ORIGINAL ARTICLE





Monitoring metabolism of synthetic cannabinoid 4F-MDMB-BINACA via high-resolution mass spectrometry assessed in cultured hepatoma cell line, fungus, liver microsomes and confirmed using urine samples

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Abstract

Purpose A *tert*-leucinate derivative synthetic cannabinoid, methyl (2S)-2-([1-(4-fluorobutyl)-1*H*-indazole-3-carbonyl] amino)-3,3-dimethylbutanoate (4F-MDMB-BINACA, 4F-MDMB-BUTINACA or 4F-ADB) is known to adversely impact health. This study aimed to evaluate the suitability of three different modes of monitoring metabolism: HepG2 liver cells, fungus *Cunninghamella elegans* (*C. elegans*) and pooled human liver microsomes (HLM) for comparison with human invivo metabolism in identifying suitable urinary marker(s) for 4F-MDMB-BINACA intake.

Methods Tentative structure elucidation of in-vitro metabolites was performed on HepG2, *C. elegans* and HLM using liquid chromatography–tandem mass spectrometry and high-resolution mass spectrometry analysis. In-vivo metabolites obtained from twenty authentic human urine samples were analysed using liquid chromatography–Orbitrap mass spectrometry.

Results Incubation with HepG2, *C. elegans* and HLM yielded nine, twenty-three and seventeen metabolites of 4F-MDMB-BINACA, respectively, formed via ester hydrolysis, hydroxylation, carboxylation, dehydrogenation, oxidative defluorination, carbonylation or reaction combinations. Phase II metabolites of glucosidation and sulfation were also exclusively identified using *C. elegans* model. Eight in-vivo metabolites tentatively identified were mainly products of ester hydrolysis with or without additional dehydrogenation, *N*-dealkylation, monohydroxylation and oxidative defluorination with further oxidation to butanoic acid. Metabolites with intact terminal methyl ester moiety, i.e., oxidative defluorination with further oxidation to butanoic acid, were also tentatively identified.

Conclusions The in-vitro models presented proved useful in the exhaustive metabolism studies. Despite limitations, HepG2 identified the major 4F-MDMB-BINACA ester hydrolysis metabolite, and *C. elegans* demonstrated the capacity to produce a wide variety of metabolites. Both *C. elegans* and HLM produced all the in-vivo metabolites. Ester hydrolysis and ester hydrolysis plus dehydrogenation 4F-MDMB-BINACA metabolites were recommended as urinary markers for 4F-MDMB-BINACA intake.

 $\textbf{Keywords} \ \ \, 4F\text{-}MDMB\text{-}BINACA \ metabolism} \cdot 4F\text{-}MDMB\text{-}BUTINACA \ \, \\ (4F\text{-}ADB) \cdot HepG2 \cdot \textit{Cunninghamella elegans} \cdot Human \ \, liver \ microsomes \cdot Urinary \ marker \ \, of \ \, \\ 4F\text{-}MDMB\text{-}BINACA \ \, intake$

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Introduction

Synthetic cannabinoids (SCBs) are agonists at cannabinoid receptor type 1 (CB₁) and type 2 (CB₂), where they elicit their main effects. Due to their similar physiological effects to the principal psychoactive component of cannabis, Δ^9 -tetrahydrocannabinol (THC), SCBs are gaining popularity and are often abused as recreational drugs. However, most of the SCBs are full agonists at CB1 and CB₂ receptors, having a higher risk of undesirable side effects when compared to THC which is a partial agonist [1]. Methyl (2S)-2-([1-(4-fluorobutyl)-1H-indazole-3-carbonyl]amino)-3,3-dimethylbutanoate (4F-MDMB-BINACA, 4F-MDMB-BUTINACA or 4F-ADB), found in numerous SCB product seizures, has been reported by various law enforcement since 2018 [2]. The tert-leucine derived indazole-3-carboxamide analog 4F-MDMB-BINACA postulated to be a potent agonist of the CB₁ receptor based upon its structural similarity to methyl-2-([1-(5-fluoropentyl)-1*H*-indazole-3-carbonyl]amino)-3,3-dimethylbutanoate (5F-MDMB-PINACA) [3, 4], has significant abuse potential and causes central nervous system depression. Death involving these drugs have been reported [5-9], and this raises public health and social concerns. Moreover, 4F-MDMB-BINACA has been listed in Schedule II of the Convention on Psychotropic Substances of 1971 recently.

