# **Comparison of Nanofiltration with Reverse Osmosis in Reclaiming Tertiary Treated Municipal Wastewater for Irrigation Purposes**



### **Abstract**

 This study investigates the performance of reverse osmosis (RO) and nanofiltration (NF) for the reclamation of ultra-filtered treated sewage effluent (TSE) for irrigation of food crops. RO and NF technologies were evaluated at different applied pressures, the performance of each technology was evaluated in terms of water flux, recovery rate, specific energy consumption and quality of permeate. It was found that the permeate from the reverse osmosis (RO) process complied with the Food and Agriculture Organization (FAO) standards at applied pressures between 10 bar and 18 bar. At an applied pressure of 20 bar the permeate quality did not comply with irrigation water standards in terms of chloride, sodium and calcium concentration. It was found that the nanofiltration process was not suitable for the reclamation of wastewater as the concentration of chloride, sodium and calcium exceeded the allowable limits at all applied pressures. In the reverse osmosis process, the highest recovery rate was 36% achieved at an applied pressure of 16 bar. The specific energy consumption at 44 this applied pressure was  $0.56 \text{ kWh/m}^3$ . The lowest specific energy of 0.46 kWh/m<sup>3</sup> was achieved at an applied pressure of 12 bar with a water recovery rate of 32.7%.

 **Keywords:** Irrigation water; Reverse osmosis; Nanofiltration; Treated sewage effluent; Water reuse.

### 1. Introduction

 Water scarcity is one of the most challenging problems that affect agriculture worldwide, especially in arid areas. The united nations estimates that agriculture accounts for 70% of water usage around the world (Hafiz et al., 2019). Treated wastewater is an economical solution to be used as irrigation water and a source of nutrients (Shanmuganathan et al., 2015). Treated wastewater can improve soil health and reduce fertilizers consumption. However, treated wastewater may also damage the soil because of the excess salts, pathogens, organics, sodium, and chloride content. The water quality for irrigation water is mainly characterized in terms of total dissolved salts, pH, and different ions and cations concentration (e.g. Na, Cl, NO3, SO4, PO4, K, Ca, and Mg). Enhancing the quality of treated wastewater to meet irrigation standards has become a must practice. In order to reach the required quality of treated wastewater membrane technologies are considered to be a critical element.

 (Shanmuganathan et al., 2015) studied the possibility of enhancing the quality of micro-filtered treated sewage effluent using nanofiltration (NF) and reverse osmosis (RO). The study showed that using NF and RO alone could not produce permeate, which meets irrigation standards. Irrigation suitable permeate was produced using an NF-RO hybrid system. Also, it was found that utilizing NF before RO reduced the RO membrane fouling. (Li et al., 2016) studied the performance of advanced treatment of municipal wastewater by nanofiltration. The study evaluated the effect of operating pressure and feed solution pH. The experimental results showed that optimum performance was achieved using a 12 bar 75 pressure,  $pH = 4$  and a flow rate of 8 LPM. Protein-like substances of high molecular weight (MW) are the dominant foulants on the membrane surface. A pilot-scale study conducted by Oron et al. (2006) showed that by using a hybrid ultrafiltration- reverse osmosis (UF-RO) technology, water suitable for irrigation could be produced from secondary treated municipal

