Plant) and MBR-PAFDO considered in this study. Plant capacity: 400 m³/day. HP: High pressure, BP: booster pump, LP: low pressure pump.

285

286 3 Results and discussion

287 3.1 FDFO performance evaluation

288 3.1.1 Short-term FDFO operation with DI water as FS

289

290 The short-term FO tests were conducted to demonstrate the ability of the DB fertiliser DS using
 291 DI as FS. The average water flux as a function of the concentration of the fertiliser DS is
 292 presented in Fig. 3 (a). Results showed that the water flux increased non-linearly with the
 293 increase in the DS concentrations, which is similar to our earlier study with ammonium
 294 sulphate fertiliser (Sahebi et al. 2015). This non-linear correlation between water flux and DS
 295 concentration (i.e. osmotic driving force) can be attributed to the severity of dilutive internal
 296 concentration polarisation (ICP) that significantly reduces the effective osmotic pressure
 297 difference across the FO membrane (Cath et al. 2006).

298

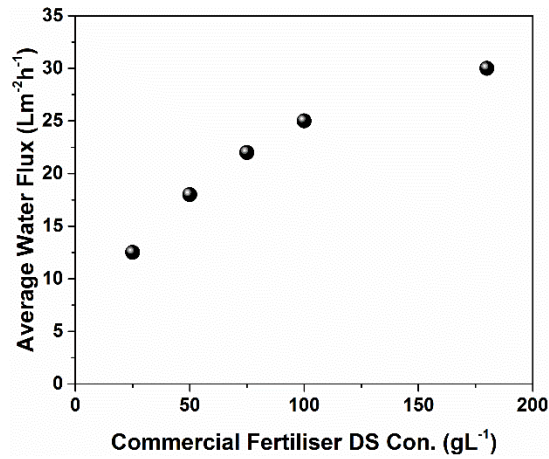
299 In addition, the RSF (J_s) for the commercial fertiliser was determined as explained in Section
300 2.3. The ratio of RSF to water flux (J_s/J_w , SRSF) was therefore found for the major nutrients
301 in the DB fertiliser and shown in Fig. 3 (b). The result shows that PO_4^{3-} , Mg^{2+} and SO_4^{2-} had
302 SRSF comparatively much lower than K^+ , NO_3^- and NH_4^+ , which can be explained by their
303 larger hydrated radius and thus lower reverse diffusion toward the FS. In fact, monovalent ions
304 are more subjected to reverse permeation than multivalent ions.

305

306 Moreover, larger-sized hydrated anions such as PO_4^{3-} and SO_4^{2-} diffuse less across the semi-
307 permeable membrane because of electrostatic repulsion forces (less than 0.1 g/L). The
308 difference in SRSF between potassium (K^+) and ammonia (NH_4^+) can be explained by their
309 concentration in the initial fertiliser DS. Lower solute concentration in the DS can result in
310 lower SRSF and vice versa. As shown in Table 1, the concentration of ammonia (i.e. 16.9 g/L)
311 in the commercial DB fertiliser was lower than the one of potassium (i.e. 26.44 g/L).
312 Consequently, the reverse permeation of NH_4^+ was lower than for K^+ . The ratio J_s/J_w plays an
313 important role in determining the draw solute loss during FO operation. This is directly related
314 to the draw solute replenishment cost. The recent study conducted by Chekli et al. (Chekli et
315 al. 2017) demonstrated that the enhanced water permeability reduces the RSF across the
316 membrane. For instance, under PAO mode (2 bar applied pressure), the reverse diffusion of
317 NH_4^+ was reduced by 80% and that of K^+ was reduced by more than 90%. The result clearly
318 showed that the RSF can be reduced by integrating PAO in the FDFO process. Therefore, in
319 this study, the DS replenishment cost was not considered for economic evaluation of the
320 PAFDO process as it is a minor contributor to the total cost of the process.