The proliferation of SCBs has become a global challenge as new compounds are rapidly introduced into the illegal drug market to evade existing drug laws. Identifying SCB intake and its correlating specific adverse effects require rapid elucidation of these SCBs. Since most SCBs are found extensively in metabolized forms in urine, the identification of metabolites is of vital importance for forensic and clinical toxicologists. Furthermore, some hydroxylated urinary metabolites have been found to be active and more toxic than the parent SCBs themselves, emphasizing the importance of metabolism study in identifying potential toxic metabolites [10, 11].

Traditional in-vivo metabolism studies to generate human metabolites of drugs relied heavily on the use of whole animal model systems, which are expensive, limited by drug administration amount, influenced by species variation and faced by many ethical issues. In-vitro metabolism studies are generally used to complement these data using perfused organs, tissue or cell cultures and microsomal preparations amongst which pooled human liver microsomes (HLM) have been frequently used to elucidate metabolism of SCBs [12–16]. The hepatic cell line HepG2 is often used as an initial screen as it is known to produce high reproducibility results with relatively stable enzyme concentration, although they are limited by the low-level

expression of several metabolizing enzymes, including the cytochrome P450 (CYP) class of proteins [17, 18]. The metabolic competence of HepG2 was described to be sufficient for in-vitro studies after an incubation time of at least 72 h [19]. An alternative to this was the use of a microbial model with phase I and II metabolite-generating capabilities such as the zygomycete fungus, Cunninghamella elegans (C. elegans). C. elegans has the advantage of more closely mimicking mammalian metabolism [20] and has been used as a complementary model in metabolism studies of new psychoactive substances, in particular, the SCBs [21–26]. This class of fungi has an enzymatic activity to facilitate phase I and II metabolism including hydroxylation, carboxylation, dihydrodiol formation, oxidative defluorination, N-dealkylation, glucuronidation, glucosidation, and sulfation [23, 27]. The advantages offered by the microbial transformation over other approaches include the low setting up cost, reliability and reproducibility as well as possible economical scale-up affording both phase I and II metabolites in higher yield. This scaled production is useful for further characterization of metabolites that are required to be distinguished from its isomeric species and in the identification of potentially toxic metabolites [28].

Although there were reports on the metabolism of 4F-MDMB-BINACA using in-vivo and various in-vitro models, studies were either conducted using small in-vivo sample size such as 1 to 4 samples [5, 29] or in closed environments such as forensic psychiatric wards and prisons [30]. In our study, twenty in-vivo urine samples screened positive for 4F-MDMB-BINACA intake were employed and the findings were compared to those obtained via HepG2, *C. elegans*, and HLM in-vitro models. The aim of this study was to assess the suitability of these in-vitro models in the in-vitro – in-vivo extrapolations and to suggest suitable urinary marker(s) for 4F-MDMB-BINACA intake.

Materials and methods

Chemicals and reagents

4F-MDMB-BINACA was provided by Health Sciences Authority (Singapore). 4F-MDMB-BINACA 3,3-dimethylbutanoic acid metabolite was purchased from Cayman Chemical (Ann Arbor, MI, USA). In-vitro metabolism study: one hundred fifty (mixed gender) - donor HLM pool, NADPH system solution A and NADPH system solution B, T75 tissue culture flasks, and 96-well cell culture plates were purchased from Corning (Corning, NY, USA); *C. elegans* ATCC 10028b from Cryosite Ltd. (South Granville, NSW, Australia); HepG2 cell line from CellBank (Sydney, NSW, Australia). Minimum Essential Medium - Eagle with