wastewater (Oron et al., 2006). The cost of the process was between 0.16 and 0.24 US\$/ $m<sup>3</sup>$  water. Mrayed et al. (2011) applied a hybrid nanofiltration-reverse osmosis (NF-RO) system to produce irrigation water from secondary treated effluent. They used polyacrylic acid (PAA) as a chelating agent. The addition of PAA helped in the formation of covalent bonds among different nutrients in the feed, which improved the rejection rate for those nutrients. (Egea-Corbacho et al., 2019) tested the performance of a pilot-scale nanofiltration membrane for the treatment of secondary treated wastewater effluent. It was found that the product water quality complies with Spanish Royal Decree 1620/2007. This was concluded by considering E Coli, TSS and turbidity, but the authors did not compare the concentration of various elements in the permeate water with allowable limits (i.e. phosphates, nitrates, total dissolved solids, ammonium, sodium and chloride). A study from (Chon et al., 2012) used a hybrid technology comprising of membrane bioreactor and nanofiltration to produce irrigation water from municipal wastewater. It was found that the physicochemical properties and molecular weight cut off were the most critical aspect in the removal of nutrients from the water. (Gu et al., 2019) evaluated the performance of trihybrid anaerobic membrane bioreactor (AnMBR)-reverse osmosis (RO)-ion exchange (IE) process for reclamation of microfiltered municipal wastewater to high-grade clean water. The net energy consumption 96 of the process was  $1.16 \text{ kWh/m}^3$  and product water was found to be suitable for industrial and indirect potable applications. Hafiz et al. (2019) used FO to produce irrigation water from treated sewage effluent (TSE). The feed solution and draw solution for the FO was TSE and 99 an engineered fertilizing solution (0.5 M NaCl & 0.01 M (NH<sub>4</sub>)<sub>2</sub>HPO<sub>4</sub>), respectively. The draw solution was regenerated using RO. The specific power consumption was between 2.18 101 and 2.58 kWh/m<sup>3</sup>. (Liu et al., 2011) evaluated the effectiveness of nanofiltration and reverse osmosis in the treatment of treated textile effluent in terms of salinity reduction and COD rejection. The results showed that nanofiltration exhibited more severe flux decline compared

 to reverse osmosis (RO) because of the higher porosity and membrane fouling of the nanofiltration membrane. RO showed higher total salts rejection compared to NF. (Qi et al., 2020) analyzed pollutants removal efficiency and operating costs of municipal wastewater treatment plants in china. Significant difference in removal efficiencies was observed among various pollutants, with the highest removal efficiency in BOD<sup>5</sup> and lowest removal efficiency in TN. Higher nitrogen removal should be achieved to obtain the desired water 110 quality outcomes.

 So far, previous studies evaluated the performance of various membrane processes for the reclamation of secondary treated sewage effluent. Little information is available for the performance of the nanofiltration and the reverse osmosis membranes in the generation of irrigation water from tertiary treated wastewater. It is recommended to select a single membrane process that can generate high-quality irrigation water from treated sewage effluent at minimal energy requirement. The product water quality must comply with the Food and Agriculture Organization (FAO) standards.

 The objective of this paper is to evaluate the efficiency of nanofiltration and reverse osmosis to treat ultra-filtered tertiary treated sewage effluent (TSE) for the product water to be used in irrigation for food crops. The performance of each technology was evaluated under different applied pressures in terms of water flux, recovery rate, energy consumption and quality of permeate.

2. Materials and setup

2.1 Feedwater

 Ultra-filtered tertiary treated sewage effluent (TSE) was used as feedwater to the nanofiltration and reverse osmosis processes. TSE was collected from a wastewater treatment plant located in Doha, Qatar. The wastewater treatment plant consists of preliminary,

128	secondary and tertiary treatment processes. The tertiary treatment process consists of a
129	multimedia filter followed by ultrafiltration and UV disinfection. The characteristics of the
130	collected, treated sewage effluent are summarized in Table 1. The max limit of the listed
131	parameters was recommended by FAO (Ayres and Westcot, 1985; Lejalem et al., 2018;
132	Parlar et al., 2019). The use of this feed water on food crops was unsuitable because of
133	excessive TDS and high ions-cations. The concentration of heavy metals was below the
134	maximum limit recommended by FAO. The conductivity of samples was measured using
135	OAKTON PCD650 multi-meter. Anions concentration was measured by ion chromatography
136	(Metrohm 850 Professional IC), and cations concentration was measured using plasma
137	emission spectroscopy (iCAP 6500-ICP-OES CID) (Thermo Scientific). Before measuring
138	the concentration of anions and cations, samples with a conductivity value above 1 mS/cm
139	were diluted using deionized water to a conductivity value below 1 mS/cm. This is done to
140	eliminate the interference of high peaks of Na and Cl, which may affect the readings of other
141	elements. The turbidity was measured using a turbidity meter (Hach 2100p). Metal
142	concentration was measured using ICP-MS (Nexion 300D).

