Comparison of Nanofiltration with Reverse Osmosis in Reclaiming Tertiary Treated Municipal Wastewater for Irrigation Purposes

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31 Abstract

32 This study investigates the performance of reverse osmosis (RO) and nanofiltration 33 (NF) for the reclamation of ultra-filtered treated sewage effluent (TSE) for irrigation of food 34 crops. RO and NF technologies were evaluated at different applied pressures, the 35 performance of each technology was evaluated in terms of water flux, recovery rate, specific 36 energy consumption and quality of permeate. It was found that the permeate from the reverse 37 osmosis (RO) process complied with the Food and Agriculture Organization (FAO) standards 38 at applied pressures between 10 bar and 18 bar. At an applied pressure of 20 bar the permeate 39 quality did not comply with irrigation water standards in terms of chloride, sodium and 40 calcium concentration. It was found that the nanofiltration process was not suitable for the 41 reclamation of wastewater as the concentration of chloride, sodium and calcium exceeded the 42 allowable limits at all applied pressures. In the reverse osmosis process, the highest recovery 43 rate was 36% achieved at an applied pressure of 16 bar. The specific energy consumption at this applied pressure was 0.56 kWh/m³. The lowest specific energy of 0.46 kWh/m³ was 44 45 achieved at an applied pressure of 12 bar with a water recovery rate of 32.7%.

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47 Keywords: Irrigation water; Reverse osmosis; Nanofiltration; Treated sewage effluent;
48 Water reuse.

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54 1. Introduction

55 Water scarcity is one of the most challenging problems that affect agriculture 56 worldwide, especially in arid areas. The united nations estimates that agriculture accounts for 57 70% of water usage around the world (Hafiz et al., 2019). Treated wastewater is an 58 economical solution to be used as irrigation water and a source of nutrients (Shanmuganathan 59 et al., 2015). Treated wastewater can improve soil health and reduce fertilizers consumption. 60 However, treated wastewater may also damage the soil because of the excess salts, 61 pathogens, organics, sodium, and chloride content. The water quality for irrigation water is 62 mainly characterized in terms of total dissolved salts, pH, and different ions and cations 63 concentration (e.g. Na, Cl, NO₃, SO₄, PO₄, K, Ca, and Mg). Enhancing the quality of treated 64 wastewater to meet irrigation standards has become a must practice. In order to reach the 65 required quality of treated wastewater membrane technologies are considered to be a critical 66 element.

67 (Shanmuganathan et al., 2015) studied the possibility of enhancing the quality of 68 micro-filtered treated sewage effluent using nanofiltration (NF) and reverse osmosis (RO). 69 The study showed that using NF and RO alone could not produce permeate, which meets 70 irrigation standards. Irrigation suitable permeate was produced using an NF-RO hybrid 71 system. Also, it was found that utilizing NF before RO reduced the RO membrane fouling. 72 (Li et al., 2016) studied the performance of advanced treatment of municipal wastewater by 73 nanofiltration. The study evaluated the effect of operating pressure and feed solution pH. The 74 experimental results showed that optimum performance was achieved using a 12 bar 75 pressure, pH = 4 and a flow rate of 8 LPM. Protein-like substances of high molecular weight 76 (MW) are the dominant foulants on the membrane surface. A pilot-scale study conducted by 77 Oron et al. (2006) showed that by using a hybrid ultrafiltration- reverse osmosis (UF-RO) 78 technology, water suitable for irrigation could be produced from secondary treated municipal

