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1 **Biomethane production from anaerobic co-digestion at wastewater treatment plants: A**  
2 **critical review on development and innovations in biogas upgrading techniques**

3  
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25 **Abstract**

26 Anaerobic co-digestion (AcoD) can utilise spare digestion capacity at existing wastewater  
27 treatment plants (WWTP) to generate surplus biogas beyond the plant's internal energy  
28 requirement. Data from industry reports and the peer-reviewed literature show that through  
29 AcoD, numerous examples of WWTPs have become net energy producers, necessitating other  
30 high-value applications for surplus biogas. A globally emerging trend is to upgrade biogas to  
31 biomethane, which can then be used as town gas or transport fuel. Water, organic solvent and  
32 chemical scrubbing, pressure swing adsorption, membrane separation, and cryogenic  
33 technology are commercially available CO<sub>2</sub> removal technologies for biogas upgrade.  
34 Although water scrubbing is currently the most widely applied technology due to low capital  
35 and operation cost, significant market growth in membrane separation has been seen over the  
36 2015-2019 period. Further progress in materials engineering and sciences is expected and will  
37 further enhance the membrane separation competitiveness for biogas upgrading. Several  
38 emerging biotechnologies to i) improve biogas quality from AcoD; ii) accelerate the absorption  
39 rate, and iii) captures CO<sub>2</sub> in microalgal culture have also been examined and discussed in this  
40 review. Through a combination of AcoD and biogas upgrade, more WWTPs are expected to  
41 become net energy producers.

42

43 **Keywords:** Biogas upgrading; Anaerobic co-digestion; Biomethane; Biogas utilisation;  
44 Bioenergy.

45

## 46 **1. Introduction**

47 Securing affordable and clean energy from sustainable sources is a global challenge of our  
48 time. Addressing this challenge has resulted in a paradigm shift in many aspects of the  
49 economy, including organic waste management. The conventional view of waste as a  
50 disposable material is no longer suitable. In a circular economy, organic waste is a resource for  
51 energy and nutrient recovery. Indeed, carbon, nitrogen, phosphorus, and energy can be  
52 sustainably and economically extracted from organic wastes such as food wastes, sewage  
53 sludge. A globally emerging practice is to valorise urban organic waste via anaerobic co-  
54 digestion (AcoD) using the spare capacity at wastewater treatment plants (WWTPs) (Nghiem  
55 et al. 2017, Xie et al. 2018, Miryahyaei et al. 2020, Chan et al. 2019, Battle-Vilanova et al.  
56 2019).

57 Recent success in full-scale AcoD implementation demonstrates the potential role of  
58 WWTPs as energy producers. Anaerobic digestion facilities at WWTPs are used to treat  
59 sewage sludge with low organic content. Thus, their capacity is governed by hydraulic rather  
60 than organic loading. To utilise the spare digestion capacity (Schwarzenbeck et al. 2008),  
61 organic waste can be co-digested with sewage sludge to increase biogas production.

62 AcoD increases biogas production by 2.5 to 4 times compared to the digestion of only  
63 sewage sludge (Shen et al. 2015). Several WWTPs have become net energy producers (Nghiem  
64 et al. 2017, Shen et al. 2015, Macintosh et al. 2019). The Grevesmuhlen WWTP (Germany)  
65 converts a mixture of primary sludge, waste activated sludge, and grease to biogas, then  
66 through gas engines, produces 20% surplus energy (Schwarzenbeck et al. 2008). The  
67 Köhlbrandhöft plant (Germany's largest WWTP, serving 1.85 million residents in Hamburg)  
68 has also produced 15% more electricity than it consumes on an annual basis. Encouraging  
69 success in AcoD implementation at WWTPs has become an impetus for new applications of  
70 the surplus biogas.

71 Raw biogas contains about 65% CH<sub>4</sub>, 35% CO<sub>2</sub>, and a trace quantity of hydrogen sulfide,  
72 water vapor, ammonia, and siloxane depending on the types of feedstock and digestion process  
73 (Mattioli et al. 2017, Wickham et al. 2018, Jang et al. 2015, Martínez et al. 2012). The presence  
74 of CO<sub>2</sub> and other trace gases reduces the economic value and limits beneficial applications of  
75 biogas. Thus, biogas must be pretreated to remove hydrogen sulfide, water vapor, and other  
76 trace gases before the most beneficial applications. In addition to pretreatment, high-value  
77 applications such as transport fuel or natural gas grid injection require complete removal of  
78 CO<sub>2</sub> for biomethane production. The process of CO<sub>2</sub> removal to produce biomethane is called  
79 biogas upgrading. Biogas upgrading technologies such as water or organic physical scrubbing,  
80 chemical scrubbing, pressure swing adsorption, membrane separation, and cryogenic  
81 technology are available for commercial applications but can be very energy-intensive. The  
82 selection of both pretreatment and upgrading technologies depends on the biogas composition,  
83 the available resources, and the final product quality.

84 This paper reviews the state-of-the-art knowledge on the biomethane production processes  
85 that can combine with AcoD process at WWTPs to leverage existing infrastructure. This review  
86 focuses on the biogas production capacity of AcoD and the potential utilisation of biogas and  
87 the associated quality requirements. A major focus is given to the pre-treatment and upgrading  
88 technologies since they are essential for beneficial utilisation of the produced biogas.  
89 Additional benefits emerging from these techniques are also reviewed. This critical review  
90 expects to guide practitioners, water engineers, and scientists on future sustainable  
91 development endeavours.

## 92 **2. Anaerobic co-digestion at WWTPs**

### 93 **2.1 AcoD at WWTPs**

94 AcoD at WWTPs refers to the digestion of sewage sludge with one or more co-substrates  
95 with high organic content. These co-substrates are essential organic waste such as food and

96 kitchen waste, organic fraction of municipal solid waste (OFMSW), fat oil and grease (FOG),  
97 food/beverage processing waste, and biofuel by-products (i.e., crude glycerol, microalgae, corn  
98 silage). The theoretical principle of AcoD is the complementarity between nutrient-rich sewage  
99 sludge and carbon-rich organic wastes to boost the anaerobic digestion (AD) performance (Xie  
100 et al. 2018, Mattioli et al. 2017, Solé-Bundó et al. 2019, Salama et al. 2019, Siddique et al.  
101 2018, Wang et al. 2020, Aichinger et al. 2015).

102 AcoD of sewage sludge with organic wastes significantly increases the organic loading rate  
103 (OLR) with only a marginal increase in hydraulic loading to enhance biogas production  
104 (Wickham et al. 2018, Nghiem et al. 2014a). Nghiem et al. (2014a) demonstrated in a pilot-  
105 scale AcoD that intermittent injection of crude glycerol (i.e., byproducts from oil refinery  
106 industry) at 0.63 and 3% v/v in sewage sludge led to an increment of 50 and 80% in biogas  
107 production. Co-digestion of soft drink beverage waste at 10 and 20% of feed volume increased  
108 biogas production by 89 and 191%, respectively (Wickham et al. 2018). [Cavinato et al. \(2013\)](#)  
109 assessed both the pilot and full-scale AcoD of sewage sludge and OFMSW and achieved an  
110 enhancement in biogas production by nearly 40-50%. Kim et al. (2011a) reported that an 80%  
111 increase in the biogas production was attained at WWTP in Velenje, Slovenia. An uplift in  
112 specific biogas production was observed at 230%, resulting in a 130% increase in electricity  
113 production and 55% in heat energy (Zupančič et al. 2008). Koch et al. (2015) reported that 78%  
114 of the energy requirement of WWTPs could be gained from AcoD of sewage sludge with food  
115 wastes at ratio of 90:10% feed volume. AcoD at WWTPs have provided 100% required energy  
116 in a number of examples (Table 1). Based on the biogas production, AcoD enhances the  
117 utilisation of digester volume by 2.5 to 4 times. Shen et al. (2015) reported that AcoD plants  
118 produce the biogas at the rate of 2.5 to 4 m<sup>3</sup> per m<sup>3</sup> digester volume compared to 0.9 to 1.1 m<sup>3</sup>  
119 in anaerobic digester (AD) plants. Likewise, Wickham et al. (2018) suggested that AD of  
120 sewage sludge can receive an additional 2 kg chemical oxygen demand (COD)/m<sup>3</sup>.d from

121 beverage waste to achieve an OLR of 3.8 kg COD/m<sup>3</sup>.d with proportional increase in biogas  
122 production.

123 Financial benefits from AcoD can be realised through energy production and gate fee.  
124 Electricity and heat generated from biogas can be used for onsite consumption. Excess energy  
125 can also be sold. [About \\$2 million per year in electricity revenue was achieved through AcoD  
126 of fat-oil-and-grease, food waste, and sewage sludge at the East Bay Municipal Utility District,  
127 USA \(Shen et al. 2015\).](#) Another example is to utilise the produced biogas in an adjacent facility  
128 to WWTPs, minimising gas transportation and investment cost. The Des Moines Metropolitan  
129 Wastewater Reclamation Authority WWTPs (Iowa, USA) sells 40-50% of the produced biogas  
130 to a nearby oilseed processing facility, providing an income of 0.8 million USD per year (Zhu  
131 et al. 2015).

132 Gate fee (i.e. or tipping fee – a charge upon a given quantity of waste at waste processing  
133 facility) can also generate revenue to support AcoD. In the US, the food waste tipping fee varies  
134 from 50 to 170 USD/ton (Shen et al. 2015). In Australia, gate fee typically consists of landfill  
135 levy (which is then reinvested to activities that divert waste away from landfill) and operation  
136 cost. The current high landfill levy and potential increase in near future (i.e., significant landfill  
137 shortages) will create greater incentives for co-digestion of residual municipal solid waste. As  
138 an example, the estimated gate fee in New South Wales is \$110 USD/ton ([Source from  
139 Australian Paper's Energy from Waste feasibility study – Fact Sheet 6](#)). Although numerous  
140 WWTPs have adapted AcoD in their operation (Table 1), economic data are commercially  
141 sensitive and thus rarely available in the literature. Several technical aspects, considerations as  
142 well as possible solutions raised from implementation of AcoD at WWTPs have been available  
143 in the literature (Nghiem et al. 2017, Xie et al. 2018, Solé-Bundó et al. 2019, Salama et al.  
144 2019, Siddique & Wahid 2018), in the favour of supporting AcoD.