143 *Table 1. Characteristics of tertiary treated sewage effluent (feed water)*





144

### 145 2.2 Experimental setup

 A schematic sketch for the bench-scale membrane testing skid is shown in Fig.1. A crossflow CF042D cell made of natural acetal copolymer (Delrin) provided by Sterlitech was used in the nanofiltration and reverse osmosis processes. The cell dimensions are 12.7 x 8.3 x 149 10 cm with active inner dimensions of 4.6 x 9.2 cm and 0.23 cm slot depth. Two tanks were used to store the feed and the permeate water. A M-03S HYDRACELL pump (230V, 50HZ, 3PH, 6.7 LPM) was used to pressurize the feed solution through the membrane. A water chiller (PolyScience Chiller) was used to maintain the feedwater temperature at room 153 temperature ( $25 \pm 2$  °C). A concentrate/back pressure control valve was used to control the water flow through the system and to regulate pressure in the system. Flow meters (Sterlitech Read Panel Mount Flow Meter) were used to measure the flow rate at specific points in the system. A digital balance (Mettler Toledo – ICS 241) was connected to a computer to measure the permeate flux in the system. A specific quantity (3 L) of tertiary treated sewage effluent (TSE) was used as a feed solution in both processes. The applied pressure in the RO







*Figure 1. Schematic diagram of cross flow lab scale membrane test skid*

166 2.3 Error estimation

167 All experiments were performed in triplicates and the reported results are the average 168 of the three experimental trials. The error bars represent the standard deviation of each

169 reading.

## 3. Results and Discussion

3.1 Effect of feed pressure on water flux and recovery rate

172 The water flux  $(J_w)$  in the RO process and the NF process was calculated using Eq.1 (Thabit et al., 2019):

$$
J_w = \left(\frac{V_p}{A_m \times t}\right) \tag{1}
$$

174 Here,  $V_p$  is the volume of the permeate (L),  $A_m$  is the area of the membrane (m<sup>2</sup>), t is the operating time (h). [Figure 2](#page-9-0) (a) and (b) shows the change of water flux with time in the RO and NF process, respectively. It can be seen from Fig.2 (a) and (b) that the water flux decreased with time at all applied pressures. The decrease in the water flux with time is due to the concentration of the feed solution where the reject solution was recycled back into the system and membrane fouling. TSE contains traces of organic matter, which could accumulate on the surface of the membrane and cause fouling (Ortega-Bravo et al., 2016). It can also be seen from Fig.2 (a) that in the RO process the water flux at an applied pressure of 12 – 20 bar was within the same range, but at an applied pressure of 10 bar the water flux was much lower. In the NF process, the water flux at an applied pressure of 12 bar was higher than the water flux at the other applied pressures. This was done to compare the performance of the RO and the NF processes at the different applied pressures; the average water flux was calculated.



<span id="page-9-0"></span>*Figure 2. Water flux using TSE as feed water at different applied pressure in (a) Reverse osmosis (b)Nanofiltration*

187 [Figure 3](#page-11-0) shows the average water flux for RO and NF at different feed pressures. In 188 RO, when the applied pressure was 10 bar the average water flux was  $21.3 \text{ L/m}^2$ .h. When the 189 applied pressure increased to 12 bar the average water flux increased by 69%, to reach a 190 value of 68.1 L/m<sup>2</sup>.h. At a 14 bar applied pressure, the average water flux further increased to 191 reach a value of 71.5 L/m<sup>2</sup>.h. At an applied pressure of 16 bar the average water flux reached