321

(a)



(b)

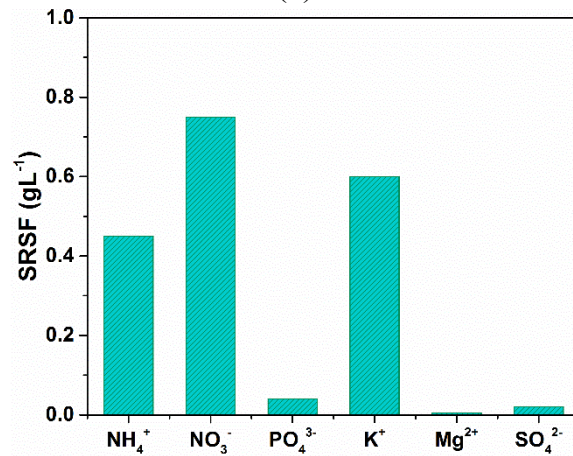


Fig. 3. (a) Experimental water flux of each concentration of commercial DB fertiliser (25, 50, 75, 100, and 175 g/L) and (b) specific reverse solute flux of the commercial DB fertiliser of 175 g/L. Feed and draw flow rate: 500 mL/min. The temperature of feed and draw sides: 25 °C. Feed: DI water.

322

323

3.1.2 Long-term FDFO operation with different wastewater streams as FS

324

325 The key parameter for FDFO technology is the draw dilution factor which can be calculated
 326 based on Equation (2) presented in Section 2.3. The principle of the osmotic equilibrium
 327 between the FS and the DS limits this pivotal factor during the FO process since the initial FS
 328 concentration governs the final DS concentration (i.e. final osmotic equilibrium), which shows
 329 direct implications for the end use of the final product. A series of long-term FDFO
 330 experiments were carried out to identify the optimal FS with the goal of diluting as much as
 331 possible the DS (i.e. maximum dilution factor).

332

333 Fig. 4 shows the variations of the water flux and dilution factor of the fertiliser DS when
334 different wastewater streams are used in the FDFO process. Fig. 4 (a) shows that the water flux
335 with MBR supernatant and effluent appeared quite similar for operation time, demonstrating a
336 consistent performance of the FDFO process under each long-term test. This can be seen that
337 the flux decline with MBR supernatant and MBR effluent mainly caused by DS dilution effect
338 rather than membrane fouling.

339

340 However, the water flux with raw wastewater is considerably lower than that with other streams
341 (i.e. the sharper flux decline). Such flux decline with raw wastewater in Fig. 4 (a) was expected
342 since the feed water used for the FDFO process had higher turbidity of 63.6 NTU (Table 2)
343 and much lower dilution factor compared to the others. It is important to note that although
344 MBR effluent and supernatant showed lower turbidity, organic compounds in the feed streams
345 may cause severe fouling, thus resulting in flux decline. Meanwhile, there was a sharp flux
346 decline in the first 10 h and this is mainly attributed to the effect of DS dilution while after 10
347 h operation, the flux decline with raw wastewater was more severe than others. This indicates
348 the occurrence of fouling caused by high turbidity in the feed water at the initial stage of the
349 operation.

350

351 In addition, Fig. 4 (b) presents the dilution factor with the operation, indicating the maximum
352 dilution factor achieved during the same operation time. Corresponding to the water flux trend,
353 when raw wastewater is used as FS, it shows the lowest DS dilution factor (DF 7.07) and
354 followed by the MBR supernatant and MBR effluent (11.20 and 13.11, respectively). In general,
355 a higher dilution factor is expected to have a higher flux decline; however, the performance
356 using raw wastewater shows the lowest dilution factor but the highest flux decline among three

357 feed solutions (Fig. 4). This indicates that the flux decline with raw wastewater was mainly due
358 to fouling occurrence on the membrane surface rather than the dilution effect. The lowest
359 diluted DS concentration (i.e. the highest dilution effect) can be achieved when the MBR
360 effluent is used as FS and thus showing that the MBR effluent is the best FS candidate for
361 FDFO application.

362

363 A recent study by Sahebe et al. (Sahebi et al. 2015) proved the application of PAO in the FDFO
364 process to improve the dilution of the fertiliser DS. Results from this investigation indicated
365 that the PAO application can provide further dilution of the final fertiliser DS due to increased
366 water flux thereby achieving the fertigation standard. Consequently, PAO could potentially
367 remove the need for additional treatment.