145  
146

147

148 **Table 1:** Examples of AcoD implementations at WWTP to achieve 100% energy self-  
 149 sufficiency and become net energy producers (Nghiem et al. 2017, Shen et al. 2015)

Location	Feedstock (V/V ratio)	Capacity (m <sup>3</sup> )	Biogas production (GWh/y)
Point Loma WWTP – USA	Mixed PS + WAS	8 x 13,600	193
Gloversville–Johnstown Joint WWTP – USA	Sludge + (yogurt/cheese whey wastewater	1 <sup>st</sup> : 5700 2 <sup>nd</sup> : 4900	28
Sheboygan Regional WWTP – USA	Sludge + FOG + dairy waste	N/A	32
Gresham WWTP – USA	Sludge (87%) + FOG (13%)	2 x 3800	17.2
East Bay Municipal Utility District WWTP - USA	Sludge + FOG/Food waste/HSW	12 x 7500	90
Strass im Zillertal WWTP – Austria	Mixed BNR WAS + Grease trap + Crude glycerol + Food waste	N/A	10
Grevesmuhlen WWTP – Germany	PS (10%) +WAS (60%) + Grease skimming sludge (30%)	2 x 1000	1.95
Zürich Werdhölzli WWTP – Switzerland	Sludge (79%) +FOG (21%) = 23,000 t TS/y	4 x 7250	41.4

150 WAS = waste activated sludge; PS = primary sludge; FOG = fat oil and grease; BNR = biological nutrient  
 151 removal; TS = total solid; HSW = high strength waste

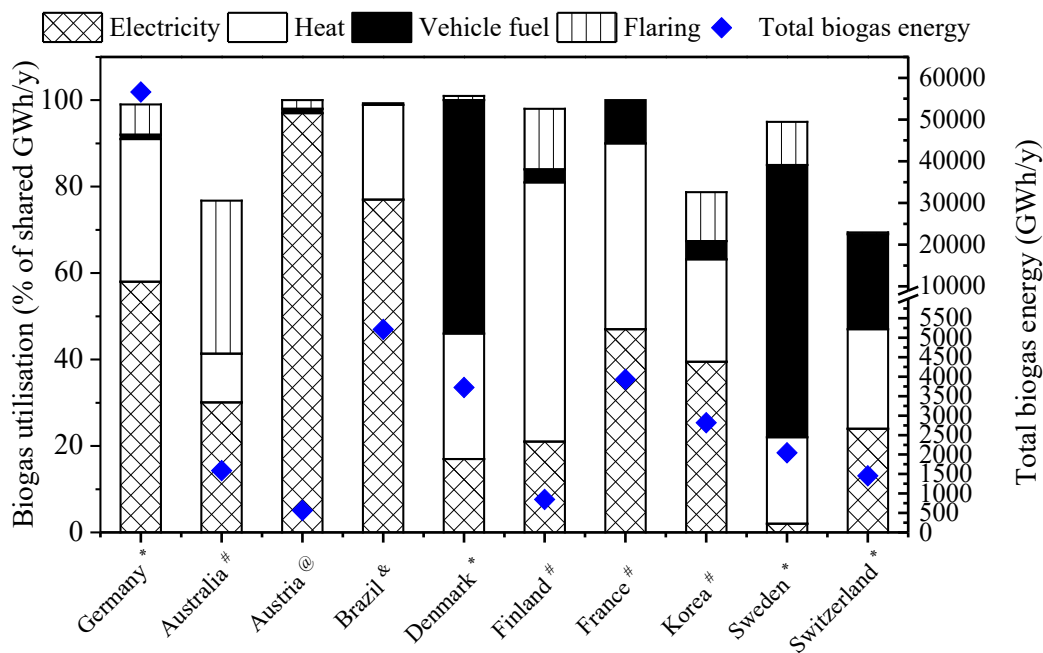
152

153 2.2. Utilisation of biogas

154 Most of the produced biogas is currently utilised for heat and electricity generation (Fig. 1).  
 155 Biogas upgrade to biomethane has only been implemented in a few countries for transport fuel  
 156 and natural gas grid injection. A notable example is Sweden, where more than half of the  
 157 produced biogas is used as a transport fuel, supporting 44,000 light vehicles, 750 buses, and  
 158 2,200 trucks (data in 2017 from CNG Europe). Germany is currently the world largest biogas  
 159 producer. Thus, although a small portion of biogas is purified and used as transport fuel, it is  
 160 enough to power about 96,000 light vehicles, 1,700 buses, and 200 trucks (data in 2017 from  
 161 CNG Europe). An emerging biomethane market has also been seen in several countries such  
 162 as Denmark, France, Switzerland, and South Korea (Fig. 1).



163 Biogas utilisation options are supported initially by government incentives such as feed-in  
 164 tariffs and tax exemptions, and energy policy. For example, the feed-in tariffs for electricity  
 165 resulted in biogas being used to produce electricity in Germany, UK, and Austria. Unlike  
 166 Sweden, the tax exemption favours the transport fuel application. France, Denmark, Sweden,  
 167 and the UK have strong financial support for biogas injection into gas grids.



168 **Figure 1:** Biogas utilisation (i.e. % of shared GWh/y) in some countries. The symbol \*, #,  
 169 &, @ indicated data source from 2018, 2017, 2016, and 2013, respectively (Source: IEA  
 170 Bioenergy Task 37).  
 171

172 Wastewater treatment is an energy intensive process. The process accounts for about 3% of  
 173 consumed electrical energy annually in USA (Wan et al. 2016). It is estimated that the energy  
 174 demand for wastewater treatment is between 20 and 30 KWh per person annually. The  
 175 wastewater treatment process also contributes to 5% of global greenhouse gas emission  
 176 (Nghiem et al. 2017, Gude 2015). In this regard, AD of sewage sludge can produce biogas to  
 177 compensate 15 to 18 KWh per person. Biogas conversion to heat and energy also reduce the  
 178 greenhouse gas emission volume at WWTP. Current approach is to intensify the capacity of  
 179 AD facility via AcoD at WWTP to produce more biogas. Indeed, WWTPs produce a significant

180 volume of biogas (Table 2). For example, WWTPs in Germany contributes above 50% of total  
 181 biogas production in 2019. It is expected that the amount of biogas production will exceed the  
 182 heat and energy requirement onsite, necessitating other applications for this renewable energy.

183 **Table 2:** Number of WWTPs with AD and AcoD of sewage sludge with organic waste for  
 184 biogas production and the relative biogas production (Source: IEA Bioenergy Task 37:  
 185 <https://task37.ieabioenergy.com/country-reports.html>).

Country	WWTPs with AD	WWTPs with AcoD	Biogas production at WWTPs (GWh/y)	Biogas from WWTPs (% of total production)	Year of data collection
Australia	52	2	381	24	2017
Brazil	10	3	210	4	2016
Denmark	51	n.a	308	8.3	2018
Finland	16	n.a	162	23	2017
France	88	n.a	442	26	2017
Germany	1274	n.a	3657	56.2	2019
Norway	27	n.a	240	33	2015
South Korea	78	21	630	22	2017
Sweden	138	n.a	715	35	2018
Switzerland	473	n.a	633	43.5	2018
The Netherlands	80	n.a	640	37.2	2018
United Kingdom	163	n.a	1280	15.4	2018
Canada	31	n.a	n.a	20.7	2019
USA	1241	216	n.a	n.a	2015

186  
 187 Providing electricity to the power grid or injecting biomethane to the natural gas grid for  
 188 distribution and transport fuel are potential applications of the surplus biogas at WWTPs.  
 189 Feeding electricity to the power grid is not always feasible. There have been some government  
 190 incentives especially Europe that allows WWTPs to feed surplus electricity to the power grid  
 191 at a favourable tariff. However, many of these incentives have expired or about to expire. In  
 192 some countries, WWTP utilities may not have a power generator license to inject electricity  
 193 into the power grid. There are also major technical challenges to the existing energy  
 194 infrastructure, which was originally designed only for energy distribution rather than a flexible  
 195 feed in and sharing network. Synchronisation of multiple power sources to the distribution grid

216 needs to match the voltage, frequency, and phases. In addition, the imbalance between supply  
217 and demand as well as inappropriate load management can destabilise the distribution grid.

218 Instead of electricity production, upgrading to biomethane for domestic consumption and  
219 transport fuel appears to be an appealing alternative for the surplus biogas from WWTPs. The  
220 methane economy is mature and many countries have extensive natural gas grid distribution.  
221 The surplus biomethane production at WWTPs can be fed into the gas grid for distribution and  
222 storage. This also takes the advantages that storage capacity and duration of storage for  
223 methane is significantly higher than other energy storage systems (e.g. electricity in battery).  
224 Facilities such as natural gas refuel stations and increase in number of natural gas vehicles in  
225 line with the production of biomethane. An example of a complete production line includes  
226 biogas production, biogas upgrading and a refuelling station has been operated at a Swedish  
227 WWTP. Thus, biogas upgrading to biomethane for natural gas grid injection or transport fuel  
228 is probably a preferred option for the surplus gas at WWTPs.

229

### 230 **3. Methane and other gases in biogas**

231 Raw biogas typically contains about 65% CH<sub>4</sub> and 35% CO<sub>2</sub> (Mattioli et al. 2017, Wickham  
232 et al. 2018). The energy content of biogas is defined by the methane concentration – the higher  
233 the methane, the higher the calorific energy value of the gas. For example, the calorific value  
234 (i.e., Wobbe index) of biogas with 70% of CH<sub>4</sub> content is 21.5 MJ/Nm<sup>3</sup>, whereas that of  
235 biomethane (100% CH<sub>4</sub>) is 35.8 MJ/Nm<sup>3</sup>. The high volume of CO<sub>2</sub> in biogas does not only  
236 reduce the calorific but also make it uneconomic for compression and transportation for off-  
237 site utilisation.

