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1 **Validation of a cationic polyacrylamide flocculant for the harvesting fresh and seawater**  
2 **microalgal biomass**

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26

27 **Abstract**

28 A simple, efficient, and fast settling flocculation technique to harvest microalgal biomass was  
29 demonstrated using a proprietary cationic polyacrylamide flocculant for a freshwater (*Chlorella*  
30 *vulgaris*) and a marine (*Phaeodactylum tricornutum*) microalgal culture at their mid-stationary  
31 growth phase. The optimal flocculant doses were 18.9 and 13.7 mg/g of dry algal biomass for  
32 *C. vulgaris* and *P. tricornutum*, respectively (equivalent to 7 g per m<sup>3</sup> of algal culture for both  
33 species). The obtained optimal dose was well corroborated with changes in cell surface charge,  
34 and culture solution optical density and turbidity. At the optimal dose, charge neutralization of  
35 64 and 86% was observed for *C. vulgaris* and *P. tricornutum* algal cells, respectively. Algae  
36 recovery was independent of the culture solution pH in the range of pH 6 to 9. Algal biomass  
37 recovery was achieved of 100 and 90% for *C vulgaris* and *P. tricornutum* respectively, and over  
38 98% medium recovery was achievable by simple decanting.

39

40 **Keywords:** *Chlorella vulgaris*; *Phaeodactylum tricornutum*; Cationic polyacrylamide  
41 polymer; Biomass recovery; Algae harvesting.

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## 54 **1. Introduction**

55 Microalgal biomass is a renewable feedstock for the production of biochemicals for food  
56 additives and the biotechnology industry, animal feed, and biofuel (Vo et al., 2018; Vadivel et  
57 al., 2019; Jacob-Lopes et al., 2019; Poddar et al., 2018; Khalid et al., 2019; Ma et al., 2018).  
58 Microalgae production includes two major steps, namely culturing and harvesting. As progress  
59 has been achieved to optimize growth condition and nutrient requirements for effective  
60 microalgae culturing, the second step has emerged as a major bottleneck for cost-effective  
61 microalgal biomass production. Large-scale harvesting of microalgal biomass is challenging  
62 due to low cell concentrations (less than 1 g/L in a mature culture), small cell sizes (3–30  $\mu\text{m}$ ),  
63 stability of cell suspension, and complex culturing solution matrix (Vandamme et al., 2010;  
64 Zheng et al., 2012; Li et al., 2017a; Augustine et al., 2019; Muylaert et al., 2017). An estimation  
65 of 30% of the total production cost is attributed to microalgal harvesting (Şirin et al., 2012),  
66 which is arguably the most energy-intensive step in the production of microalgal-based  
67 materials.

68 Several microalgae harvesting techniques including membrane filtration, centrifugation and  
69 flocculation have been explored and reported in the literature (Vandamme et al., 2010; Şirin et  
70 al., 2012; Rashid et al., 2013; Bilad et al., 2012). Amongst them, flocculation is the most  
71 promising option for low cost microalgae harvesting, although the biomass recovery efficiency  
72 is often low (Vandamme et al., 2010; Şirin et al., 2012; Rashid et al., 2013; Pandey et al., 2019;  
73 Ummalyma et al., 2017). Common flocculants for microalgae harvesting can be, divided into  
74 three groups: i) inorganics such as ferric chloride and aluminium sulfate; ii) synthetic polymers  
75 such as polyacrylamide and polyethyleneimine; and iii) bio-agent flocculants such as fungi,  
76 protein, and chitosan (Vandamme et al., 2010; Rashid et al., 2013; Horiuchi et al., 2003; Li et  
77 al., 2017b). Inorganic flocculants are required at high doses (up to g/L) and increase the  
78 impurity of microalgal biomass, limiting its application and necessitating downstream  
79 processing (Şirin et al., 2012). Recently, the second and third groups of flocculants have been

80 extensively studied. Performances of these flocculants are often dependent on pH, long settling  
81 time and growth medium matrix (freshwater vs seawater). Vandanme et al. (2010) reported that  
82 cationic starch could effectively recover freshwater *Parachlorella* and *Scenedesmu* but not  
83 marine microalgae such as *Phaedactylum* and *Nannochloropsis*. Likewise, chitosan can be  
84 effective for harvesting marine microalgae at high dose (e.g. 40 mg/L or more) (Cheng et al.,  
85 2011) . The culture media matrix (i.e. ionic strength, cell structure of fresh and marine  
86 microalgae) influences the efficiency of previous flocculants (Pandey et al., 2019; Roselet et  
87 al., 2015). Moreover, previous studies reported that a relatively higher dose of inorganic and  
88 polymer flocculant is needed for marine microalgae (Bilanovic et al., 1988; Fabrizi et al., 2010;  
89 Uduman et al., 2010). Given the performance of current available flocculants and the wide  
90 range of applications of marine microalgal-based products, it is essential to identify a versatile  
91 flocculant that can be used for both freshwater and marine microalgae, over a wide pH range  
92 and at low doses.