368

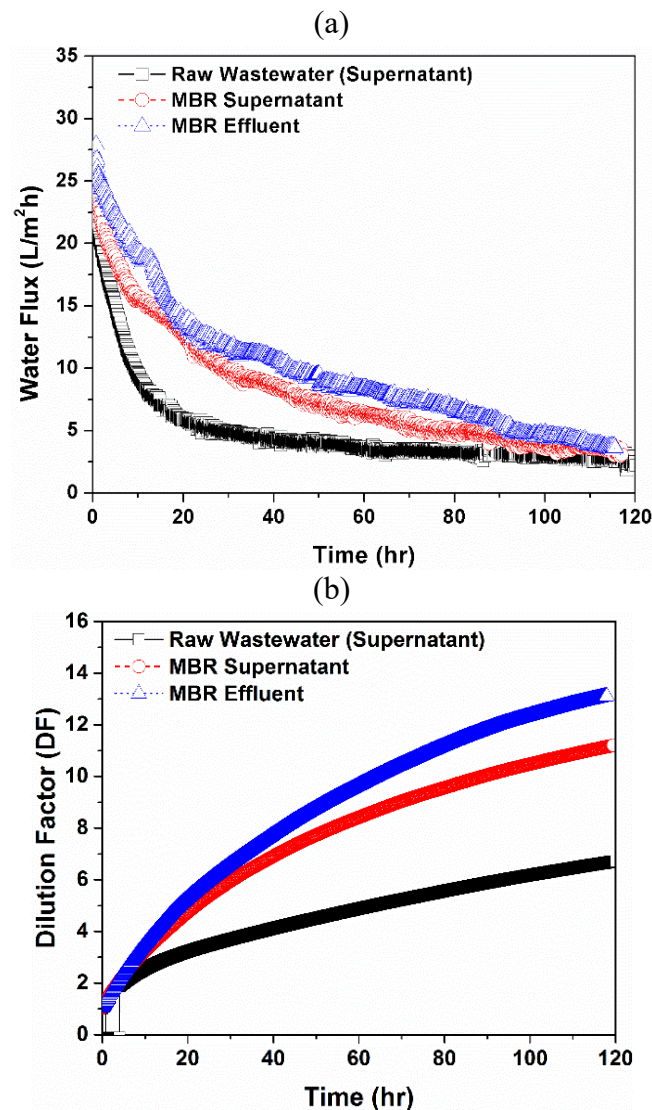


Fig. 4. The effect of wastewater quality (feed solution) on the water flux (a) and the dilution factor (b) as a function of the operation time. The initial FS and DS volumes were 5 L and 200 mL, respectively. Operation time: 5 days.

370

371 3.1.3 Suitability of the final FDFO nutrient solution for direct fertigation of 372 greenwall plants

373

374 Based on the results of the short- and long-term experiments with different FSs (Section 3.1.1
375 and 3.1.2), the final long-term FDFO experiment was conducted using the MBR effluent as FS
376 and DB fertiliser with an initial concentration of 175 g/L as DS. When the MBR effluent was

377 used as the FS, it was observed that the fouling effect on the flux decline was not significant
378 compared to the DS dilution effect. As mentioned earlier in Section 2.4, the final permeate
379 water quality should be less than 1 g/L. Thus, it is obvious that the commercial fertiliser
380 solution needs to be diluted 175 times to fulfill the requirement. The results in terms of water
381 permeation and dilution factor during the operation of FDFO process are presented in Fig. 5.
382 The operational parameters and the schematic diagram of the experimental procedure for this
383 long-term FDFO operation can be found in Figure S1 in the supplementary information (SI).