Earle's BSS (MEM Eagle with EBSS) from Lonza (Basel, Switzerland); liquid chromatography – mass spectrometry (LC-MS) grade acetonitrile (ACN) from Honeywell (Muskegon, MI, USA); reagent grade dichloromethane (CH₂Cl₂) and sodium chloride (NaCl) from Chem-Supply (Gilman, SA, Australia); LC-MS grade formic acid from Sigma-Aldrich (St. Louis, MO, USA); glycerol, potassium dihydrogen phosphate (KH₂PO₄) and dipotassium hydrogen phosphate (K₂HPO₄) from Ajax Chemicals (Auburn, NSW, Australia); potato dextrose agar, glucose, peptone, and yeast extract from Oxoid Australia (Adelaide, SA, Australia). In-vivo metabolism study: chemicals for preparing reagents and buffers were obtained from Fluka, Sigma-Aldrich (St. Louis, MO, USA). Optima LC-MS grade ACN from ThermoFisher Scientific (Fair Lawn, NJ, USA); purified water was used, and all solvents used were of LC-MS grade or better; β-glucuronidase from abalone (Haliotis rufescens) from KURA Biotec (Rancho Dominguez, CA, USA); Clean Screen FASt® filter tubes (3 mL packed with 200 mg solid-phase sorbent) from UCT, Inc (Bristol, PA, USA).

Monitoring in-vitro metabolism

HepG2 liver cells

Cryofrozen HepG2 cells were thawed and subcultured (up to passage 7) in a T75 tissue culture flask containing MEM Eagle with EBSS media supplemented with 10% heat-inactivated fetal bovine serum (HI-FBS), L-glutamine (2 mM), penicillin:streptomycin (100 U/mL:100 μg/mL) and non-essential amino acid solution (NEAA, 1x). The cells were cultured at 37 °C in a humidified atmosphere incubator containing 95% air and 5% CO₂, until ~ 80% confluent. The media was changed every 2 days, and when required, cells were harvested by subculture with media containing trypsin/EDTA (0.12% trypsin/ 0.02% EDTA, w/v). The following method was adopted [31] with slight modifications: cells were seeded at a density of 9600 cells/ well onto 96-well plates overnight. Cells were treated with 120 μM of 4F-MDMB-BINACA media solution (100 μL) in triplicates for 72 h. The supernatant (60 μ L) was transferred into a 0.5 mL capped tube containing ice-cold ACN (60 μL). The mixture was vortex-mixed and cooled for another 15 min at −30 °C before centrifuging at 10,000 g for 5 min. The filtered sample (0.22 μm, 100 μL) was diluted with ACN (500 µL) and 10 µL was subjected to liquid chromatography – quadrupole time-of-flight mass spectrometry (LC – QTOF-MS). A 2 μL aliquot was also subjected to liquid chromatography-triple quadrupolemass spectrometry (LC - QQQ-MS). A control without HepG2 and/or without 4F-MDMB-BINACA were also treated in the identical fashion.

Fungus C. elegans

C. elegans was cultured on potato dextrose agar plates at 27 °C for 5 days. The mycelia of the fungus were mixed in sterile physiological saline solution (1 plate of mycelia/10 mL). Growth medium consisting of NaCl/KH₂PO₄/ yeast extract/peptone/glucose (1:1:1:1:2, v/v/v/v), water and glycerol was prepared as previously described [32] and autoclaved (121 °C, 15 min) before use. A 1.5-mL volume of the homogenate was inoculated into 100 mL of growth medium in a conical flask. The culture was incubated for 48 h at 26 °C and 180 rpm on a Multitron rotary shaker (Infors AG, Bottmingen, Switzerland). 4F-MDMB-BIN-ACA (1 mg in 0.5 mL ACN) was added to the conical flask in triplicate and incubated for another 72 h. The solution was filtered in a Buchner funnel, extracted with CH₂Cl₂ (3×50 mL). The combined extracts were evaporated to dryness using a rotary evaporator followed by a vacuum pump. The sample was reconstituted in 2 mL ACN and filtered using a 0.22 µm filter and was further diluted in ACN tenfold before injected (5 µL) into the LC – QTOF-MS instrument. An aliquot of the filtrate (100 µL) was diluted in 900 μL water/ACN (90:10, v/v) and 2 μL was also injected into the LC - QQQ-MS instrument. Control samples in the absence of C. elegans and/or 4F-MDMB-BINACA were also treated in the identical fashion.