238 Trace gases especially hydrogen sulfide (H<sub>2</sub>S), water vapor, and siloxane can be detrimental  
239 to downstream biogas utilisation processes. H<sub>2</sub>S is corrosive to co-generators, biogas storage  
240 facilities, compressors, and pipelines. The combustion of H<sub>2</sub>S produces sulfur dioxide, which  
241 is a major air pollutant (Zhu et al. 2015).

222 H<sub>2</sub>S level in biogas from AcoD is an important parameter affecting the usage of biogas.  
223 Under anaerobic conditions, sulphur-bearing organic compounds and sulphate in organic  
224 wastes are reduced to sulphide, which is released to biogas in the form of H<sub>2</sub>S (Cirne et al.  
225 2008) . High sulphur content substrate produces high H<sub>2</sub>S content in biogas (Hansen et al.  
226 2004) . The H<sub>2</sub>S content in biogas from a WWTP digesting only sewage sludge is in the range  
227 from 500 to 2,500 ppmv. Animal wastes, slaughterhouse wastes, dairy milk industry wastes  
228 can produce biogas with 20,000 to 30,000 ppmv H<sub>2</sub>S (Hansen et al. 2004). Up to 31,000 ppmv  
229 H<sub>2</sub>S in biogas from AcoD with seaweed has been reported (Hansen et al. 2004). By contrast,  
230 most internal combustion engines manufactures limit H<sub>2</sub>S to 100 ppmv in biogas. According  
231 to the European biomethane standards, the concentration of H<sub>2</sub>S is required to be less than 1  
232 ppmv for gas grid injection and transport fuel (European biomethane standards).

233 Water vapor in biogas can also lead to corrosion problems (Ryckebosch et al. 2011) . Water  
234 content in biogas depends on the digester's operating temperature (e.g., mesophilic or  
235 thermophilic). The water content is about 4 to 5% (v/v) of raw biogas from the mesophilic  
236 digester. At higher temperatures (i.e., 55 to 60°C in thermophilic digester), 7 - 8% (v/v)  
237 (between 30 and 100 g water per m<sup>3</sup>) of the water content has been recorded. Water vapor  
238 removal is necessary to avoid corrosion in biogas upgrading and utilisation. The permissible  
239 water content is below 10 mg/Nm<sup>3</sup> for gas grid injection.

240 The presence of siloxanes in biogas can lead to the formation of siloxane dioxide particles.  
241 Siloxane dioxide particles are abbrasive and are adhesive to metal surface, causing excessive  
242 wear and tear of the the co-generator engines. Siloxanes concentration in biogas is between 1  
243 to 400 mg(Si)/Nm<sup>3</sup>, while the maximum permissibl siloxane concentration in natural gas is 5  
244 mg (Si)/Nm<sup>3</sup> (Ryckebosch et al. 2011).

245 Biomethane for gas grid injection and transport fuel must meet very stringgent standards  
246 and are due to the relevance of the technical execution of installations, planning, construction,

247 and operation. The European Commission has begun to develop the European biomethane  
248 standards for grid injection and transport fuel. The new standard is set to bring legal and  
249 technical security, market assessment, and precondition free trade amongst countries.  
250 Requirements such as total H<sub>2</sub>S < 1 ppmv, and siloxanes < 0.5 mg(Si)/Nm<sup>3</sup> are examples in the  
251 new standards.

#### 252 **4. Biomethane market**

253 Biomethane market has gained significant momentum in recent years. The number of  
254 new biogas upgrading plants increases worldwide (Table 3). *It results from combined factors  
255 including i) advanced in biogas upgrading technology; ii) paradigm shift from a low economic  
256 electricity and heat production to new opportunities for use biomethane in the transport sector,  
257 and iii) moving towards a green economy model (Zhu et al. 2019).* As an upgraded product of  
258 biogas, biomethane is essentially identical to natural gas. Thus, biomethane can be injected into  
259 the natural gas grid or used as transport fuel. Germany and Sweden have the largest markets  
260 for biomethane in the world. A growing interest can also be seen in other countries, especially  
261 the UK, France, and Switzerland (Table 3).

262 The global biomethane market was valued at USD 0.62 billion in 2017. With the annual  
263 growth rate of 26%, a \$4.96 billion market size is estimated by 2026. Several countries have  
264 set ambitious target for biomethane as natural gas replacement for household consumption  
265 (Hoo et al. 2020). France plans to provide 8 TWh of biomethane energy by 2023 (Herbes et al.  
266 2018). In the UK, biomethane is expected to be a major source of the future clean gas supply  
267 (Richards et al. 2020).

268 Biomethane can also be liquefied or compressed biomethane for storage or used as  
269 transport fuels. Liquefied biomethane also has higher energy content, suitable for heavy  
270 vehicles and providing long distance transportation. Biomethane utilisation as transport fuel  
271 has continued to increase over the years. Sweden has set a target of 100% of transport fuel from

272 biomethane by 2030. This target appears realistic. In 2016, 82.8% of transport fuel was from  
 273 biomethane. This number increased to 90.8% in 2018.

274

275 **Table 3:** Increase in the number of biogas upgrading plants in selected countries over the  
 276 2014 – 2019 period (Source: IEA Bioenergy Task 37:  
 277 <https://task37.ieabioenergy.com/country-reports.html>).

Country	Number of plants in 2014	Number of plants 2016	Number of plants in 2019
France	8	30	47
Denmark	12	32	34
United Kingdom	37	85	96
Italy	5	7	8
Finland	9	12	17
Switzerland	24	31	45
Netherlands	21	26	53
Germany	178	194	203
Austria	14	15	13
Sweden	59	63	69
Hungary	2	2	n/a
Luxembourg	3	3	3
Spain	1	1	
Norway	n/a	n/a	9
Australia	0	0	0
South Korea	n/a	n/a	10
Japan	n/a	6	6
China	n/a	n/a	2
USA	n/a	n/a	50

278 n/a = not available

## 279 5. Biogas pretreatment

### 280 5.1 H<sub>2</sub>S removal

281 Desulphurisation of raw biogas is considered the essential step before further processing  
 282 and the use of biomethane. Methods to remove H<sub>2</sub>S from biogas can be categorised into three  
 283 groups: i) biological desulphurisation; ii) absorption to a liquid solution (water or chemical  
 284 scrubbing), and iii) adsorption on a solid absorbent (iron sponge, iron oxide pellets, activated  
 285 carbon).

286 Biological desulphurisation can be performed *in-situ* to the anaerobic digester. Air or  
287 oxygen is injected into the digester to provide oxygen molecules (Ryckebosch et al. 2011,  
288 Nghiem et al. 2014b). Nghiem et al. (2014b) injected oxygen to regulate the oxidation redox  
289 potential between -320 to -270 mV to reduced H<sub>2</sub>S in biogas from 6000 to below 30 ppm  
290 without any observable effect on digester performance. Biological desulphurisation can be  
291 carried out ex-situ in biofilters, which are packed bed scrubbers containing immobilized  
292 microorganisms. H<sub>2</sub>S is captured in the liquid film and biologically translated to sulphur and  
293 sulphate. The liquid of the scrubber bed can be regenerated if the pH level decreases below 7.  
294 This system is unable to supply a stable H<sub>2</sub>S content <100 ppmv, and this value varies with the  
295 qualities of the raw biogas. Hence, this process can not be confidently employed for  
296 biomethane production (Pettersson et al. 2009).

297 H<sub>2</sub>S in biogas can be removed by absorption in the scrubbing technologies (i.e., physical  
298 and chemical scrubbing) (Table 4). Water and alkaline solution (e.g., sodium hydroxide,  
299 calcium hydroxide, and potassium hydroxide) and amines can absorb H<sub>2</sub>S. In this regard, H<sub>2</sub>S  
300 can be simultaneously removed during biogas upgrading (i.e. CO<sub>2</sub> removal). However, it is  
301 worth mentioning that chemical reaction between H<sub>2</sub>S with absorbent is an irreversible process,  
302 limiting the absorbent regeneration.

303 Adsorption of H<sub>2</sub>S in iron oxide pellets/sponge, and activated carbon column is an effective  
304 method for biogas treatment. Iron oxide reacts with H<sub>2</sub>S in biogas to form ferric sulfide (Wang  
305 et al. 2011). Wang et al. (2011) reported that iron oxide could uptake large amount of H<sub>2</sub>S (e.g.,  
306 0.25 kg H<sub>2</sub>S per kg iron oxide). The formed ferric sulfide can be changed to ferric oxide and  
307 elemental sulfure by exposing it to air or oxygen during the regeneration process. Adsorption  
308 is the most applied method for H<sub>2</sub>S removal because of its outstanding performance,  
309 regeneration capacity, and easy to use.

310

311 **Table 4.** Considerations for selection of desulphurisation techniques

Method	Considerations	References
Air/oxygen injection	<ul style="list-style-type: none"> <li>- Potential over oxygenation affect anaerobic conditions</li> <li>- High cost of pure oxygen bottles</li> <li>- Limited full scale experience</li> <li>- High volume of N<sub>2</sub> in biogas</li> </ul>	(Nghiem et al. 2014b, Jeníček et al. 2017)
Bio-trickling filter	<ul style="list-style-type: none"> <li>- Unstable performance</li> <li>- Slow reaction rate</li> <li>- Difficult to set up the filter</li> </ul>	(Montebello et al. 2013)
Water absorption (scrubbing)	<ul style="list-style-type: none"> <li>- High water volume</li> <li>- Unstable performance</li> <li>- Can remove part of CO<sub>2</sub></li> </ul>	(Angelidaki et al. 2018)
Chemical absorption (scrubbing)	<ul style="list-style-type: none"> <li>- Ongoing chemical cost (no regeneration)</li> <li>- Performance is predictable</li> <li>- Partial CO<sub>2</sub> removal</li> </ul>	(Ryckebosch et al. 2011)
	<ul style="list-style-type: none"> <li>- Addition of catalyst solution (Fe(III)-EDTA to reduce chemical consumption</li> </ul>	(Horikawa et al. 2004)
Iron sponge (Iron oxide/hydroxide) adsorption	<ul style="list-style-type: none"> <li>- High operating costs</li> <li>- Excessive heat generation</li> <li>- H<sub>2</sub>SO<sub>4</sub> formation</li> </ul>	(Angelidaki et al. 2018)
Activated carbon adsorption	<ul style="list-style-type: none"> <li>- H<sub>2</sub>SO<sub>4</sub> formation</li> <li>- Impregnated with NaOH, KOH, Na<sub>2</sub>CO<sub>3</sub></li> <li>- Regeneration requirements more frequent</li> <li>- Modification of AC with CuSO<sub>4</sub>, KOH, ZnAc<sub>2</sub></li> </ul>	(Zulkefli et al. 2019, Ciahotný et al. 2019)

312

313 5.2 Water vapour removal

314 There are several methods to remove water vapour. Biogas can be cooled down to allow  
315 water vapor to condense (Ryckebosch et al. 2011). The condensed water is returned to the  
316 digester or drained out. Moisture in biogas can also be removed by adsorption dryers. These  
317 are high water adsorbent materials (e.g. silica gel, aluminum oxides, and molecular sieves).  
318 The used adsorbent can be regenerated by heating. Moisture can also be removed by increasing  
319 the biogas pressure. In this process, water vapor is not completely removed but the relative  
320 humidity of biogas is reduced. It is also noted that water vapour removal is performed after  
321 CO<sub>2</sub> removal in scrubbing technologies.