384

385 The osmotic pressure difference between the DS and the FS became lower and lower which
386 decreased the driving force across the FDFO process. It is worth noting that after 8-day
387 operation, the water flux was almost zero meaning that the draw solution could not be further
388 diluted because the osmotic equilibrium between the FS and the DS occurred (Phuntsho et al.
389 2014). At the end of the operation, the final DS concentration was reached to 3.83 g/L (total
390 dilution of around 84.41). This results in the final fertiliser solution that contain insufficient
391 nutrients for irrigation purpose (i.e. less than 1 g/L total dissolved solids).

392

393 In this study, PAO has been therefore considered as an integrated process to FDFO (referred
394 to as PAFDO). Operating the PAO mode in FDFO can provide a trade-off between savings in
395 total membrane area required (i.e. CAPEX cost) and the increased energy consumption (i.e.
396 OPEX cost). Blandin et al. (Blandin et al. 2015) already reported the effect of hydraulic
397 pressure on reducing the total membrane area. The results also showed that at the same recovery
398 rate the required membrane area could be significantly reduced by increasing the applied
399 pressure and thus savings in CAPEX cost. Recently, Kook et al. (Kook et al. 2018) investigated
400 the optimum operating condition of PAO for PAO-RO hybrid system at a pilot scale level and

401 its economic potential for wastewater purification and seawater dilution. From the practical
 402 aspect, the results showed that the PAO-RO hybrid system can be economically favourable if
 403 two FO membrane elements are connected in series in a housing. So, this is the first study to
 404 investigate an optimum point to apply PAO in the fertiliser driven FO application and its
 405 economic effect. The following section will discuss the optimum PAFDO process
 406 configuration for osmotic dilution of fertiliser draw solution using wastewater to reach
 407 economic viability.
 408

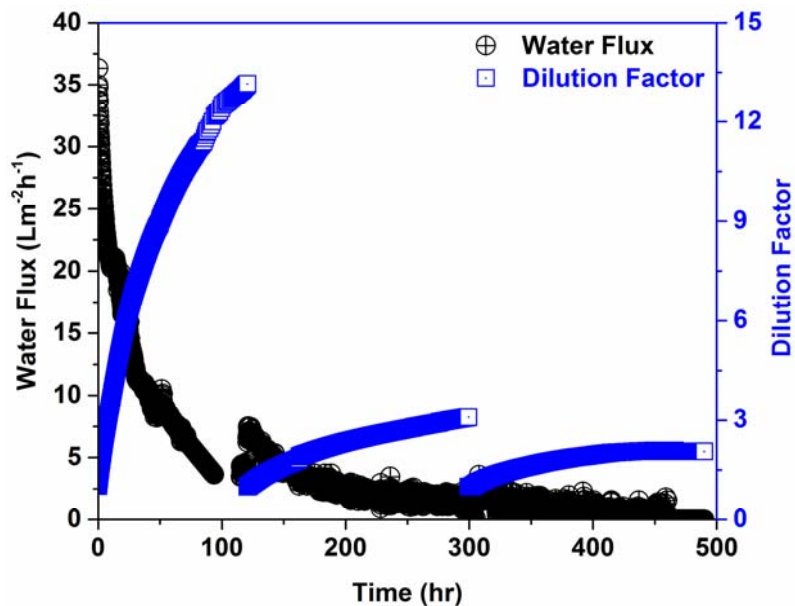


Fig. 5. Water flux and dilution factor with operating time. Experimental conditions: Commercial DB fertiliser as DS and MBR effluent as FS, initial FS and DS volumes were 15 L and 0.2 L respectively, and operation time: 500 hr.

409

3.2 Economic evaluation

410

3.2.1 MBR-UV-RO disinfection system

411

412

413 Fig. 6 (a) displays that the UV-RO accounts for 40% of the total energy consumption of the
 414 plant with 25% from RO process and 15% from the UV disinfection step. Fig. 6 (b) shows the
 415 RO and UV systems made up the largest proportion for all three costs. Compared to the MBR,

416 the RO and UV systems had a 28% higher OPEX and a 10% higher CAPEX. Consequently,
417 the contribution of the RO and UV units to the total cost of the plant was significant, around
418 60%.