Pooled human liver microsomes

The incubation mixture using 4F-MDMB-BINACA solution in ACN/phosphate buffer (25 µL, final concentration 1.7 µM in 0.03% ACN), phosphate buffer (0.1 M, pH 7.4, 855 μL), NADPH-A (50 µL) and NADPH-B (20 µL) was mixed in a 2 mL capped tube, to which HLM (50 μ L = 1 mg protein) was added. The mixture was vortexed mixed and incubated in triplicate at 37 °C in a shaking water bath (100 rpm) for 1 h. The reaction was quenched by adding ice-cold ACN (1 mL) to the mixture, vortex mixed and then centrifuged at 13,000 g for 10 min. The sample was filtered (0.22 µm) and 5 µL was injected into the LC - QTOF-MS instrument. A 100-µL volume of the filtrate was diluted in 900 μL water/ACN (90:10, v/v), and 2 μL was also injected into the LC – QQQ-MS instrument. Controls samples without HLM and/or without 4F-MDMB-BINACA were also incubated and subsequently worked up under the identical conditions.



LC - QTOF-MS

Chromatographic separation was performed on an Agilent 1290 LC system with a Poroshell 120 EC-C18 analytical column (2.7 μm, 75×2.1 mm; Agilent Technologies, Santa Clara, CA, USA). The mobile phase consisted of 0.1% formic acid in water (A) and 0.1% formic acid in acetonitrile (B). The gradient elution was as follows: 10% B until 2 min, increased to 40% B until 4 min, ramped to 60% B until 15 min, and further increased to 80% B at 18 min and ramped down to 10% B until 19 min and held until 21 min. The flow rate was 0.3 mL/min and the column temperature was kept at 35 °C. High-resolution QTOF-MS data were acquired on an Agilent 6510 Accurate Mass QTOF mass spectrometer (Agilent Technologies) equipped with dual electrospray ionization (ESI) source operated in both positive and negative ion modes, to determine accurate masses of the metabolites. The parameters were as follows: scanning mass range, m/z 100-700 (MS), m/z 100-600 (MS/ MS); capillary voltage, 3500 V; nebulizer pressure, 40 psi; gas temperature, 310 °C; drying gas flow, 13 L/ min; fragmentor voltage, 160 V; collision energy (CE) for product ion scan analysis, 10, 20 and 40 eV; skimmer voltage, 65 V. Mass calibration was performed with the mixture provided by the manufacturer. Real-time mass calibration was enabled using the following reference masses: m/z 118.086255 and 322.048121 (positive ion mode); 112.985587 and 301.998139 (negative ion mode). Extracted ion chromatograms and product ion spectra were analysed using Agilent MassHunter Workstation Software Qualitative Analysis (version B.06.00). The criteria for metabolites elucidation were as follows: mass error of the precursor ion \leq 5.00 ppm, mass error of product ions ≤ 20.00 ppm, consistent fragmentation pattern with the proposed structure, reasonable retention time, and absence of the specific peak in controls.