79 wastewater (Oron et al., 2006). The cost of the process was between 0.16 and 0.24 US\$/m³ 80 water. Mrayed et al. (2011) applied a hybrid nanofiltration-reverse osmosis (NF-RO) system to produce irrigation water from secondary treated effluent. They used polyacrylic acid 81 82 (PAA) as a chelating agent. The addition of PAA helped in the formation of covalent bonds 83 among different nutrients in the feed, which improved the rejection rate for those nutrients. 84 (Egea-Corbacho et al., 2019) tested the performance of a pilot-scale nanofiltration membrane 85 for the treatment of secondary treated wastewater effluent. It was found that the product 86 water quality complies with Spanish Royal Decree 1620/2007. This was concluded by 87 considering E Coli, TSS and turbidity, but the authors did not compare the concentration of 88 various elements in the permeate water with allowable limits (i.e. phosphates, nitrates, total 89 dissolved solids, ammonium, sodium and chloride). A study from (Chon et al., 2012) used a 90 hybrid technology comprising of membrane bioreactor and nanofiltration to produce 91 irrigation water from municipal wastewater. It was found that the physicochemical properties 92 and molecular weight cut off were the most critical aspect in the removal of nutrients from 93 the water. (Gu et al., 2019) evaluated the performance of trihybrid anaerobic membrane 94 bioreactor (AnMBR)-reverse osmosis (RO)-ion exchange (IE) process for reclamation of 95 microfiltered municipal wastewater to high-grade clean water. The net energy consumption of the process was 1.16 kWh/m³, and product water was found to be suitable for industrial 96 97 and indirect potable applications. Hafiz et al. (2019) used FO to produce irrigation water from 98 treated sewage effluent (TSE). The feed solution and draw solution for the FO was TSE and 99 an engineered fertilizing solution (0.5 M NaCl & 0.01 M (NH₄)₂HPO₄), respectively. The 100 draw solution was regenerated using RO. The specific power consumption was between 2.18 101 and 2.58 kWh/m³. (Liu et al., 2011) evaluated the effectiveness of nanofiltration and reverse 102 osmosis in the treatment of treated textile effluent in terms of salinity reduction and COD 103 rejection. The results showed that nanofiltration exhibited more severe flux decline compared

to reverse osmosis (RO) because of the higher porosity and membrane fouling of the
nanofiltration membrane. RO showed higher total salts rejection compared to NF. (Qi et al.,
2020) analyzed pollutants removal efficiency and operating costs of municipal wastewater
treatment plants in china. Significant difference in removal efficiencies was observed among
various pollutants, with the highest removal efficiency in BOD₅ and lowest removal
efficiency in TN. Higher nitrogen removal should be achieved to obtain the desired water
quality outcomes.

So far, previous studies evaluated the performance of various membrane processes for the reclamation of secondary treated sewage effluent. Little information is available for the performance of the nanofiltration and the reverse osmosis membranes in the generation of irrigation water from tertiary treated wastewater. It is recommended to select a single membrane process that can generate high-quality irrigation water from treated sewage effluent at minimal energy requirement. The product water quality must comply with the Food and Agriculture Organization (FAO) standards.

The objective of this paper is to evaluate the efficiency of nanofiltration and reverse osmosis to treat ultra-filtered tertiary treated sewage effluent (TSE) for the product water to be used in irrigation for food crops. The performance of each technology was evaluated under different applied pressures in terms of water flux, recovery rate, energy consumption and quality of permeate.

123 2. Materials and setup

124 2.1 Feedwater

125 Ultra-filtered tertiary treated sewage effluent (TSE) was used as feedwater to the
126 nanofiltration and reverse osmosis processes. TSE was collected from a wastewater treatment
127 plant located in Doha, Qatar. The wastewater treatment plant consists of preliminary,

128	secondary and tertiary treatment processes. The tertiary treatment process consists of a
129	multimedia filter followed by ultrafiltration and UV disinfection. The characteristics of the
130	collected, treated sewage effluent are summarized in Table 1. The max limit of the listed
131	parameters was recommended by FAO (Ayres and Westcot, 1985; Lejalem et al., 2018;
132	Parlar et al., 2019). The use of this feed water on food crops was unsuitable because of
133	excessive TDS and high ions-cations. The concentration of heavy metals was below the
134	maximum limit recommended by FAO. The conductivity of samples was measured using
135	OAKTON PCD650 multi-meter. Anions concentration was measured by ion chromatography
136	(Metrohm 850 Professional IC), and cations concentration was measured using plasma
137	emission spectroscopy (iCAP 6500-ICP-OES CID) (Thermo Scientific). Before measuring
138	the concentration of anions and cations, samples with a conductivity value above 1 mS/cm
139	were diluted using deionized water to a conductivity value below 1 mS/cm. This is done to
140	eliminate the interference of high peaks of Na and Cl, which may affect the readings of other
141	elements. The turbidity was measured using a turbidity meter (Hach 2100p). Metal
142	concentration was measured using ICP-MS (Nexion 300D).