### 322 5.3 Ammonia removal

323 Ammonia can be present in biogas at a trace level of up to 100 ppm (Ryckebosch et al.  
324 2011). Ammonia can be removed simultaneously with water because of its high solubility in  
325 water. Scrubbing technologies (i.e. physical and chemical scrubbing) are effective in  
326 achieving complete removal of ammonia from raw biogas (Allegue et al. 2012). Therefore, the  
327 pretreatment to removal of water vapor and ammonia is not required if these methods are  
328 applied during CO<sub>2</sub> removal (i.e. biogas upgrading).

### 329 5.4 Siloxanes removal

330 Absorption and adsorption are the two common methods for the removal of siloxanes from  
331 biogas. Organic solvents, strong acids or bases solution can provide up to 97% siloxanes  
332 removal efficiency (Ryckebosch et al. 2011). However, the use of organic solvents, acids or  
333 bases could cause corrosion and produce hazardous chemicals that requires additional  
334 treatment. Adsorption on solid materials such as silicagel and activated carbon is preferred  
335 option. Activated carbon adsorption reduces siloxanes level to 0.1 mg/Nm<sup>3</sup> (Ajhar et al. 2010).  
336 Adsorbent regeneration is possible at high temperature (i.e., above 250 °C). Siloxanes removal  
337 is usually performed after water vapor since high moisture gas can decrease the removal  
338 efficiency (Schweigkofler et al. 2001).

## 339 **6. Biogas upgrading technologies**

340 Once pre-treated, biogas can be further processed to remove CO<sub>2</sub> to produce biomethane.  
341 Several biogas upgrading methods are already available at commercial scale. They include  
342 scrubbing (i.e. water, organic solvent, and chemical scrubbing), pressure swing adsorption,  
343 membrane separation, and cryogenic technology. The maturity of these methods varies widely.  
344 It is noteworthy that some biogas upgrade methods can result in the removal of impurities,  
345 especially H<sub>2</sub>S. For example, water scrubbing (at high pH) and pressure swing adsorption can  
346 remove both H<sub>2</sub>S and CO<sub>2</sub> together. On the other hand, a pretreatment step to remove H<sub>2</sub>S is

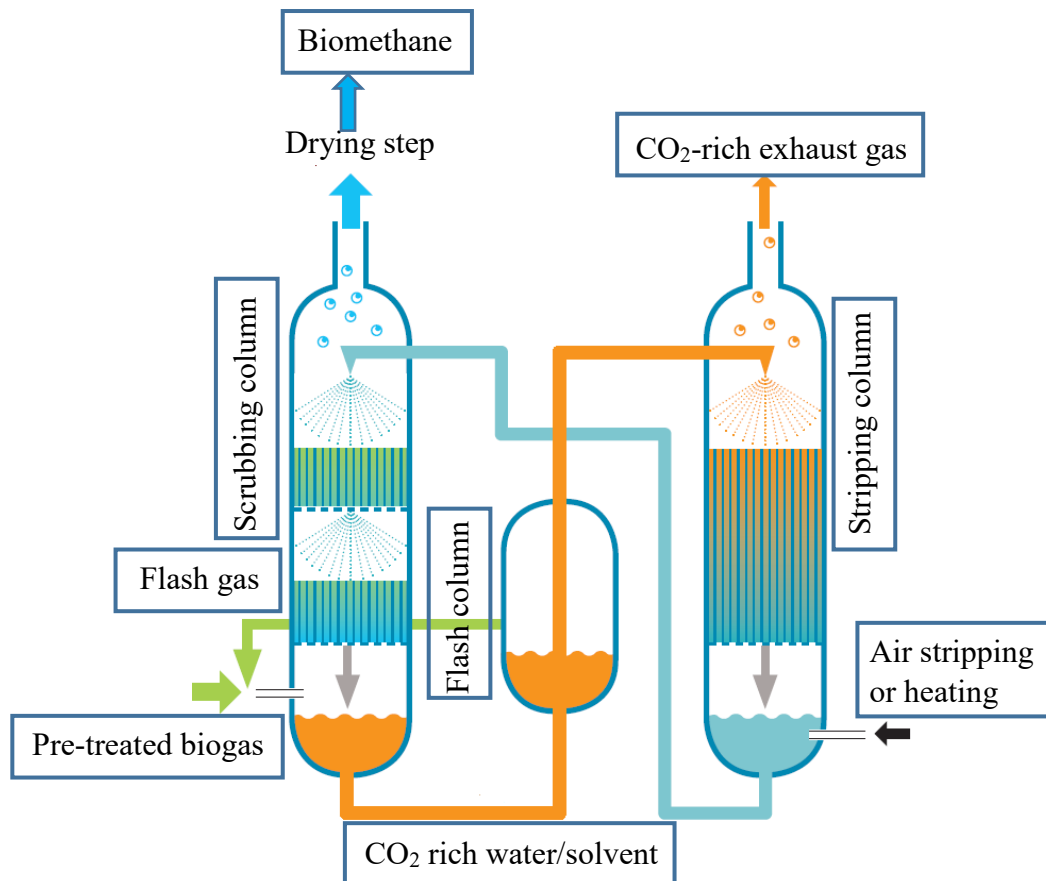
347 required for CO<sub>2</sub> removal by chemical scrubbing using amines. H<sub>2</sub>S removal is also required  
348 to avoid membrane poisoning in membrane separation techniques.

349 The high CH<sub>4</sub> purity is required for natural gas grid injection and vehicle fuel, meeting a  
350 few criteria such as high-energy content, gas transportation, storage, and technical restrictions.  
351 For example, biomethane is compressed in pressurised gas cylinders at 200 to 250 bar for  
352 storing and transporting purposes. While CH<sub>4</sub> can be readily compressed, a mixture of CO<sub>2</sub> and  
353 CH<sub>4</sub> has very different thermodynamic properties and cannot be readily compressed at high  
354 pressure for storage.

## 355 6.1 Scrubbing Technologies

### 356 6.1.1 Water or organic physical scrubbing

357 Water or solvent scrubbing relies on the difference in the solubility of gasses (CO<sub>2</sub> and CH<sub>4</sub>)  
358 in a wash solution ([Andriani et al. 2014](#)). The wash solution can be water (water scrubbing) or  
359 organic solvent (e.g. polyethylene glycol dimethyl ether, trade name as Genosorb or Seloxol).  
360 This method involves no chemical reaction. Since the gas solubility improves with increasing  
361 pressure, pre-treated biogas is pressurised and injected into the scrubbing column (Fig. 2).



362

363 **Figure 2:** Schematic of scrubbing technologies for the separation of CO<sub>2</sub> and CH<sub>4</sub>

364

365 In the water scrubbing process, the pretreated biogas is maintained 6-10 bar and 40 °C. At  
 366 this condition, the solubility of CO<sub>2</sub> is approximately 26 times higher than that of CH<sub>4</sub>. The gas  
 367 is injected via the bottom side of the column, while water is provided from the top. The counter-  
 368 current injection increases the gas and water interaction in the scrubbing column. This  
 369 configuration also allows CH<sub>4</sub> venting out from the top while CO<sub>2</sub> rich water is circulated into  
 370 a flash column from the bottom. At the flash column, the gas pressure decreases to 2.5 – 3.5  
 371 bars, releasing CH<sub>4</sub> gas to be recovered. The CO<sub>2</sub> rich water is pumped into a stripping column.  
 372 In this column, the air is injected at atmospheric pressure, resulting in the removal of CO<sub>2</sub> from  
 373 water (i.e., regeneration). The ventilated CH<sub>4</sub> is subjected to a drying step to produce final  
 374 biomethane (Angelidaki et al. 2018).

375 Although water scrubbing is a simple process with low energy consumption, high water  
376 consumption, and methane loss are major drawbacks. A total of 3-5% of methane can be lost,  
377 and the combustion of the exhaust gas is required to maintain emission regulation (Ryckebosch  
378 et al. 2011) . The water scrubbing method requires a large amount of water (200 m<sup>3</sup>/h for a gas  
379 flow of 1000 Nm<sup>3</sup>/h) (Sun et al. 2015). Thus, water regeneration is crucial for the economic  
380 viability of this technology. Water scrubbing can be advantageous when apply at WWTPs.  
381 Secondary and tertiary effluent can be used as water source without regeneration (Angelidaki  
382 et al. 2018).

383 An organic solvent can also be used as the wash solution. The process configuration is  
384 similar to water scrubbing (Fig. 2). CO<sub>2</sub> has a higher solubility in some solvent such as  
385 polyethylene glycol dimethyl ether than in water. Consequently, a smaller volume of solvent  
386 is required and the size of the scrubbing column can be reduced. The absorption process also  
387 occurs at lower pressure (4 to 8 bars) resulting in a lower energy demand compared to water  
388 scrubbing (6 to 10 bars. However, organic solvent regeneration is a complex process compared  
389 to water regeneration. Air stripping and pressure release are not effective to regenerate the  
390 organic solvents. In practice, organic solvent is heated to between 40 and 80 °C, requiring  
391 additional energy of 0.1 to 0.15 kWh/Nm<sup>3</sup> of biogas for regeneration (Ryckebosch et al. 2011,  
392 Gupta 2003, Singhal et al. 2017).