192 a maximum value of  $77.7 \text{ L/m}^2$ .h. As the applied pressure further increased the average water flux decreased. Where at an applied pressure of 18 bar and 20 bar the average water flux was 194 71.6 L/m<sup>2</sup>.h and 67.5 L/m<sup>2</sup>.h, respectively. A different trend was observed in the NF process. Excluding the 10 bar applied pressure experiment, it was found that the average water flux decreased as the applied pressure increased (Fig.3). The maximum average water flux was  $\,$  44.5 L/m<sup>2</sup>.h obtained at an applied pressure of 12 bar. When the applied pressure increased to 198 14 bar the average water flux decreased by almost 37% to reach a value of  $28.1 \text{ L/m}^2$ .h. As the applied pressure further increased the average water flux kept decreasing to reach a 200 minimum value of 20.6 L/m<sup>2</sup>.h at an applied pressure of 20 bar. The water permeability is expected to increase as the feed pressure increases; however, applying excessive pressure may result in excessive accumulation of foulants on the surface of the membrane which may result in lower average water flux (Jiang et al., 2017). The average water flux of RO was higher than NF. In effect, the NF membrane is prone to fouling compared to RO membrane, because the nanofiltration membrane has a rougher, thicker and hydrophobic surface when 206 compared to the reverse osmosis membrane (Xu et al., 2010). The lowest average water flux for RO and NF was obtained at an applied pressure of 10 bar. The water diffusion through the membrane starts to occur when the applied pressure exceeds the natural osmotic pressure of the solution. The osmotic pressure of the feed solution was almost 9 bar; consequently, a feed pressure of 10 bar was not enough to acquire high water flux.



<span id="page-11-0"></span>*Figure 3. Average water flux of RO and NF at different feed pressure*

 $V_F$ 

211 The recovery rate  $(R\%)$  has been calculated using Eq.2 (Singh et al., 2019):

$$
\%R = \left(\frac{V_P}{V}\right) \times 100\%
$$
\n<sup>(2)</sup>

212 Here,  $V_P$  and  $V_F$  are the volume of the permeate and the feed solution, respectively. [Figure 4](#page-12-0) shows the recovery rate for RO and NF at different feed pressures. In the RO process, the highest water recovery was 36% obtained at an applied pressure of 16 bar. The lowest water recovery was 10.2% obtained at an applied pressure of 10 bar. When the applied pressure was 12 bar, the water recovery was 32.7% which is 3.3% lower than the maximum water recovery obtained at 16 bar feed pressure. When the applied pressure was 14 bar, the water recovery was 35.2% which is only 0.8% lower than the maximum water recovery obtained at 16 bar feed pressure. As the feed pressure increased more than 16 bar, the water recovery decreased. At a feed pressure of 18 bar the water recovery was 34.3%, and at a feed pressure of 20 bar the water recovery was 33.8%. As mentioned earlier, applying excessive pressure may result in the accumulation of foulants on the membrane surface, which could result in lower water

 production (Qasim et al., 2019). In nanofiltration, the maximum recovery rate was 21.9% obtained at a feed pressure of 12 bar. The recovery rate decreased dramatically from 21.9% to 14.9% at an applied pressure of 14 bar. As the pressure increased, the recovery rate decreased to reach a value of 10.6% at an applied pressure of 20 bar. The recovery rate of nanofiltration was found to be lower than reverse osmosis. In addition, the effect of pressure is more apparent in nanofiltration. The NF membrane has a higher negative charge compared to the RO membrane, which makes the NF membrane more prone to fouling (Zou et al., 2018). In addition, the nanofiltration membrane has a rougher, thicker and less hydrophilic surface 231 when compared to the reverse osmosis membrane (Chen et al., 2017). The used nanofiltration 232 membrane (NF 90) had a contact angle of  $67.5^{\circ}$  while the used reverse osmosis membrane (BW30LE) had a contact angle 33° (Gryta et al., 2012). This could have also caused the lower recovery rates obtained by NF compared to RO.