143 Table 1. Characteristics of tertiary treated sewage effluent (feed water)

Parameter	Value	Max Limit	Standard Method
TDS (nnm)	1461 ± 5	750	APHA 2540 C. Total Dissolved
TDS (ppin)	1401 ± 3		Solids Dried at 180 °C
Turbidity (NTU)	0.2 ± 0.1	2	APHA 2130 B. Nephelometric
Turblatty (NTO)	0.2 ± 0.1	2	Method
EC (mS/cm)	2.56 ± 0.2	0.7	APHA 2510 B. Conductivity
Fluoride (ppm)	0.27 ± 0.2	1.5	
Chloride (ppm)	897.5 ± 0.2	106.5	- APHA 4110 Determination
Bromide (ppm)	0.96 ± 0.2	1	of anions by ion chromatography
Nitrate (ppm)	25.84 ± 0.2	20	
Sulfate (ppm)	320.3 ± 0.2	400	-

Sodium (ppm)	200.3 ± 0.2	69	$\Delta PH \Delta 3120$ Determination
Potassium (ppm)	12.4 ± 0.2	10	of metals by plasma emission
Calcium (ppm)	87.7 ± 0.2	40	spectroscopy
Magnesium (ppm)	21.4 ± 0.2	24	speciloscopy
Boron (ppb)	158.97 ± 0.1	500	
Vanadium (ppb)	0.11 ± 0.1	100	_
Manganese (ppb)	11.54 ± 0.1	200	_
Cobalt (ppb)	0.17 ± 0.1	50	
Nickel (ppb)	23.11 ± 0.1	200	EPA Method 200.8
Copper (ppb)	13.08 ± 0.1	200	
Zinc (ppb)	151.58 ± 0.1	2000	
Cadmium (ppb)	0.2 ± 0.1	10	
Beryllium (ppb)	2.02 ± 0.1	100	

145 2.2 Experimental setup

146 A schematic sketch for the bench-scale membrane testing skid is shown in Fig.1. A 147 crossflow CF042D cell made of natural acetal copolymer (Delrin) provided by Sterlitech was 148 used in the nanofiltration and reverse osmosis processes. The cell dimensions are 12.7 x 8.3 x 149 10 cm with active inner dimensions of 4.6 x 9.2 cm and 0.23 cm slot depth. Two tanks were 150 used to store the feed and the permeate water. A M-03S HYDRACELL pump (230V, 50HZ, 151 3PH, 6.7 LPM) was used to pressurize the feed solution through the membrane. A water 152 chiller (PolyScience Chiller) was used to maintain the feedwater temperature at room temperature (25 ± 2 °C). A concentrate/back pressure control valve was used to control the 153 154 water flow through the system and to regulate pressure in the system. Flow meters (Sterlitech 155 Read Panel Mount Flow Meter) were used to measure the flow rate at specific points in the 156 system. A digital balance (Mettler Toledo – ICS 241) was connected to a computer to 157 measure the permeate flux in the system. A specific quantity (3 L) of tertiary treated sewage 158 effluent (TSE) was used as a feed solution in both processes. The applied pressure in the RO

and the NF processes varied between 10 bar and 20 bar with an increase of 2 bar for each
experiment. The flow rate was 3.5 LPM and the experimental running time was 4 h. The used
RO membrane was BW30LE produced by DOW FILMTEC. The used RO membrane is a
polyamide – TFC membrane with a pore size of 100 Da. The used NF membrane was NF90
produced by DOW FILMTEC. The used NF membrane is a polyamide – TFC membrane
with a pore size of 200-400 Da.





Figure 1. Schematic diagram of cross flow lab scale membrane test skid

166 2.3 Error estimation

167 All experiments were performed in triplicates and the reported results are the average168 of the three experimental trials. The error bars represent the standard deviation of each

169 reading.

170 3. Results and Discussion

171 3.1 Effect of feed pressure on water flux and recovery rate

172 The water flux (J_w) in the RO process and the NF process was calculated using Eq.1 173 (Thabit et al., 2019):

$$J_w = \left(\frac{V_p}{A_m \times t}\right) \tag{1}$$

Here, V_P is the volume of the permeate (L), A_m is the area of the membrane (m²), t is the 174 operating time (h). Figure 2 (a) and (b) shows the change of water flux with time in the RO 175 176 and NF process, respectively. It can be seen from Fig.2 (a) and (b) that the water flux decreased with time at all applied pressures. The decrease in the water flux with time is due 177 178 to the concentration of the feed solution where the reject solution was recycled back into the 179 system and membrane fouling. TSE contains traces of organic matter, which could 180 accumulate on the surface of the membrane and cause fouling (Ortega-Bravo et al., 2016). It 181 can also be seen from Fig.2 (a) that in the RO process the water flux at an applied pressure of 182 12 - 20 bar was within the same range, but at an applied pressure of 10 bar the water flux was much lower. In the NF process, the water flux at an applied pressure of 12 bar was higher 183 184 than the water flux at the other applied pressures. This was done to compare the performance 185 of the RO and the NF processes at the different applied pressures; the average water flux was 186 calculated.