393

394

395

396

397

398

399 **Table 5:** Advantages and disadvantages of physical scrubbing technologies (Singhal et al.  
 400 2017, Kadam et al. 2017, Niesner et al. 2013)

Method	Advantages	Disadvantages
Water scrubbing	<ul style="list-style-type: none"> <li>- Many years of experience</li> <li>- Numerous plants are under operation</li> <li>- Less costly (i.e. water is a low-cost solvent).</li> <li>- Environmentally friendly solvent</li> <li>- Technically simple method</li> <li>- No additional heat</li> </ul>	<ul style="list-style-type: none"> <li>- Energy consumption: 0.2 to 0.5 kWh/Nm<sup>3</sup> of biogas.</li> <li>- High pressure 4-10 bars</li> <li>- Methane is up to 5 % by volume</li> <li>- Water is less selective (i.e. absorbent rate and loading is low)</li> </ul>
Solvent scrubbing	<ul style="list-style-type: none"> <li>- High absorption rate and loading per volume of solvent</li> <li>- Smaller footprint</li> </ul>	<ul style="list-style-type: none"> <li>- Energy consumption: 0.1 to 0.33 kWh/Nm<sup>3</sup> of biogas.</li> <li>- Additional heat to achieve effective regeneration</li> <li>- Potential environment pollution of used solvent</li> <li>- Methane loss is up to 4 % by volume</li> </ul>

401

402 6.1.2 Chemical scrubbing

403 Chemical scrubbing or chemical absorption is based on a reversible reaction between CO<sub>2</sub>  
 404 with a chemical adsorbent. Common chemical absorbents are monoethanolamine (MEA),  
 405 diethanolamine (DEA), methyldiethanolamine (MDEA), and other amine compounds. Their  
 406 solutions have high selectivity against CO<sub>2</sub>. Since chemical adsorbents are only reactive with  
 407 CO<sub>2</sub>, the CH<sub>4</sub> loss is minimal after its dissolution in solvent solution (0.1 to 0.2%) (Sun et al.  
 408 2015). Thus, a post-combustion of lean gas is not required. Chemical scrubbing can produce  
 409 high CH<sub>4</sub> purity (99% by volume). H<sub>2</sub>S removal upstream must be conducted because of the  
 410 corrosive reaction of H<sub>2</sub>S with amine solution, i.e., degradation of amine (Vega et al. 2014).

411 Regeneration of an amine solution is an energy-intensive process compared to the physical  
 412 scrubbing due to the strong binding between the gas molecules. The CO<sub>2</sub> saturated amine  
 413 solution is heated to above 110 °C for regeneration. The regenerated amine solution then cooled  
 414 down to 40 °C before starting a new absorption cycle. Regeneration consumes 0.4 to 0.8  
 415 kWh/Nm<sup>3</sup> of biogas, or about 15 to 30% of the energy generated from the biomethane (Leung

416 et al. 2014). Recent research has focused on minimising thermal energy requirement for  
417 regenerating amine solution. It is achieved by developing a new amine solution, optimising  
418 heat-exchanging equipment, and modifying operation conditions (temperature and gas flow  
419 rate) (Aroonwilas et al. 2009, Kim et al. 2011b). Amine degradation, equipment corrosion, and  
420 potential generation of the volatile compound into the atmosphere are other limitations of the  
421 chemical scrubbing method. Moreover, amine can degrade into nitrosamines and nitramines,  
422 potentially harmful to human health and the environment (Stowe et al. 2017).

## 423 6.2 Pressure swing adsorption

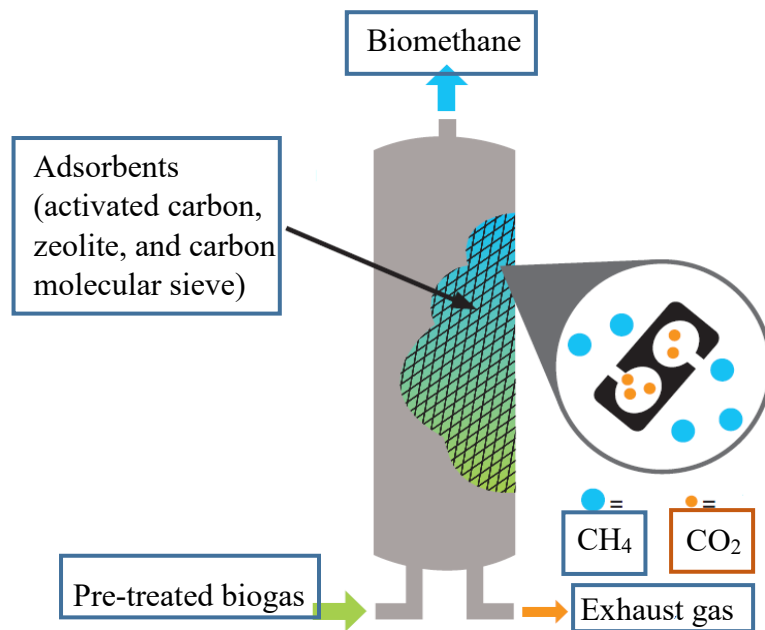
424 Pressure swing adsorption relies on the principle that  $\text{CH}_4$  and  $\text{CO}_2$  adsorb differently to  
425 specific surfaces or pores of adsorbents. Since the adsorption of  $\text{CO}_2$  is proportionally to high  
426 pressures and low temperatures, the pressure swing adsorption process utilise  
427 pressure/temperature differences, i.e., pressure-temperature swing, to carry out the separation  
428 (Ntiamoah et al. 2016). The system main part is a column, filling with adsorbents such as  
429 activated carbon, zeolites, calcium oxides, hydrotalcites, and carbon molecular sieves (Fig. 3).  
430 These materials are porous and of high surface areas to enhance adsorption capacity (Leung et  
431 al. 2014). Desulphurisation is required before adsorption since  $\text{H}_2\text{S}$  is irreversibly adsorbed by  
432 the adsorption substance and produces toxic effects (Patterson et al. 2011).

433 In the pressure swing adsorption, pre-treated biogas is compressed to 2-7 bar and cooled to  
434 about  $70\text{ }^\circ\text{C}$  to improve the adsorption. The pressured gas is injected into the adsorption column  
435 from the bottom.  $\text{CO}_2$  molecules, which are smaller than methane molecules, accumulate to a  
436 much greater degree on the surfaces or in the pores than  $\text{CH}_4$ . At the same time,  $\text{CH}_4$  remains  
437 primarily in the gas phase and escapes from the column head, resulting in a methane-rich  
438 product gas. Once the methane is released, the pressure inside the column decreases to  
439 atmospheric pressure. The adsorbed  $\text{CO}_2$  dissolves from the surfaces and returns into the gas  
440 phase. This gas is blown off ( $\text{CO}_2$ - rich exhaust gas) via a valve at the column bottom (Fig. 3).

441 The column is then filled with biogas to begin a new cycle. Pressure swing adsorption has been  
 442 in operation for many years at many reference plants for biogas upgrading. The biomethane  
 443 quality is nearing 96-98%, with 1.5 to 2.5% methane loss (Allegue et al. 2012). Therefore,  
 444 post-combustion of exhaust gas is required to minimise methane release in the atmosphere (Sun  
 445 et al. 2015). Overall, the energy requirement of the pressure swing adsorption is between 0.15  
 446 to 0.35 kWh/Nm<sup>3</sup> of biogas, making it a competitive method for biogas upgrading.

447 In the temperature swing adsorption, pre-treated biogas is injected into the column at  
 448 ambient temperature and pressure allowing CO<sub>2</sub> molecules to adsorb on the materials.  
 449 Regeneration, on the other hand, is conducted by directly heating the column or injecting hot  
 450 air, N<sub>2</sub> gas, or steam into the column (Ntiamoah et al. 2016). The regeneration rate is dependent  
 451 on temperature. Indeed, higher temperature results in a faster regeneration rate. In comparison  
 452 to pressure swing adsorption, the regeneration time usually is longer. After regeneration, the  
 453 column is cooled down to ambient temperature. N<sub>2</sub> gas is applied to both cool and clean the  
 454 column for the next adsorption cycle. Temperature swing adsorption is mainly applied to  
 455 capture CO<sub>2</sub> from the power station and utilise the wasted heat in the regeneration process.