93 This study aims to validate the efficiency of a cationic polymer on the recovery of the  
94 freshwater (*Chlorella vulgaris*) and marine (*Phaeodactylum tricornutum*) microalgae. The  
95 polymer has been widely used in the water industry but has not been applied to harvest  
96 microalgae. A dose-response experiment was performed to determine the optimal polymer  
97 dose. Optical density removal, turbidity, zeta potential and biomass recovery were examined to  
98 evaluate flocculation efficiency and mechanisms.

## 99 **2. Materials and methods**

### 100 2.1 Microalgae strains and growth conditions

101 The freshwater green algae *C. vulgaris* (CS-41) was obtained from  
102 the Australian National Algae Culture Collection, CSIRO Microalgae Research (Hobart, TAS,  
103 Australia) and marine diatom *P. tricornutum* (CCMP 632) was obtained from the National  
104 Centre for Marine Algae and Microbiota (NCMA) (East Boothbay, ME, USA). They were  
105 maintained at the Climate Change Cluster (C3) culture collection at University Technology

106 Sydney in freshwater MLA media and f/2 (marine) media (Algaboost; Wallaroo, SA, Australia)  
107 using seawater collected from Sydney Harbour (salinity of 33-35 g/L) respectively. Seed  
108 cultures were grown up to early stationary phase in 1-L Schott's bottles, followed by 10-L  
109 carboys, bubbled with air in ~20 °C and ~100  $\mu\text{mol}/\text{m}^2/\text{s}$  light in a 16:8 hour light:dark cycle.  
110 *P. tricornutum* was harvested from the carboy, while *C. vulgaris* was further scaled up with a  
111 1/50 inoculation in a 400-L bag bioreactor using dechlorinated tap water with Jaworski medium  
112 (Fresh by Design; Moss Vale, NSW, Australia), in a temperature controlled room of ~20 °C  
113 and ~400  $\mu\text{mol}/\text{m}^2/\text{s}$  light in a 16:8 hour light:dark cycle. This seeding protocol was adapted to  
114 sequentially scale up the reactor volume, maintaining sufficient initial microalgal biomass  
115 (Pereira et al., 2018). For media under 10 L, the media were sterilized before inoculation by  
116 filtering the seawater through a Whatman 0.2  $\mu\text{m}$  filter, and both fresh and marine water was  
117 then autoclaved, followed by addition of filter sterilized stock media components through a  
118 Whatman 0.2  $\mu\text{m}$  filter. For the large scale bag, the water was sterilized with 100 mL 12%  
119 sodium hypochlorite, followed by 100 mL 2 M sodium thiosulphate. The algae were checked  
120 for pH twice a day, and if the pH reached above 9.4, the culture was sparged with CO<sub>2</sub> up to 3  
121 minutes. The rate of microalgae growth was monitored every day by measuring optical density  
122 at wavelengths of 680 and 730 nm for *C. vulgaris* and *P. tricornutum*, respectively (Nguyen &  
123 Rittmann, 2018). The harvesting experiments were performed with the microalgae culture at a  
124 mid-stationary phase.

## 125 2.2 Experimental set up for flocculation

126 A cationic polyacrylamide flocculant (FO3801) with high-charge (>80% charge) high-  
127 molecular weight (>15 Megadalton) (SNF Pty Ltd; Corio, VIC, Australia) was used in this  
128 study. A stock solution of the polymer (0.4% v/v) was prepared in Milli-Q water with  
129 continuous mixing at 100 rpm for 1 h and stored at room temperature and used within 1 day of  
130 preparation.