LC - QQQ-MS

Chromatographic equipment and conditions were the same as described above. MS was run in multiple reaction monitoring (MRM) mode on an Agilent 6490 Triple Quadrupole mass spectrometer with ESI source in positive ion mode (Agilent Technologies). An optimised CE 8–40 eV was applied for each individual MRM transition. The source parameters were as follows: the capillary and nozzle voltages, 3500 and 1500 V, respectively; the sheath gas temperature and flow set to 400 °C and 12 L/min, respectively; nebulizer pressure, 40 psi; gas temperature, 290 °C; drying gas flow, 12 L/min; fragmentor voltage, 380 V. Mass calibration was routinely performed via check tune before analysis. Extracted ion chromatograms and MRM mass chromatograms were analysed using Agilent MassHunter Workstation Software Qualitative Analysis (version B.06.00). Percentage

(%) peak area ratio was obtained by dividing the peak area of the detected metabolite by the peak area of the most abundant metabolite in the same sample. Major metabolites were defined as those with greater than 10% average peak area ratios.

In-vivo metabolism

Authentic human urine samples

Antemortem urine samples from routine toxicology cases (n = 20) confirmed with the 4F-MDMB-BINACA ester hydrolysis metabolite (4F-MDMB-BINACA 3,3-dimethylbutanoic acid metabolite) were processed and analysed for in-vivo metabolites in collaboration with the in-vitro metabolism findings. Urine sample (500 µL) was incubated with 250 μ L of pH 5 acetate buffer (1 M) and 50 μ L β-glucuronidase (10,000 Fisherman units/mL) for 1 h at 60 °C. After incubation, the mixture was cooled at room temperature, and 150 µL of purified water was added. The mixture was vortex-mixed and 500 µL of this mixture and 500 µL of methanol were loaded onto the Clean Screen FASt® tube. About 10-20 bar pressure was applied on the tube using a positive pressure manifold to allow all the samples to flow through the tube. The eluent was collected in a 2 mL sample vial for analysis.

LC - Orbitrap-MS

Chromatographic separation was achieved on a Kinetex XB-C18 column (2.6 µm, 100 × 2.1 mm; Phenomenex, Torrance, CA, USA) with an identically packed defender guard cartridge (2.6 μ m, 10×2.1 mm). The mobile phase consisted of 10 mM ammonium formate, pH 3 (A) and 0.1% formic acid in ACN (B). The gradient elution was as follows: 5% B until 1 min, increased to 95% B until 10 min, ramped to 99% B at 10.1 min, held until 11.5 min, ramped down to 5% B at 11.6 min and held until 13.0 min. The flow rate was 0.5 mL/ min for the whole run except from 10.1 to 11.5 min which was set at 0.8 mL/min. The identification of the metabolites was performed using liquid chromatography coupled with Thermo Scientific Q-Exactive hybrid quadrupole - Orbitrap mass spectrometer (Thermo Scientific, Fremont, CA, USA) equipped with heated electrospray ionization source (HESI-II) operated in positive ionization with full MS and datadependent MS² acquisition mode. The parameters were as follows: scanning mass range, m/z 120–1000 (MS); sheath gas, 40 arbitrary units (AU); auxiliary gas, 10 AU; sweep gas, 1 AU; spray voltage, 3.5 kV; capillary temperature, 270 °C; heater temperature, 425 °C. The in-vitro metabolites elucidated from the cultured HepG2 cells, C. elegans and HLM were added to the compound database. Data obtained from the twenty urine samples were retrospectively analysed



and processed using TraceFinder software based on the identification criteria of mass errors less than ± 5 ppm for full MS peaks and MS/MS peaks from the theoretical mass and matching of MS/MS spectra. Percentage (%) peak area ratios of the metabolites were similarly calculated as the in-vitro metabolism study.

Results

A total of 25 different metabolites (B1–B25) was tentatively identified and listed in Table 1 in the order of retention time (RT) with postulated biotransformation. 4F-MDMB-BINACA was eluted at 12.0 min, whilst the metabolites eluted between 5.3 and 9.6 min. Metabolites were identified according to their precursor ions, product ions, and fragmentation patterns (Fig. 1). The proposed biotransformation pathways of 4F-MDMB-BINACA are presented in Fig. 2.

In-vitro metabolism

Twenty-three, seventeen and nine metabolites were tentatively identified after incubation of 4F-MDMB-BINACA with *C. elegans*, HLM and HepG2 cells, respectively.