456



457

458 **Figure 3.** Basic principle of pressure swing adsorption

459

### 460 6.3. Membrane separation

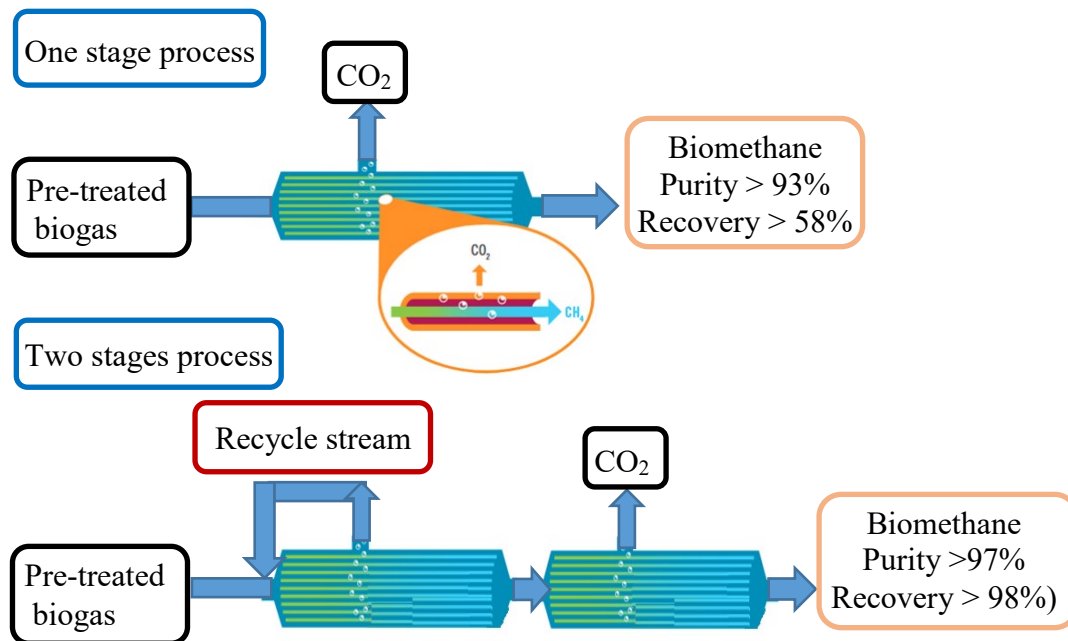
461 The principle of membrane separation methods is that gases permeate through the  
462 membrane pores at different selectivity, i.e., highly permeable to CO<sub>2</sub> (small molecule) and  
463 impermeable to CH<sub>4</sub> (large molecule). In general, membrane suitable for biogas upgrading is  
464 20 more permeable to CO<sub>2</sub> than to CH<sub>4</sub>. The CO<sub>2</sub>-rich exhaust gas from the membrane  
465 separation can be used to produce highly pure CO<sub>2</sub> suitable for the food and beverage industry  
466 (Esposito et al. 2019). Esposito et al. (2019) evaluated the first large scale industrial biogas  
467 upgrading plant to produce CH<sub>4</sub> and CO<sub>2</sub> from membrane separation simultaneously,  
468 liquefying and cryogenic units. The residual CO<sub>2</sub> from five membranes line was combined and  
469 subjected to a liquefying, compression, drying, and cooling. High purity CO<sub>2</sub> (99.9%) was  
470 achieved after cooling to -30°C to separate N<sub>2</sub>, O<sub>2</sub>, and trace CH<sub>4</sub>.

471 Membrane separation is available in different designs. Typical operating pressures are 7 to  
472 20 bars (Peppers et al. 2019). To achieve high methane purities, the tube bundles are connected  
473 in two-stage or three-stage cascades. The two-stage cascade provides higher CH<sub>4</sub> and maintains  
474 higher recovery than a single cascade. In the two-stage cascade, a circulation loop returns the  
475 gas from the first membrane back to the inlet, while the enriched CH<sub>4</sub> continue to the second  
476 membrane (Fig. 4).

477 Key advantages of membrane separation include modular and compact design with less  
478 moving parts. However, membrane separation is still an emerging technology with limited  
479 practical experience. Moreover, the energy requirement is between 0.18 to 0.33 kWh/Nm<sup>3</sup> of  
480 biogas (Makaruk et al. 2010). Methane loss of up to 2% has been reported in some laboratory-  
481 scale studies (Baena-Moreno et al. 2020). Peppers et al. (2019) recently investigated the  
482 feasibility of membrane separation for 100 Nm<sup>3</sup>/h biogas plant. The results demonstrated that  
483 pre-treatment of other gas is necessary to protect the membrane and ensure high CH<sub>4</sub> purity  
484 (Baena-Moreno et al. 2020). Although biogas quality satisfied the standard, the overall cost



485 analysis revealed low economic viability at small scale ( $< 100 \text{ Nm}^3/\text{h}$  biogas flowrate) (Peppers  
486 et al. 2019).



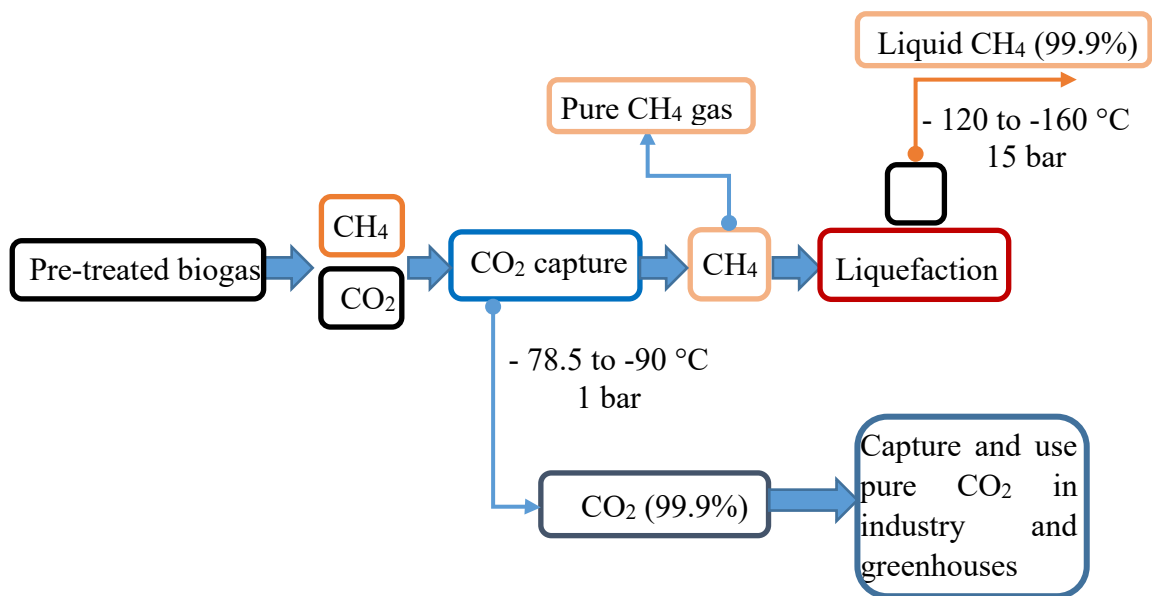
487  
488 **Figure 4:** Physical and technical principle of membrane separation

489  
490 6.4 Cryogenic technology

491 Cryogenic treatment is based on the principle that gases condense (become liquid) or re-  
492 sublimate (become solid) at low temperatures or high pressures. CO<sub>2</sub> and CH<sub>4</sub> have different  
493 condensation temperatures. The CO<sub>2</sub> re-sublimates at -78.5 °C and 1 bar while CH<sub>4</sub> remains  
494 gaseous. The solid CO<sub>2</sub> and gaseous CH<sub>4</sub> can be separated through rectification (i.e., counter-  
495 current distillation). Because of this principle, cryogenic treatment can achieve very pure CH<sub>4</sub>  
496 (up to 99.9% by volume), CO<sub>2</sub> (up to 99.9% by volume) with less than 1% methane loss.  
497 However, the technology is still under development, and its market readiness has not yet been  
498 fully established.

499 A ubiquitous and significant obstacle to this technology is the high energy required for  
500 refrigeration and compression of the raw biogas. The energy consumption is approximately  
501 10% of the generated methane. Another challenge is to ensure that frozen CO<sub>2</sub> does not clog

502 the equipment in the gas refrigeration process. In this regard, other biogas impurities must be  
 503 carefully removed. However, possible options to strengthen this technology are available. The  
 504 energy used to condense initial biogas can be recovered if the produced biomethane is to be  
 505 liquefied. Biomethane liquefaction at  $-125\text{ }^{\circ}\text{C}$  and 15 bar leverage synergies in the production  
 506 of cold gas, thus minimising the energy consumption in both steps. Likewise, frozen  $\text{CO}_2$  can  
 507 be utilised as dry ice in some industrial applications (Fig. 5) (Esposito et al. 2019). Thus,  
 508 cryogenic treatment is starting to become commercially competitive.



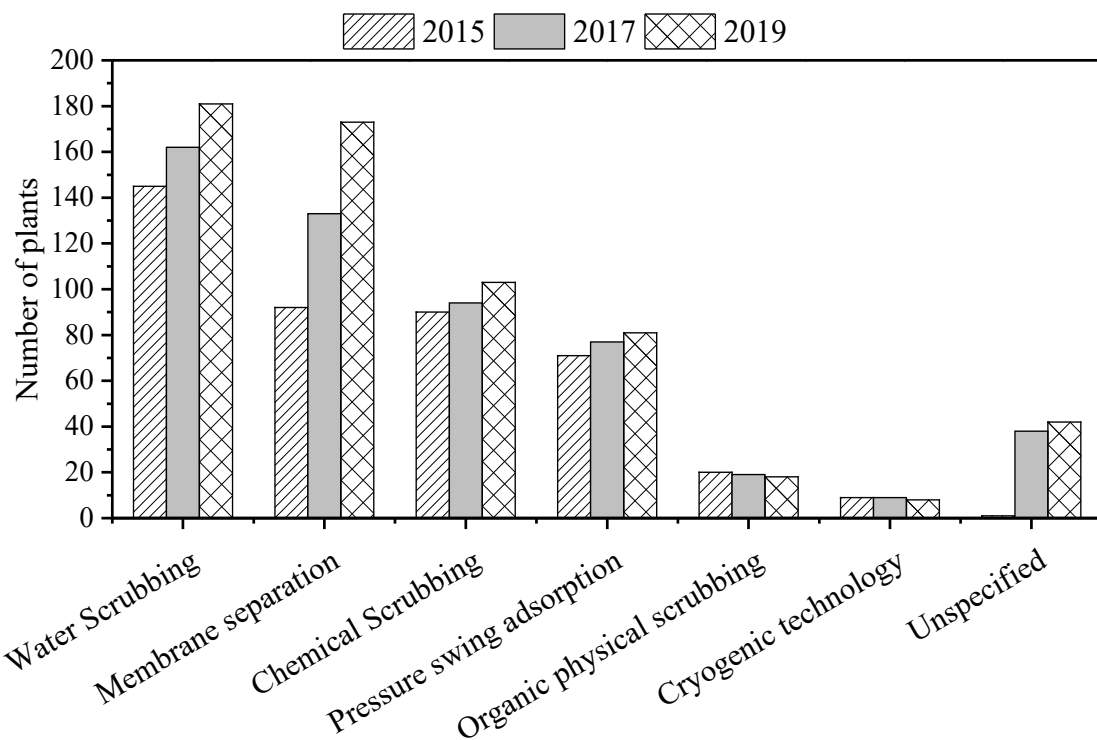
509  
 510 **Figure 5.** Principle of cryogenic biogas upgrading with potential to capture pure  $\text{CO}_2$  and  
 511 liquefy  $\text{CH}_4$

### 512 6.5 Current full-scale application

513 The number of full-scale plants utilising biogas upgrading technologies is increasing (Fig.  
 514 6). Physical scrubbing using water (i.e. water scrubbing) is the dominant technology. In 2019,  
 515 there have been 181 plants in operation. Water scrubbing is a simple process in comparison to  
 516 others technologies. Its major drawback is high water volume requirement. Reusing secondary  
 517 or tertiary effluent for scrubbing can reduce overall cost. The market share of membrane  
 518 separation technology has grown significantly over the last five years. The number of plants

519 increased from 92 (2015) to 173 (2019) (Fig. 6). Key advantages of membrane separation  
 520 include robust design with less moving parts, modular design, and a small physical footprint.  
 521 Recent scientific progress in materials engineering and science has also resulted in better  
 522 membrane performance. With the accumulation of practical experience, it is expected that  
 523 membrane separation will be highly adopted in the near future. Chemical scrubbing, organic  
 524 physical scrubbing, pressure swing adsorption, and cryogenic technology have a small number  
 525 of new instalment over last five years. Cryogenic technology could hold promise for future  
 526 development once the benefit of pure CO<sub>2</sub> harvesting for dry ice production is realised  
 527 (Esposito et al. 2019).