131 Microalgae cell suspensions at a mid-stationary growth phase (Section 2.1) were used for all  
132 harvesting experiments. Different polymer doses of 2.7 to 54 mg polymer/g biomass (dry  
133 weight) (i.e. 1 to 20 mg/L) were added in glass bottles containing *C. vulgaris*. Likewise,  
134 polymer doses of 1.97 to 39.4 mg polymer/g biomass (dry weight) (i.e. 1 to 20 mg/L) was used  
135 for *P. tricornutum*. The bottles were then gently mixed by hand for one minute and then allowed  
136 to settle for another minute. An aliquot (10 mL) of the culture in the bottle was pipetted from a  
137 height of one- and two-thirds from the bottom for evaluating the flocculation performance.  
138 The flocculation efficiency was calculated based on the change in the optical density at  
139 wavelength of 680 and 730 nm before and after each polymer addition, as shown in the  
140 following equation.

$$141 \quad \text{Flocculation efficiency (\%)} = \left( \frac{OD_i - OD_f}{OD_i} \right) \times 100$$

142 Where  $OD_i$  and  $OD_f$  is the optical density of the culture before and after flocculant addition.  
143 Each polymer dose was repeated three times for two different microalgae cultures.

144

### 145 2.3 Analytical methods

146 The optical density of microalgae medium before and after was measured by the UV  
147 spectrophotometer (UV 6000 Shimadzu; Ermington, NSW, Australia). The wavelengths were  
148 680 and 730 nm for *C. vulgaris* and *P. tricornutum*, respectively.

149 The biomass concentration was determined gravimetrically by drying the sample to a  
150 constant mass at 60 °C within 4 h. A 150 mL aliquot of microalgae cells suspension was filtered  
151 through a 1.1 µm pre-weighed glass fiber filter paper. The weight of the final filter paper was  
152 used to calculate the dry microalgal biomass. This protocol was also applied for the  
153 determination of biomass volume after flocculation. For marine *P. tricornutum* culture, Milli-  
154 Q water (100 mL) was used to rinse (i.e. salt removal) before drying.

155 Zeta potential of the microalgae solutions before and after flocculation was measured using  
156 a Zetasizer nano instrument (Nano ZS Zen 3600; Malvern, UK). The zeta potential values were  
157 measured using a 10 mL sample of cells suspensions (i.e. initial sample) and 10 mL of  
158 supernatant after flocculation. All samples were at pH 9.5.

159 The solution pH was measured using a pH/conductivity meter (Orion 4-Star Plus Thermo  
160 Scientific; Waltham, MA, USA). The pH adjustment was achieved by using 0.1 M NaOH and  
161 0.1 M H<sub>2</sub>SO<sub>4</sub>. Turbidity of the microalgae solution before and after flocculation was measured  
162 using a portable turbidity meter kit (Apera TN400; Columbus, OH, USA) with accuracy ±1%  
163 or 0.02 NTU.

164 Statistical analysis was performed in Microsoft Excel using Student's unpaired *t*-Test, with  
165 a two-tailed distribution.

### 166 **3. Results and discussion**

#### 167 3.1 Polymer dose optimisation

168 A dose-response relationship revealed the optimal flocculation efficiency at polymer dose  
169 of 18.9 and 13.7 mg/g dry biomass for *C. vulgaris* and *P. tricornutum* (i.e. equivalent to 7 mg/L  
170 for both freshwater *C. vulgaris* and marine *P. tricornutum* (Fig. 1)). The optical density OD<sub>680</sub>  
171 removal increased gradually from 44 to 90% with polymer dose of 2.7 and 18.9 mg/g in the *C.*  
172 *vulgaris* solution. A further increase in polymer dose up to 54 mg/g did not result in the  
173 improvement of optical density removal (Fig. 1). Similarly, the optical density removal  
174 increased from 82 to 99% with polymer dose of 1.9 to 13.7 mg/g in the *P. tricornutum* solution.  
175 At 39.4 mg/g, the OD<sub>730</sub> removal was 75%, which is statistically significantly lower than that  
176 at 13.7 mg/g (unpaired *t*-test, *p*-value < 0.05). The data in Fig. 1 suggest that polymer over-  
177 dosing can be counterproductive.