419

420 For OPEX, the RO cost ($\$0.321/\text{m}^3$) was significant, followed by MBR and UV, $\$0.277/\text{m}^3$
421 and $\$0.167/\text{m}^3$ respectively. For CAPEX, the MBR cost ($\$0.302/\text{m}^3$) was highest and followed
422 by RO and UV disinfection units ($\$0.242/\text{m}^3$ and $\$0.130/\text{m}^3$ respectively). As expected, the
423 major factors responsible for such high OPEX cost of the plant are the energy consumption of
424 the RO and UV units (Fig. 6 (a)). From the total water cost ($\$1.439/\text{m}^3$), the most economically
425 feasible scenario would be ultimately made by reducing the operating cost of RO and UV units
426 ($\$0.86/\text{m}^3$) under given plant operating conditions. Hence, one way of reducing the operating
427 cost of the plant is to replace UV-RO unit with a low-energy technology, like FDFO. Several
428 FDFO configurations have been therefore proposed and evaluated in the following section.

429

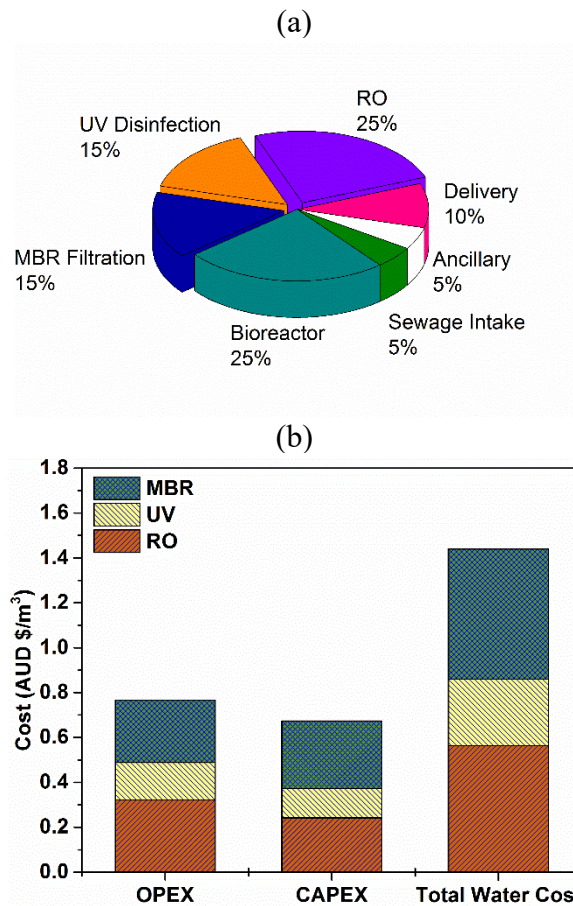


Fig. 6. (a) Energy consumption breakdown in Central Park wastewater treatment plant and (b) CAPEX and OPEX costs contributed to the total water cost of the plant.

430

431 3.2.2 FDFO system: sensitivity analysis

432 A sensitivity analysis is conducted on two approaches that can be used to make the PAFDO
 433 process economically favourable as mentioned above in Section 2.5.2; (i) PAO mode in
 434 different DS dilution stages in the FDFO process and (ii) PAO mode with three different
 435 applied pressures (2, 4, and 6 bar).

436

437 The benchmark for the UV-RO system is \$0.86/m³ and thus the cost of PAFDO should be
 438 lower than that of UV-RO. Fig. 7 shows the estimation of capital and operational costs for
 439 PAFDO as a function of FDFO dilution factor based on the results of the long-term experiments
 440 (as shown in Fig. 5). In the figure, “FDFO” refers to a FDFO stand-alone system while

441 “PAFDO (DF)” refers to a system in which PAO is integrated at a certain DS dilution factor of
442 5, 10, 20, 50 and 80. These values correspond to the FO water flux of 23.04, 14.89, 11.34, 5.57,
443 and 4.37 LMH, respectively. It was also assumed that once the PAO mode is applied at a certain
444 point in FDFO, the DS is continuously diluted to reach the final DS concentration of less than
445 1 g/L TDS, irrigation standard (Bauder et al. 2011).