Figure 2. Water flux using TSE as feed water at different applied pressure in (a) Reverse osmosis (b)Nanofiltration

Figure 3 shows the average water flux for RO and NF at different feed pressures. In RO, when the applied pressure was 10 bar the average water flux was 21.3 L/m^2 .h. When the applied pressure increased to 12 bar the average water flux increased by 69%, to reach a value of 68.1 L/m².h. At a 14 bar applied pressure, the average water flux further increased to reach a value of 71.5 L/m².h. At an applied pressure of 16 bar the average water flux reached

a maximum value of 77.7 L/m².h. As the applied pressure further increased the average water 192 193 flux decreased. Where at an applied pressure of 18 bar and 20 bar the average water flux was 71.6 L/m².h and 67.5 L/m².h, respectively. A different trend was observed in the NF process. 194 195 Excluding the 10 bar applied pressure experiment, it was found that the average water flux 196 decreased as the applied pressure increased (Fig.3). The maximum average water flux was 44.5 L/m^2 .h obtained at an applied pressure of 12 bar. When the applied pressure increased to 197 14 bar the average water flux decreased by almost 37% to reach a value of 28.1 L/m².h. As 198 199 the applied pressure further increased the average water flux kept decreasing to reach a minimum value of 20.6 L/m².h at an applied pressure of 20 bar. The water permeability is 200 201 expected to increase as the feed pressure increases; however, applying excessive pressure 202 may result in excessive accumulation of foulants on the surface of the membrane which may 203 result in lower average water flux (Jiang et al., 2017). The average water flux of RO was 204 higher than NF. In effect, the NF membrane is prone to fouling compared to RO membrane, 205 because the nanofiltration membrane has a rougher, thicker and hydrophobic surface when 206 compared to the reverse osmosis membrane (Xu et al., 2010). The lowest average water flux 207 for RO and NF was obtained at an applied pressure of 10 bar. The water diffusion through the 208 membrane starts to occur when the applied pressure exceeds the natural osmotic pressure of 209 the solution. The osmotic pressure of the feed solution was almost 9 bar; consequently, a feed 210 pressure of 10 bar was not enough to acquire high water flux.



The recovery rate (R%) has been calculated using Eq.2 (Singh et al., 2019):

Figure 3. Average water flux of RO and NF at different feed pressure

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$$\%R = \left(\frac{V_P}{V_F}\right) \times 100\% \tag{2}$$

212 Here, V_P and V_F are the volume of the permeate and the feed solution, respectively. Figure 4 213 shows the recovery rate for RO and NF at different feed pressures. In the RO process, the 214 highest water recovery was 36% obtained at an applied pressure of 16 bar. The lowest water 215 recovery was 10.2% obtained at an applied pressure of 10 bar. When the applied pressure was 216 12 bar, the water recovery was 32.7% which is 3.3% lower than the maximum water recovery 217 obtained at 16 bar feed pressure. When the applied pressure was 14 bar, the water recovery 218 was 35.2% which is only 0.8% lower than the maximum water recovery obtained at 16 bar 219 feed pressure. As the feed pressure increased more than 16 bar, the water recovery decreased. 220 At a feed pressure of 18 bar the water recovery was 34.3%, and at a feed pressure of 20 bar 221 the water recovery was 33.8%. As mentioned earlier, applying excessive pressure may result 222 in the accumulation of foulants on the membrane surface, which could result in lower water

223 production (Qasim et al., 2019). In nanofiltration, the maximum recovery rate was 21.9% 224 obtained at a feed pressure of 12 bar. The recovery rate decreased dramatically from 21.9% to 225 14.9% at an applied pressure of 14 bar. As the pressure increased, the recovery rate decreased 226 to reach a value of 10.6% at an applied pressure of 20 bar. The recovery rate of nanofiltration 227 was found to be lower than reverse osmosis. In addition, the effect of pressure is more 228 apparent in nanofiltration. The NF membrane has a higher negative charge compared to the 229 RO membrane, which makes the NF membrane more prone to fouling (Zou et al., 2018). In 230 addition, the nanofiltration membrane has a rougher, thicker and less hydrophilic surface 231 when compared to the reverse osmosis membrane (Chen et al., 2017). The used nanofiltration 232 membrane (NF 90) had a contact angle of 67.5° while the used reverse osmosis membrane 233 (BW30LE) had a contact angle 33° (Gryta et al., 2012). This could have also caused the 234 lower recovery rates obtained by NF compared to RO.