178 [FIGURE 1]

179 Results from Fig. 1 demonstrate that the proprietary high charge, high molecular weight  
180 cationic polyacrylamide can be used for both fresh and marine microalgae flocculation. The



181 optimal doses were 18.9 and 13.7 mg/g dry biomass for *C. vulgaris* and *P. tricornutum*,  
182 respectively. The mid stationary biomass were at 370 mg/L for *C. vulgaris*, and 508 mg/L for  
183 *P. tricornutum*. As such, the optimal dose is equivalent to 7 mg/L of culture solution for both  
184 species. Several previous studies have established 90% optical density removal as the  
185 benchmarking value for effective flocculation (Ma et al., 2018; Roselet et al., 2015; Uduman et  
186 al., 2010). In this study, 90 to 99% optical density removal was achievable for both freshwater  
187 and marine microalgae.

188 Although the optimal dose varies from flocculant to flocculant, a considerably lower optimal  
189 dose was obtained in this study compared to the literature. For example, the optimal dose of an  
190 organic chitosan flocculant reported by several previous studies (Augustine et al., 2019; Rashid  
191 et al., 2013) was 120 mg/L (equivalent to approximately 120 mg/g dry algae biomass), which  
192 is 6.3 times higher than the optimal dose for the *C. vulgaris* in this study. Flocculation efficiency  
193 of chitosan for the marine microalgae *P. tricornutum* was below 30% at a dose of 30 mg/L (Ma  
194 et al., 2018; Şirin et al., 2012).

195 Another indicator of the efficiency of cationic polyacrylamide is the settling time. In this  
196 study, effective flocculation was observed within 1 min of polymer dosing. This appears to be  
197 the fastest settling time in the literature (Augustine et al., 2019; Rashid et al., 2013; Pandey et  
198 al., 2019). Pandey et al. (2019) observed maximum flocculation after 40 to 60 min settling time  
199 using  $Al^{3+}$ ,  $Ca^{2+}$  and egg cell powder. The fast settling time achieved with the proprietary  
200 polymer in our study provides an opportunity to integrate the harvesting process into the  
201 microalgal culturing in a continuous system that likely enhances the commercialization of  
202 microalgal industry. In addition, Xiong et al. (2018) suggested that the polyacrylamide  
203 backbone of this proprietary polymer is non-toxic and can readily be hydrolyzed once dissolved  
204 in water. This property can potentially expand the usage of cationic polyacrylamide on stringent  
205 applications of the harvested microalgae (e.g. food additives and cosmetic reagents).  
206 Nevertheless, further study is necessary to assert these applications.

## 207 3.2 Flocculation mechanisms

208 Microalgae cell neutralisation is the main mechanism for the flocculation formation with  
209 cationic polyacrylamide (Fig. 2). The charges of growth media increased gradually with  
210 polymer levels. The *C. vulgaris* and *P. tricornutum* microalgae cells are negatively charged at  
211 -15.6 and -12.6 mV, respectively. The negative surface charge of microalgae cells in a culture  
212 suspension is induced by the carboxylic and sulfate functional groups on the microalgae surface  
213 (Ndikubwimana et al., 2015). When they are highly negatively charged, electrostatic repulsion  
214 maintains suspension amongst cells. When a cationic polymer is added, the electrostatic  
215 repulsion decreased (as indicated by zeta potential) (Fig 2), promoting flocculation. The zeta  
216 potential data corroborated with the optimal polymer dose (Section 3.1). At the optimal doses  
217 18.9 and 13.7 mg/g, zeta potentials changed from -15.6 to -5.6 (i.e. 64% change) and from -  
218 12.6 to -1.7 (i.e. 86% change) in *C. vulgaris* and *P. tricornutum*, respectively. Charge  
219 neutralization of the microalgal suspension has been observed as a major flocculation  
220 mechanism in previous studies using polymeric flocculants. For example, Zheng et al. (2012)  
221 observed an increase in zeta potential from -19 to 0.8 after addition of 20 mg/L  $\gamma$ -glutamic acid  
222 to a *C. vulgaris* suspension.

### 223 [FIGURE 2]

224 Results from Fig. 2 suggest that complete neutralization of microalgae cells suspension is  
225 not necessary for effective flocculation. The microalgal suspension reached an isoelectric point  
226 at polymer doses of 45.9 and 16.2 mg/g dry biomass for *C. vulgaris* and *P. tricornutum*,  
227 respectively. These doses were 2.4 and 1.2 times higher than the optimal dose for *C. vulgaris*  
228 and *P. tricornutum*, respectively. Over dosing of a polymer can cause charge reversal of the  
229 microalgae cells and thus decrease the flocculation efficiency. This process, called  
230 restabilization, has also been observed with other polymeric flocculants such as cationic starch  
231 (Vandamme et al., 2010) and chitosan (Rashid et al., 2013). The restabilization did not occur

232 in this study at a concentration range of 7 to 20 mg/L (i.e. even after the isoelectric point),  
233 suggesting that the cationic polyacrylamide can be used over a wide range of doses.