446

447 Fig. 7 clearly shows that the FDFO stand-alone process had a 23% lower total water cost
448 compared to the UV-RO process. When PAO was applied at different dilution factors in the
449 FDFO process, the total water cost of the PAFDO was shown to be lower than that of the FDFO
450 stand-alone, except when applying PAO at the dilution factor of 80. For example, when PAO
451 is applied at a 5-fold DS dilution stage in the FDFO, the cost of PAFDO showed the lowest
452 cost at \$0.293/m³ among PAFDO configurations. However, with increasing the FDFO dilution
453 factor from 5 to 80, the total cost of the PAFDO system was significantly increased to
454 \$0.920/m³, which is 6.5% higher than the UV-RO. This increase is mostly because the FDFO
455 average water flux in the FDFO process became lower as increasing the dilution of the fertiliser
456 DS and thus increasing the total membrane area required (Blandin et al. 2015, Phuntsho et al.
457 2014). This result indicates that in order to make the PAFDO process economically favorable
458 compared to the UV-RO (\$0.86/m³), the PAO mode should occur before fertiliser DS dilution
459 reaches to 80.

460

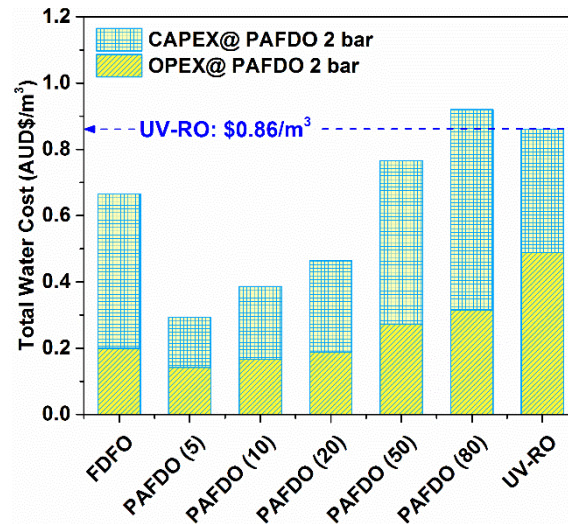


Fig. 7. Total water cost per m³ of water produced for FDFO stand-alone and PAFDO with different FDFO recovery rates and for the existing UV-RO. PAFDO (DF) refers to PAO application at different DS fertiliser dilution factors with 2 bar applied.

461

462 From the results presented in Fig. 7, coupling FDFO with PAO is confirmed to be a promising
 463 strategy to overcome current limitations of FDFO, and could help reduce the total water cost
 464 even though additional energy is required for feed pressurization. However, in such a hybrid
 465 system, an optimum trade-off between the total membrane area (i.e. CAPEX and OPEX costs)
 466 and the additional energy (i.e. OPEX cost) is important for a practical application of the
 467 PAFDO process (Blandin et al. 2015, Sahebi et al. 2015).

468

469 In order to determine the significance of applying PAO mode of FDFO operation and the FDFO
 470 average water permeation in relation to the total product cost for the PAFDO system, total
 471 water cost calculation for FDFO stand-alone and PAFDO was made over the DS dilution factor
 472 from 0 to 80 as shown in Fig. 8. Detailed calculations on the CAPEX and OPEX can be found
 473 in Figure S2 in the SI.

474

475 Fig. 8 (a) shows a strong response of the average permeation fluxes to the total water cost as
 476 the driving force inevitably decreased due to the dilution of the fertiliser and the concentration

477 of the wastewater stream. The FDFO cost is also seen to highly influence the total water cost
478 of the PAFDO. This indicates that requiring the higher DS dilution leads to a significant cost
479 increase of the FDFO process and thus it is essential to take into consideration a certain dilution
480 factor for applying PAO mode of the FDFO process (i.e. PAFDO). For instance, when the DS
481 was diluted 20-fold in the FDFO process, the water cost increases to 61% (from \$0.107/m³ to
482 \$0.273/m³). This is because an FDFO average water flux decreased from 34 L/m²h to 11.3
483 L/m²h (67% decrease). With further dilution of DS fertiliser (i.e. a 40-fold dilution), the cost
484 of the FDFO further increased to \$0.503/m³ considering the average water flux of 5.9 L/m²h.