<span id="page-12-0"></span>*Figure 4. Recovery rate of RO and NF at different feed pressure*

235 SEM images of unused and used RO and NF membranes at different applied 236 pressures are shown in Fig.5. Fig. 5 (b), (c) and (d) show the used RO membranes at an  applied pressure of 12, 16 and 20 bar, respectively. It can be seen from the SEM images that as the applied pressure increased more accumulation of foulant materials occurred on the surface of the membrane. A similar observation was detected on the nanofiltration membrane where the amount of the accumulated foulants increased on the surface of the membrane as the feed pressure increased (Fig.5 (f), (g) and (h)). From the EDX analysis shown in Table 2, it can be seen that after the use of the RO and NF membranes new elements were detected on the surface of the membranes such as sulfur, iron, and phosphorus, which also indicates the accumulation of foulant materials on the surface of the membrane. The accumulation of these elements on the surface of the membrane negatively affected the water flux. The accumulation of metal ions on the surface of the membrane may further tie the fouling materials on the membrane material resulting in enhanced compactness of the fouling layer (Sun et al., 2019).





249 *Figure 5. SEM images of (a) Clean RO membrane, (b) Tested RO membrane at a feed* 

- 250 *pressure of 12 bar, (c) Tested RO membrane at a feed pressure of 16 bar, (d) Tested RO*
- 251 *membrane at a feed pressure of 20 bar, (e) Clean NF membrane, (f) Tested NF membrane at*
- 252 *a feed pressure of 12 bar, (g) Tested NF membrane at a feed pressure of 16 bar, (h) Tested*
- 253 *NF membrane at a feed pressure of 20 bar.*
- 254

255 *Table 2. EDX analysis of elements wt% on the surface of a clean and tested RO and NF*  256 *membranes*

<b>Membrane</b>	Weight %					
	$ {\bf C} $	[O]	[S]	[Fe]	$\mathbf{[P]}$	[Other]
<b>RO-clean</b>	87.39	9.53	3.08		0	0
<b>RO-tested</b>	70.64	19.45	3.32	1.79	0.66	4.14
NF-clean	89.78	6.66	3.56		0	$\theta$
NF-tested	73.96	16.32	3.94	1.91	0.48	3.39

<sup>257</sup>

258 3.3 Energy consumption

259 The specific energy consumption  $(E_s)$  of the RO and the NF processes has been

260 calculated using Eq.3 (Shrivastava and Stevens, 2018):

$$
E_s = \left(\frac{P}{n \times \%R}\right) \tag{3}
$$

 Here, P is the applied feed pressure (bar), *n* is the pump efficiency, %R is the recovery rate. The specific energy consumption depends on the applied pressure and the recovery rate; therefore, the lowest energy consumption will be obtained at a high recovery rate using low applied pressure. [Figure 6](#page-16-0) shows the specific energy consumption of the RO process and the NF process at different feed pressures. In the NF process, the specific energy consumption 266 increased from 0.68 kWh/m<sup>3</sup> to 2.35 kWh/m<sup>3</sup> at a feed pressure of 12 bar and 20 bar, respectively. As shown in Eq.3, the specific energy is a function of applied pressure and recovery rate. At a low feed pressure of 10 bar the specific energy consumption was 1.33  $\text{kWh/m}^3$ . The high specific energy consumption at such low feed pressure is due to the low recovery rate obtained at 10 bar feed pressure (Fig.4). The maximum energy consumption 271 was 2.35 kWh/m<sup>3,</sup> which was obtained at a feed pressure of 20 bar. The high specific energy consumption at such high applied pressure is due to the low recovery (Fig.4). In the RO process, the same trend was observed where the specific energy consumption increased from 274 0.46 kWh/m<sup>3</sup> to 0.73 kWh/m<sup>3</sup> at a feed pressure of 12 bar and 20 bar, respectively. At a low 275 feed pressure of 10 bar the specific energy consumption was  $1.22 \text{ kWh/m}^3$ , which is due to the low recovery rate obtained at such low applied pressure as shown in Fig.4. It was found that the NF process at an applied pressure of 12 bar gave the highest water recovery rate and the lowest energy consumption. For the RO process, the lowest energy consumption was found at an applied pressure of 12 bar while the highest water recovery rate was at an applied pressure of 16 bar. The difference in the water recovery rate between the 12 and 16 bar applied pressure in the RO process was only 3.3%. In comparison, the energy consumption was 18% higher at an applied pressure of 16 bar when compared to 12 bar applied pressure. The quality of the produced permeate should be analyzed to investigate which process and which running conditions to be utilized.