234 The microalgal culture medium can influence neutralization and thus, flocculation  
235 efficiency. The high ionic strength of seawater (i.e. marine microalgal culture) reduced the  
236 flocculation efficiency of cationic polymers (Roselet et al., 2015; Bilanovic et al., 1988; König  
237 et al., 2014). König et al. (2014) reported that medium salinity reduction (50%) could improve  
238 the flocculation of polymeric Zetag® and marine microalgal *Chlorella stigmatophora*.  
239 However, the polymer used in this study can perform effectively well with both freshwater and  
240 marine species, thus can be potentially applied to other microalgae.

### 241 3.3 Impact of pH on flocculation efficiency

242 The flocculation efficiency of cationic polyacrylamide with *C. vulgaris* and *P. tricornutum*  
243 was independent of pH ranging from 6 to 9 (Table 1). No statistically significant difference  
244 (unpaired *t*-test, *p*-value > 0.05) in optical density removal, NTU and zeta potential was  
245 observed at pH 6 to 9. The pH of microalgal culture can vary depending on different growth  
246 phase, an alkalinity level of growth medium and the concentration of dissolved carbon dioxide.  
247 The obtained results show that no pH adjustment would be required for the microalgal  
248 harvesting in this study within a pH range of 6-9.

249 [TABLE 1]

250 Changing pH values of the microalgae growth medium is often required for effective  
251 flocculation as shown in previous studies (Horiuchi et al., 2003; Ummalyma et al., 2016; Lal &  
252 Das, 2016; Tran et al., 2017). For example, the cell surface charge of *Chlorella* species  
253 decreases at low pH (Lal & Das, 2016). At a pH of 5.5, the amine groups in extracellular  
254 polysaccharides, protein and lipids dissociate, and the carboxylic group is used to protect the  
255 dissociation. This process decreases the negative charge of microalgal cells (Tran et al., 2017).  
256 Increasing the pH values can induce chemical precipitation of calcium and magnesium salts in  
257 the culture medium. Chemical precipitation can increase the impurity of the final microalgal

258 products. Adjusting pH during the harvesting process also change the cell integrity and the  
259 downstream process (Augustine et al., 2019; Şirin et al., 2012). A flocculant that can work  
260 effectively without pH adjustment is desirable. The finding of this study, therefore, suggests  
261 such flocculant exists and, to our knowledge, this is the first time it is reported.

### 262 3.4 Final biomass harvesting

263 The microalgal biomass recoveries were 100 and  $90 \pm 2.3\%$  for *C. vulgaris* and *P.*  
264 *tricornutum*, respectively (Fig. 3a). At lab-scale testing, the addition of polymer at 7 mg/L (i.e.  
265 at a cost of 0.049 \$AUD per m<sup>3</sup>) into both freshwater and marine microalgal cultures formed  
266 very stable flocs (i.e. without any alterations after 24 h). The solutions can be easily decanted  
267 to remove water (Fig 3b&c). Simultaneously, above 98% recovery of the culture medium was  
268 achieved in this study. The results from this study provide conclusive evidence that cationic  
269 polyacrylamide that has been widely used for sludge dewatering can be also used for microalgal  
270 biomass and culture medium separation.

271 [FIGURE 3]

272

## 273 4. Conclusions

274 This study demonstrates the effectiveness of a proprietary high charge high molecular weight  
275 cationic polyacrylamide for simple, robust, and efficient recovery of freshwater and marine  
276 microalgae *C. vulgaris* and *P. tricornutum*. A dose-response relationship showed that the  
277 optimal polymer doses were 18.9 and 13.7 mg/g dry biomass. At the optimal dose, microalgal  
278 cell surface charge was neutralised at 64 and 86% for *C. vulgaris* and *P. tricornutum*,  
279 respectively. Between pH 6 and 9, the solution pH did not affect flocculation efficiency. Thus,  
280 this method can be used for both freshwater and marine microalgae species over a wide range  
281 of culture conditions.

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## 286 **6. Declarations**

287 The authors declare that there is no conflict of interest regarding the publication of this  
288 article.

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