528



529

530 **Figure 6:** Biogas upgrading plants in countries employing different technologies from over  
 531 the last 5 years.

532 The comparison of common biogas upgrading technologies is summarised in Table 6. A  
 533 direct comparison among these technologies is not possible since their selection can depend on  
 534 multiple factors beyond those summarised in Table 6. Nevertheless, some generalisation can  
 535 be made. An estimated OPEX for plant with capacity of 1000 Nm<sup>3</sup>/h indicates that water

536 scrubbing and membrane separation have low operating and maintenance costs (Table 6).  
 537 Membrane separation can also achieve the highest biomethane quality with moderate energy  
 538 consumption and methane loss. Data in Table 6 are consistent the number of full-scale biogas  
 539 upgrading plants currently in operation. Water scrubbing and membrane separation are the two  
 540 most prevalent biogas upgrading technologies (Figure 6). Figure 6 also shows a significant  
 541 increase in the number of membrane-based biogas upgrading plants over the 2014-2019 period.

542 **Table 6:** Reported energy consumption (kWh/Nm<sup>3</sup>) of different technologies. Source:  
 543 (Singhal et al. 2017, Patterson et al. 2011, Masebinu et al. 2014, Vrbová et al. 2017)

Technologies	Biomethane quality (CH <sub>4</sub> %)	Energy consumption (kWh/Nm <sup>3</sup> )	Methane loss (vol %)	Cost for 1000 Nm <sup>3</sup> /h plant	
				CAPEX (million €)	OPEX (€/year)
Water scrubbing	95–98	0.2–0.5	0.5 - 5	1	15,000
Organic physical scrubbing	93–98	0.1–0.33	1 - 4	1	39,000
Chemical scrubbing	<98	0.05–0.18	0.5	2	59,000
Pressure swing adsorption	<98	0.16–0.43	1.5 - 2.5	1.75	56,000
Membrane separation	90–99	0.18–0.35	0.5 - 2	2	25,000
Cryogenic	99	0.18–0.25	0.1	n.a	n.a

544

## 545 6.6. Emerging biotechnology platforms for biogas upgrading

### 546 6.6.1 Technologies to improve biogas quality from AcoD

547 Biological biogas upgrading targets different microbial functional groups in the AcoD  
 548 process to facilitate its function to serve a specific aim. Anaerobic digestion is a biochemical  
 549 process that involves four groups of microorganisms, namely hydrolysers, acidogens,  
 550 acetogens, and methanogens (Nguyen et al. 2019). Biological desulphurisation is one example  
 551 that facilitates the function of sulphur-oxidising microorganisms to reduce H<sub>2</sub>S in biogas. The  
 552 success of the biological desulphurisation (Section 3) sets a foundation for further exploration  
 553 to develop biological biogas upgrading technology. In this regard, the presence of  
 554 hydrogenotrophic methanogens in the AcoD is of particular interest. Hydrogenotrophic

555 methanogens mainly use H<sub>2</sub> as electron-donating sources for the reduction of CO<sub>2</sub> to methane.  
556 Thus, it is hypothesised that through the exogenous addition of H<sub>2</sub> into the digester, CO<sub>2</sub> can  
557 be converted to CH<sub>4</sub> to achieve a two-fold benefit: high CH<sub>4</sub> and low CO<sub>2</sub> content in biogas.  
558 Wahid et al. (2019) observed that the addition of H<sub>2</sub> at a ratio of 4 to 1 mole of CO<sub>2</sub> resulted in  
559 94 and 3% of CH<sub>4</sub> and CO<sub>2</sub>, respectively, in the biogas. Likewise, Bassani et al. (2015)  
560 achieved 89 and 85% CH<sub>4</sub> content in biogas from mesophilic and thermophilic digesters,  
561 respectively, after H<sub>2</sub> addition.

562 Although the methane content in the final biogas is higher after H<sub>2</sub> addition, this technology  
563 is still at its infancy with results from laboratory-scale studies only. There are many drawbacks.  
564 Residual H<sub>2</sub> in the biogas is one limitation. Wahid et al. (2019) observed up to 3% of H<sub>2</sub> in the  
565 biogas, which is higher than the biomethane quality standard for natural gas injection and  
566 transport fuel. Injection of H<sub>2</sub> into the digester can increase the pH (i.e., due to CO<sub>2</sub> depletion)  
567 and influence the process stability (Luo et al. 2013). pH over 8.5 can inhibit the methanogenic  
568 activity (Nguyen et al. 2019). In addition, H<sub>2</sub> injection can increase the hydrogen partial  
569 pressure that may inhibit the acetogenesis.

570

#### 571 6.6.2 Biocatalytic enzyme enhance CO<sub>2</sub> capture efficiency

572 Research efforts to enhance the efficiency of adsorbents (adsorbent rate and capacity) are in  
573 the trajectory to reduce the energy cost of the biogas upgrading technologies. Using enzymes  
574 such as the carbonic anhydrases to convert CO<sub>2</sub> and water to bicarbonate (reaction described  
575 below), could contribute to reducing the energy cost of the CO<sub>2</sub> removal step. In the chemical  
576 absorption method, the energy requirement is determined by the solvent specific heat of  
577 reaction and solvent capacity to take up CO<sub>2</sub>. Consequently, if a solvent has a low reaction heat  
578 and high capacity, energy saving can be achieved (Closmann et al. 2009, Gundersen et al. 2014,  
579 Kunze et al. 2015). Amines and alkali carbonates are potential solvent candidates, but they

580 suffer slow absorption kinetics (Kunze et al. 2015, Beiron et al. 2019). Bicarbonate formation  
581 is the rate-limiting step of the absorption of CO<sub>2</sub>.

582 Enzymes can act as an activator to enhance the absorption kinetics. Indeed, Kunze et al.  
583 (2015) demonstrated that the addition of carbonic anhydrase at 0.2 (wt %) to 30 wt % MEA  
584 and K<sub>2</sub>CO<sub>3</sub> improved the absorbed volume by a factor > 4. Likewise, MDEA absorption  
585 capacity was increased by a factor of 3 after the addition of the carbonic anhydrase enzyme  
586 (Vinoba et al. 2013). The pilot-scale testing performed at 70 °C, revealed that enzyme addition  
587 was a technically feasible method. Thus, the biotechnology enzyme will help to advance the  
588 enzyme addition technology through enhancing temperature resilience. However, to the best  
589 of our knowledge, no study has investigated the application of this technology for biomethane  
590 production from biogas.