485

486 Compared to the FDFO, although the PAO cost increases with increasing the dilution stages in
487 the FDFO process, its contribution to the PAFDO cost is not significant. For example, when
488 the FDFO dilution increased from 20 to 40, the PAO cost increased from \$0.213/m³ to
489 \$0.255/m³ (i.e. 16.5% increase). This corresponded to the trend of the PAO average water flux
490 reduction, suggesting that the flux at a 20-fold DS dilution was 22.4 L/m²h while that at a 40-
491 fold DS dilution was 17.0 L/m²h (i.e. 24% decrease) shown in Fig. 8 (a). This again confirms
492 the importance of average permeation flux to the total water cost of the PAFDO process and
493 thus resulting in its economic sustainability.

494

495 These results indicate that when the DS dilution factor increased, the total cost significantly
496 increased due to the much lower average permeation flux. This indicates that the total water
497 cost of the PAFDO is highly influenced by dilution factor in the FDFO process. Therefore, a
498 suitable stage to apply the PAO mode in FDFO process should be below a 60-fold DS dilution
499 with considering a threshold flux of above 10 L/m²h.

500

501 To clarify the effect of the PAO application on the total cost of PAFDO, a sensitivity analysis
502 has been conducted based on three different hydraulic pressures of 2, 4, and 6 bar. This range
503 was considered because of concern for the possibility of membrane deformation when applying
504 a pressure higher than 6 bar (Blandin et al. 2013). The results of the sensitivity analysis are
505 further presented in Fig. 8 (b).

506

507 The increase in the total water cost of the PAFDO due to additional hydraulic pressure is higher
508 at lower DS dilution factor than at higher DS dilution factor. Specifically, as expected, when
509 applying the PAO at a 10-fold DS dilution stage in the FDFO, the water cost of the PAFDO
510 due to the applied pressure of 2 bar is $\$0.364/\text{m}^3$ while that of 6 bar is $\$0.395/\text{m}^3$ (7.9%
511 increase). However, it is notable that for higher DS dilution factors of 50 and 80 the lowest
512 total cost of the PAFDO was observed with the applied pressure of 4 bar. This can be explained
513 by the results presented in Fig. 8 (a). It has been shown that the contribution of the FDFO cost
514 to the PAFDO cost is more significant than that of the PAO. This is attributed to the possibility
515 that higher DS dilution in the FDFO process could require more FO membrane areas thus
516 increasing CAPEX and OPEX costs of the FDFO process.

517

518 Concerning this, as shown in Fig. 8 (b), when applying the PAO at a 20-fold DS dilution stage
519 in the FDFO, the PAFDO cost with 2 bar applied pressure became similar to that with 4 bar
520 applied pressure ($\$0.442/\text{m}^3$ and $\$0.441/\text{m}^3$ respectively). These results, therefore, demonstrate
521 that the trade-off between savings in CAPEX costs (i.e. reduced total membrane area) when
522 using pressure and the increased OPEX costs required for feed pressurization could occur at a
523 certain DS dilution stage or factor. Thus, this finding can help define optimal operating
524 conditions for an integrated FDFO and PAO system thus economically feasible.