Figure 4. Recovery rate of RO and NF at different feed pressure

SEM images of unused and used RO and NF membranes at different applied
pressures are shown in Fig.5. Fig. 5 (b), (c) and (d) show the used RO membranes at an

237 applied pressure of 12, 16 and 20 bar, respectively. It can be seen from the SEM images that 238 as the applied pressure increased more accumulation of foulant materials occurred on the 239 surface of the membrane. A similar observation was detected on the nanofiltration membrane 240 where the amount of the accumulated foulants increased on the surface of the membrane as 241 the feed pressure increased (Fig.5 (f), (g) and (h)). From the EDX analysis shown in Table 2, 242 it can be seen that after the use of the RO and NF membranes new elements were detected on the surface of the membranes such as sulfur, iron, and phosphorus, which also indicates the 243 244 accumulation of foulant materials on the surface of the membrane. The accumulation of these 245 elements on the surface of the membrane negatively affected the water flux. The 246 accumulation of metal ions on the surface of the membrane may further tie the fouling 247 materials on the membrane material resulting in enhanced compactness of the fouling layer 248 (Sun et al., 2019).





249 Figure 5. SEM images of (a) Clean RO membrane, (b) Tested RO membrane at a feed

- 250 pressure of 12 bar, (c) Tested RO membrane at a feed pressure of 16 bar, (d) Tested RO
- 251 membrane at a feed pressure of 20 bar, (e) Clean NF membrane, (f) Tested NF membrane at
- a feed pressure of 12 bar, (g) Tested NF membrane at a feed pressure of 16 bar, (h) Tested
- 253 *NF membrane at a feed pressure of 20 bar.*
- 254

Membrane		Weight %					
	[C]	[0]	[S]	[Fe]	[P]	[Other]	
RO-clean	87.39	9.53	3.08	0	0	0	
RO-tested	70.64	19.45	3.32	1.79	0.66	4.14	
NF-clean	89.78	6.66	3.56	0	0	0	
NF- tested	73.96	16.32	3.94	1.91	0.48	3.39	

Table 2. EDX analysis of elements wt% on the surface of a clean and tested RO and NF
membranes

258 3.3 Energy consumption

259 The specific energy consumption (E_s) of the RO and the NF processes has been

260 calculated using Eq.3 (Shrivastava and Stevens, 2018):

$$E_s = \left(\frac{P}{n \times \% R}\right) \tag{3}$$

261 Here, P is the applied feed pressure (bar), n is the pump efficiency, $\Re R$ is the recovery rate. 262 The specific energy consumption depends on the applied pressure and the recovery rate; 263 therefore, the lowest energy consumption will be obtained at a high recovery rate using low 264 applied pressure. Figure 6 shows the specific energy consumption of the RO process and the 265 NF process at different feed pressures. In the NF process, the specific energy consumption increased from 0.68 kWh/m³ to 2.35 kWh/m³ at a feed pressure of 12 bar and 20 bar, 266 267 respectively. As shown in Eq.3, the specific energy is a function of applied pressure and 268 recovery rate. At a low feed pressure of 10 bar the specific energy consumption was 1.33 269 kWh/m³. The high specific energy consumption at such low feed pressure is due to the low 270 recovery rate obtained at 10 bar feed pressure (Fig.4). The maximum energy consumption 271 was 2.35 kWh/m³, which was obtained at a feed pressure of 20 bar. The high specific energy 272 consumption at such high applied pressure is due to the low recovery (Fig.4). In the RO 273 process, the same trend was observed where the specific energy consumption increased from 0.46 kWh/m³ to 0.73 kWh/m³ at a feed pressure of 12 bar and 20 bar, respectively. At a low 274 275 feed pressure of 10 bar the specific energy consumption was 1.22 kWh/m³, which is due to 276 the low recovery rate obtained at such low applied pressure as shown in Fig.4. It was found 277 that the NF process at an applied pressure of 12 bar gave the highest water recovery rate and 278 the lowest energy consumption. For the RO process, the lowest energy consumption was 279 found at an applied pressure of 12 bar while the highest water recovery rate was at an applied 280 pressure of 16 bar. The difference in the water recovery rate between the 12 and 16 bar 281 applied pressure in the RO process was only 3.3%. In comparison, the energy consumption 282 was 18% higher at an applied pressure of 16 bar when compared to 12 bar applied pressure. 283 The quality of the produced permeate should be analyzed to investigate which process and 284 which running conditions to be utilized.