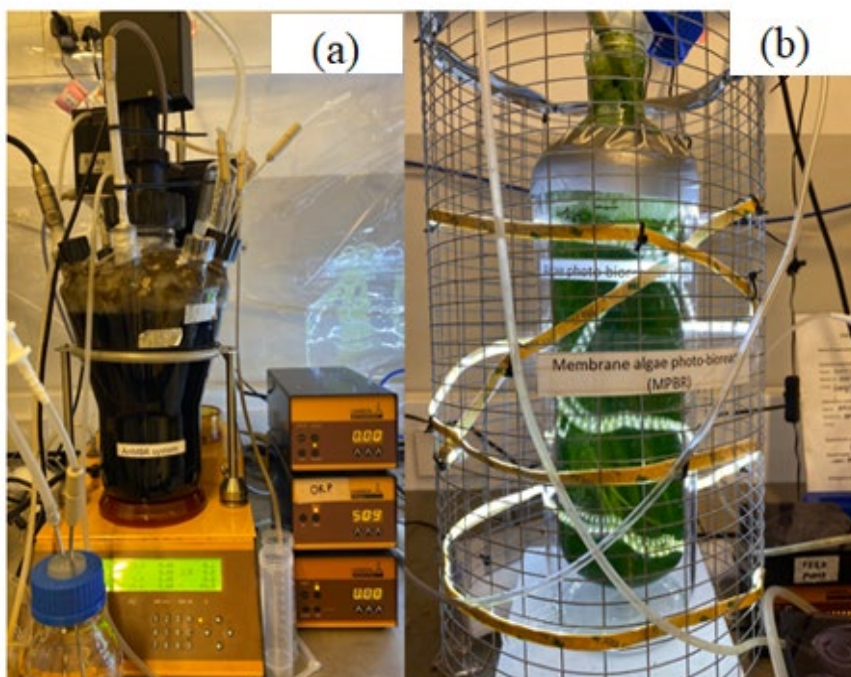
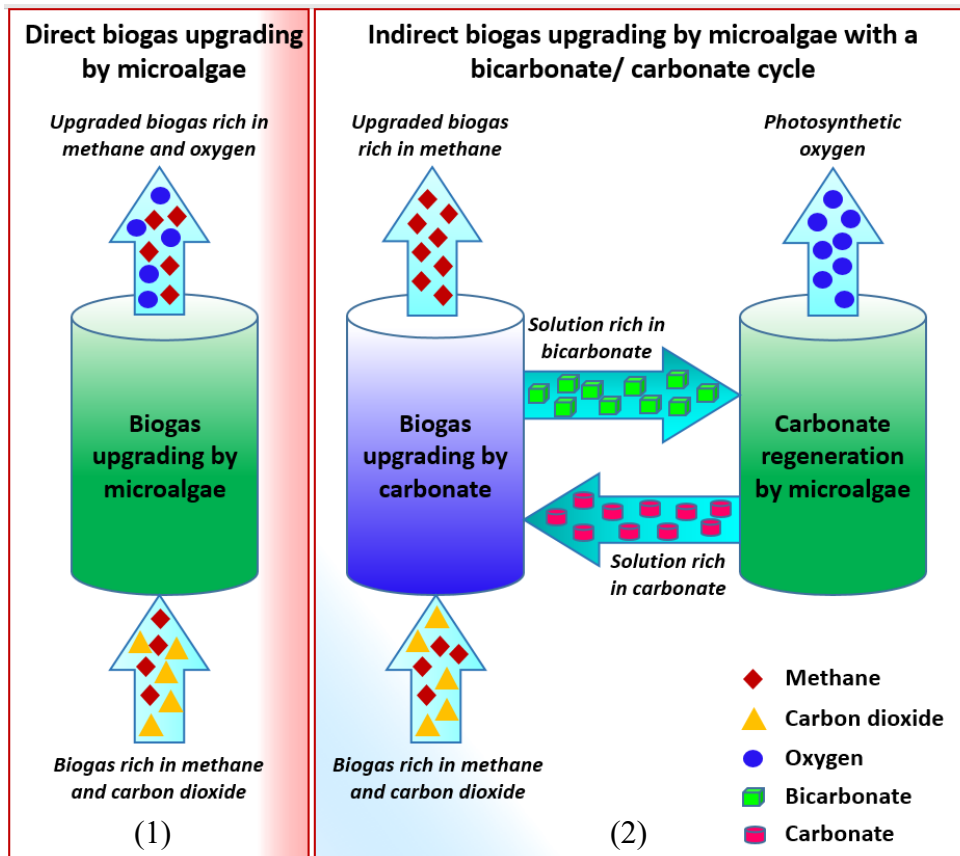
591

### 592 6.6.3 Microalgae for CO<sub>2</sub> capture from biogas

593 Microalgae are autotrophic microorganisms that can fix CO<sub>2</sub> and utilise nutrients (nitrogen  
594 and phosphorus) to produce biomass using light. Microalgae biomass can be used in an array  
595 of valuable bioproducts such as food products, nutraceuticals, feed, pharmaceuticals,  
596 biopolymers, bioplastics, and bulk chemicals (Fabris et al. 2020, Vu et al. 2020a). Therefore,  
597 the application of microalgae to capture CO<sub>2</sub> from biogas can have multi-fold benefits,  
598 including i) reduction in CO<sub>2</sub> content and the associated increase in CH<sub>4</sub> content; ii) production  
599 of valuable biomass and iii) removal of nutrients from water and wastewater (Sutherland et al.  
600 2019). In this process, the biogas generated from the anaerobic digester is fed into a  
601 photobioreactor where microalgae uptake CO<sub>2</sub> (Fig. 7) – a direct approach. This configuration  
602 was first introduced by Converti et al. (2009), who combined a mixed sludge anaerobic digester  
603 with a photobioreactor leading to biogas production with CH<sub>4</sub> content above 70%. Since then,  
604 higher methane content in the final biogas has been achieved with similar systems (Yan et al.  
605 2013, Nagarajan et al. 2019, Bose et al. 2020). Yan et al. (2013) obtained a biomethane (92%

606 CH<sub>4</sub>) through optimisation of culture conditions for the microalgae *Chlorella sp.* However, the  
607 research has identified several challenges that require future development for the emergence of  
608 this green technology (Nagarajan et al. 2019, Bose et al. 2020). [The mass transfer and CO<sub>2</sub>](#)  
609 [solubility in the microalgal culture media is the first limitation \(Bose et al. 2020\)](#). Unlike the  
610 water scrubbing process, gas is injected into the microalgal culture at atmospheric pressure and  
611 room temperature, limiting CO<sub>2</sub> solubility in the growth medium and leading up to 90% of  
612 input gas lost (de Godos et al. 2014). A second limitation is the high methane loss due to its  
613 solubility in a large volume of microalgal culture media. A third limitation is difficulty in  
614 harvesting the final biogas. If biogas is purged into the photobioreactor, an enclosed system is  
615 needed to collect the outlet gas. This requirement can limit the design for photobioreactors and  
616 microalgal growth. Another limitation is the lack of high CO<sub>2</sub> tolerant microalgal species. The  
617 high concentration of CO<sub>2</sub> in water reduces the pH value to below 6.0, which is detrimental to  
618 microalgal growth (i.e., disruption of cell membrane permeability and photosynthesis)  
619 (Sutherland et al. 2020). It is also a challenge because of the introduction of oxygen from the  
620 microalgal photosynthesis in to the final biogas.

621 Another approach to mitigate the limitation of the direct method is indirect biogas upgrading  
622 systems (Fig. 7). In this approach, CO<sub>2</sub> can be captured in a carbonate solution such as  
623 potassium carbonate. The potassium carbonate solution provides high quality of methane to be  
624 achieved. The saturated carbonate solution then is fed into the microalgal culture. Microalgae  
625 utilise bicarbonate as a carbon source for growth, regenerating the carbonate for a next biogas  
626 upgrading cycle. This approach, however; only limits to some specific microalgal species  
627 which can tolerate high ion strength and alkali environment (Xia et al. 2015).



628

629 **Figure 7:** Schematic diagram of using microalgae for biogas upgrading (1) a direct and (2)

630 indirect approach and a photography of anaerobic digester (a) coupled with a photobioreactor

631 (b) for CO<sub>2</sub> capture and microalgae (*Chlorella vulgaris*) as an example of the direct approach



632 in our laboratories (Vu et al. 2020b). The schematic diagram was adapted from Xia et al.  
633 (2015).

## 634 7. Future perspectives

635 As AcoD continues to be adopted at WWTPs around the world, the demand for biogas  
636 upgrading technology to better utilise the surplus biomethane will continue to grow. Water  
637 scrubbing is currently the most widely applied technology due to low capital and operation  
638 cost. On the other hand, membrane separation has the highest growth. It is expected that  
639 membrane separation will overtake water scrubbing to become the most dominant technology  
640 for the biogas upgrading. It is also noteworthy that the technical readiness level for biogas  
641 upgrading is high with a variety of technologies that have been implemented at full-scale. In  
642 addition to water scrubbing and membrane separation, other technologies such as chemical  
643 scrubbing and pressure swing adsorption will continue to be utilised for biogas upgrading on a  
644 case to case basis.

645 Biogas upgrading to biomethane provides opportunities to tap into potential revenue that  
646 has not been previously utilised (IEA 2020). As discussed in section 3, raw biogas contains  
647 about 35% CO<sub>2</sub>, which can be used for a range of applications. Gaseous CO<sub>2</sub> from the  
648 upgrading process can be used to produce dry ice at a temperature of – 78.5 °C. Unlike  
649 conventional ice, dry ice evaporates during the melting process, leaving no residue. Thus, dry  
650 ice is an appealing alternative to conventional ice in many industrial applications (e.g., food  
651 packing, biological samples transportation, and cleaning). The utilisation of this CO<sub>2</sub> source  
652 can provide additional revenue.

653 The quality of raw biogas could induce additional cost on overall expenditure of the pre-  
654 treatment and biogas upgrading processes. For example, high level of H<sub>2</sub>S in raw biogas can  
655 increase the cost of its removal process (i.e. shorten the lifetime of adsorption column and  
656 increase chemical usage). Technologies to improve raw biogas quality and AcoD performance

657 will have great benefits. To date, biological desulphurisation (i.e., injection air or oxygen into  
658 AcoD) to reduce H<sub>2</sub>S formation has only been demonstrated in the laboratory. The variation in  
659 the performance between laboratory and full-scale studies may suggest more full-scale  
660 experience in the future. It is also recommended to evaluate the impact of air injection on other  
661 performance parameters (e.g., solid removal and biogas yields) and biosolid quality.

662 Biocatalytic enzyme and CO<sub>2</sub> capture by microalgae currently have a low technology  
663 readiness level. There is no study on the use of biocatalytic enzymes for biomethane  
664 production. It is also expected that the industrialized production of enzymes and its stability in  
665 the biomethane application is needed. Numerous questions need to be answered before  
666 deciding on an optimal microalgal biogas upgrading system. It is likely an innovative integrated  
667 system to i) use microalgae to capture CO<sub>2</sub>; ii) to use anaerobic digestate as growth media and  
668 iii) to harvest microalgal biomass to use again as feedstocks for AcoD need to be evaluated in  
669 the upcoming studies.

670

## 671 **8. Conclusion**

672 Through anaerobic co-digesting sewage sludge and organic waste, numerous wastewater  
673 treatment plants (WWTPs) worldwide have achieved energy self-sufficiency and produced  
674 surplus biogas. Natural gas grid injection and transport fuels are attractive applications to utilise  
675 the surplus biogas from WWTPs after biogas upgrading to biomethane. Biogas upgrading  
676 technologies include water, organic and chemical scrubbing, pressure swing adsorption,  
677 membrane separation, and cryogenic are commercially available. Amongst them, water  
678 scrubbing is currently the most widely applied technology due to low capital and operation  
679 costs. On the other hand, the membrane separation is expected to be the dominant technology  
680 in the near future. In the 2015-2019 period, membrane process has a significant market growth  
681 (82% increase in new plants). Several emerging biotechnologies to improve biogas quality  
682 from co-digestion accelerate the absorption rate, and capture CO<sub>2</sub> in microalgal culture are

683 highlighted and discussed. Information corroborated in this review demonstrates the possibility  
684 to transform WWTPs to net energy producers through the combination of co-digestion and  
685 biogas upgrade.

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