525

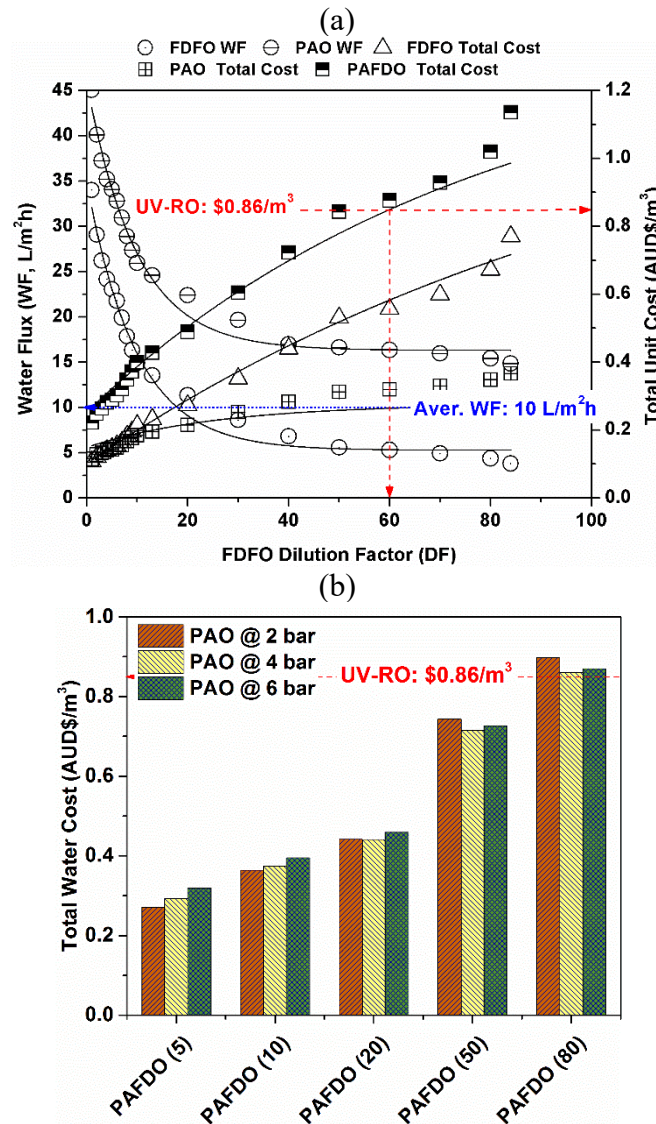


Fig. 8. Sensitivity analysis on (a) FDFO and PAO average water fluxes and total unit cost and (b) the applied pressure in PAO mode of FDFO operation related to the total water cost per m³ of water produced. PAFDO (5) refers to PAO application at the FDFO dilution factor of 5. Plant water production capacity: 400 m³/day.

526

527 **4 Conclusions**

528

529 Techno-economic analyses were conducted for a fertiliser driven forward osmosis process for
 530 commercial fertiliser dilution by biologically treated urban wastewater to produce irrigation
 531 water. Experimental investigations under the conditions used in this study showed that urban
 532 wastewater in particular the MBR effluent is the best feed stream for the FDFO application. In

533 addition, the commercial diamond blue fertiliser, whose nutrient composition is well balanced
534 for plants, turned out to generate an osmotic pressure similar to one of the inorganic salts well-
535 known for their good performance as DS. In that way, this study demonstrated that the FDFO
536 process is technically feasible with the potential to simultaneously reuse some amounts of
537 wastewater and produces water for greenwall irrigation. Economic evaluation results showed
538 that the integration of the PAO process can make PAFDO process economically favorable due
539 to the enhancement of the average FDFO water flux and thus reduction in CAPEX. However,
540 the sensitivity analysis proved that the average FDFO water flux and PAO application point in
541 the FDFO process play a crucial role in economic feasibility of the PAFDO system, indicating
542 that a coupling of higher average FDFO water flux ($\geq 10 \text{ L/m}^2\text{h}$) and PAO application at lower
543 DS dilution factor ($\text{DF} < 60$) in the FDFO process is recommended. Finally, PAO operation
544 with less than 4 bar applied is recommended since the energy penalty (i.e. increased OPEX
545 cost) caused by the PAO application of FDFO operation is compensated by improved water
546 permeation (i.e. reduced CAPEX cost). Although further work is required to validate the
547 application of PAO in FDFO process in terms of fouling behaviour, FO element arrangement
548 configurations, and a full-scale system design to control water flux and dilution factor, this
549 study offers a better understanding for the process engineers to design and operate the
550 collaborative process for the dilution of the fertiliser DS and the strategic management to lower
551 the wastewater reuse cost for greenwall fertigation.

552

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558

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