Figure 6. Specific energy consumption of RO and NF at different feed pressure.

286 3.4 Product water quality

287 Characteristics of the produced permeate were measured for the RO and the NF 288 processes. The quality of the produced permeate was compared with the Food and 289 Agriculture Organization (FAO) standards (Ayres and Westcot, 1985). As shown in Fig.7 (a) 290 the concentration of the different measured elements in the produced permeate from the RO 291 process at applied pressures between 10 and 18 bar was within the FAO standards. At an 292 applied pressure of 20 bar, multiple parameters such as TDS, conductivity, chloride, sodium 293 and calcium concentration exceeded the allowable limits. In the NF process (Fig.7 (b)), under 294 all applied pressures, the permeate quality did not comply with FAO standards. It was noticed 295 that the elements that did not comply with the standards were chloride, sodium and calcium. 296 The high concentration of these elements in return affected the TDS concentration. For 297 example, at 12 bar applied pressure where the highest water recovery rate was attained the 298 TDS concentration was almost 21% higher than the allowable limit, and the chloride, sodium

- and calcium concentrations were 84%, 61% and 13% higher than the allowable limit,
- 300 respectively. It could be noticed that the NF membrane had a low rejection rate for
- 301 monovalent ions, where the rejection rate for chloride and sodium was only 27% and 12%,
- 302 respectively.
- 303



■ Fluoride ■ Bromide ■ Potassium ■ Nitrate ■ Magnesium ■ Calcium ■ Sulfate ■ Sodium ■ Chloride ■ TDS



Figure 7. Characteristics of the permeate produced from treated sewage effluent (TSE) using (a) Reverse osmosis (b)Nanofiltration at different applied pressures.

It can be inferred that using a single-stage NF is not possible due to low water quality, and it is recommended to use RO at applied pressure between 10 and 18 bar. Selecting the most suitable applied pressure to operate the RO process depends on water flux, recovery rate, energy consumption and product water quality. It was found in the previous sections that the lowest energy consumption of RO was obtained at an applied pressure of 12 and 14 bar. After considering the water quality, it is recommended to use an applied pressure of 14 bar due to higher water quality.

311 4. Conclusions

In this paper, a comparative study was done on the reclamation of tertiary treated sewage effluent (TSE) by nanofiltration and reverse osmosis for water reuse in irrigation. The results reported show that reverse osmosis (RO) is capable of reclaiming tertiary treated 315 sewage effluent (TSE) to be used as irrigation water for food crops. NF is not suitable for the 316 reclamation of wastewater because of the low rejection of monovalent ions. In nano-filtered 317 TSE, the concentration of Na and Cl exceeded the maximum allowable limits recommended 318 by FAO. RO is suitable for the reclamation of wastewater due to high water quality and water flux. In the RO process, the highest recovery rate was 36% achieved at an applied pressure of 319 320 16 bar. The specific energy consumption at this applied pressure was 0.56 kWh/m³. At 14 bar applied pressure, the recovery rate was only 2% lower than that at an applied pressure of 16 321 322 bar, while the specific energy consumption was almost 11% lower. At 12 bar applied 323 pressure, the specific energy consumption was 8% higher than the specific energy at an 324 applied pressure of 14 bar while the recovery rate was 7% higher. It is recommended to use 325 RO process at 14 bar applied pressure for the reclamation of TSE. This is due to the high 326 recovery rate, low energy consumption and high water quality compared to other available 327 technologies. The product water could meet the quality requirements of the Food and 328 Agriculture Organization (FAO). The results from this study might lead to a paradigm shift in 329 the reclamation of tertiary treated wastewater to be used for the irrigation of food crops.

330

Declaration of competing interest

331 The authors declare that they have no known competing financial interests or personal332 relationships that could have appeared to influence the work reported in this paper.

333 CRediT authorship contribution statement

334 MhdAmmar Hafiz: Conceptualization, Data collection, Data analysis, Writing
335 manuscript. Alaa H. Hawari: Conceptualization, Methodology, Review & editing. Radwan
336 Alfahel: Data collection, Data analysis. Ali altaee: Conceptualization, Methodology, Review
337 & editing.

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