<span id="page-16-0"></span>*Figure 6. Specific energy consumption of RO and NF at different feed pressure.*

285

286 3.4 Product water quality

 Characteristics of the produced permeate were measured for the RO and the NF processes. The quality of the produced permeate was compared with the Food and Agriculture Organization (FAO) standards (Ayres and Westcot, 1985). As shown in Fig.7 (a) the concentration of the different measured elements in the produced permeate from the RO process at applied pressures between 10 and 18 bar was within the FAO standards. At an applied pressure of 20 bar, multiple parameters such as TDS, conductivity, chloride, sodium and calcium concentration exceeded the allowable limits. In the NF process (Fig.7 (b)), under all applied pressures, the permeate quality did not comply with FAO standards. It was noticed that the elements that did not comply with the standards were chloride, sodium and calcium. The high concentration of these elements in return affected the TDS concentration. For example, at 12 bar applied pressure where the highest water recovery rate was attained the TDS concentration was almost 21% higher than the allowable limit, and the chloride, sodium

- 299 and calcium concentrations were 84%, 61% and 13% higher than the allowable limit,
- 300 respectively. It could be noticed that the NF membrane had a low rejection rate for
- 301 monovalent ions, where the rejection rate for chloride and sodium was only 27% and 12%,
- 302 respectively.
- 303



Fluoride **Potassium TDS Bromide Nitrate Magnesium** Calcium ■ Sulfate Sodium Chloride



*Figure 7. Characteristics of the permeate produced from treated sewage effluent (TSE) using (a) Reverse osmosis (b)Nanofiltration at different applied pressures.*

 It can be inferred that using a single-stage NF is not possible due to low water quality, and it is recommended to use RO at applied pressure between 10 and 18 bar. Selecting the most suitable applied pressure to operate the RO process depends on water flux, recovery rate, energy consumption and product water quality. It was found in the previous sections that the lowest energy consumption of RO was obtained at an applied pressure of 12 and 14 bar. After considering the water quality, it is recommended to use an applied pressure of 14 bar due to higher water quality.

# 4. Conclusions

 In this paper, a comparative study was done on the reclamation of tertiary treated sewage effluent (TSE) by nanofiltration and reverse osmosis for water reuse in irrigation. The results reported show that reverse osmosis (RO) is capable of reclaiming tertiary treated

 sewage effluent (TSE) to be used as irrigation water for food crops. NF is not suitable for the reclamation of wastewater because of the low rejection of monovalent ions. In nano-filtered TSE, the concentration of Na and Cl exceeded the maximum allowable limits recommended by FAO. RO is suitable for the reclamation of wastewater due to high water quality and water flux. In the RO process, the highest recovery rate was 36% achieved at an applied pressure of 320 16 bar. The specific energy consumption at this applied pressure was  $0.56$  kWh/m<sup>3</sup>. At 14 bar applied pressure, the recovery rate was only 2% lower than that at an applied pressure of 16 bar, while the specific energy consumption was almost 11% lower. At 12 bar applied pressure, the specific energy consumption was 8% higher than the specific energy at an applied pressure of 14 bar while the recovery rate was 7% higher. It is recommended to use RO process at 14 bar applied pressure for the reclamation of TSE. This is due to the high recovery rate, low energy consumption and high water quality compared to other available technologies. The product water could meet the quality requirements of the Food and Agriculture Organization (FAO). The results from this study might lead to a paradigm shift in the reclamation of tertiary treated wastewater to be used for the irrigation of food crops.

#### **Declaration of competing interest**

 The authors declare that they have no known competing financial interests or personal relationships that could have appeared to influence the work reported in this paper.

### **CRediT authorship contribution statement**

 **MhdAmmar Hafiz**: Conceptualization, Data collection, Data analysis, Writing manuscript. **Alaa H. Hawari**: Conceptualization, Methodology, Review & editing. **Radwan Alfahel**: Data collection, Data analysis. **Ali altaee**: Conceptualization, Methodology, Review & editing.

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