Ester hydrolysis of 4F-MDMB-BINACA generated 4F-MDMB-BINACA ester hydrolysis metabolite, 4F-MDMB-BINACA 3,3-dimethylbutanoic acid (B22) with further dehydrogenation (B20), monohydroxylation (B8, B9, B13) with further dehydrogenation (B5/B6, B14), oxidative defluorination (B2) with subsequent oxidation to butanoic acid (B4) and *N*-dealkylation (B1). Other metabolites retained the terminal methyl ester moiety and their biotransformation tentatively identified included: monohydroxylation (B19/B21, B23/B25) with further carbonylation (B24), dihydroxylation (B10, B12/B15), oxidative defluorination (B18) and further oxidation to butanoic acid (B17) and *N*-dealkylation (B16). Phase II glucosidation (B7) and

Table 1 In-vitro metabolites of 4F-MDMB-BINACA tentatively identified using *Cunninghamella. elegans* (*C. elegans*)(F), human liver microsomes (HLM)(M) and HepG2 (H) models

ID	RT (min)	Biotransformation	Formula	Exact mass $[M+H]^+$	U	F	M	Н
B1	5.30	Ester hydrolysis, N-dealkylation	C ₁₄ H ₁₇ N ₃ O ₃	276.1343	✓	✓	✓	✓
B2	5.40	Ester hydrolysis, oxidative defluorination	$C_{18}H_{25}N_3O_4$	348.1918	✓	✓	✓	✓
В3	5.40	Ester hydrolysis, oxidative defluorination to butanoic acid, dehydrogenation	$C_{18}H_{21}N_3O_5$	360.1550		✓		
B4	5.45	Ester hydrolysis, oxidative defluorination to butanoic acid	$C_{18}H_{23}N_3O_5$	362.1710	✓	✓	✓	
B5	5.50	Ester hydrolysis, monohydroxylation (butyl), dehydrogenation	$C_{18}H_{22}FN_3O_4$	364.1667		✓		
B6	5.70	Ester hydrolysis, monohydroxylation (butyl), dehydrogenation	$C_{18}H_{22}FN_3O_4$	364.1667		✓	✓	
B7	5.80	Monohydroxylation (indazole), glucosidation	$C_{25}H_{36}FN_3O_9$	542.2508		✓		
B 8	5.85	Ester hydrolysis, monohydroxylation (tert-leucine)	$\mathrm{C_{18}H_{24}FN_3O_4}$	366.1824	✓	✓	✓	✓
B9	6.10	Ester hydrolysis, monohydroxylation (butyl)	$\mathrm{C_{18}H_{24}FN_3O_4}$	366.1824		✓	✓	
B10	6.10	Dihydroxylation (butyl)	$C_{19}H_{26}FN_3O_5$	396.1929		✓		
B11	6.40	Monohydroxylation (indazole), sulfation	$C_{19}H_{26}FSN_3O_7$	460.1548		✓		
B12	6.60	Dihydroxylation (indazole, butyl)	$C_{19}H_{26}FN_3O_5$	396.1929		✓		
B13	6.70	Ester hydrolysis, monohydroxylation (indazole)	$C_{18}H_{24}FN_3O_4$	366.1824		✓	✓	✓
B14	6.75	Ester hydrolysis, monohydroxylation (<i>tert</i> -leucine), dehydrogenation	$C_{18}H_{22}FN_3O_4$	364.1667		✓	✓	
B15	6.90	Dihydroxylation (indazole, butyl)	$C_{19}H_{26}FN_3O_5$	396.1929		✓		
B16	7.10	N-Dealkylation	$C_{15}H_{19}N_3O_3$	290.1499		✓	✓	✓
B17	7.30	Oxidative defluorination to butanoic acid	$C_{19}H_{25}N_3O_5$	376.1867	✓	✓	✓	✓
B18	7.35	Oxidative defluorination	$C_{19}H_{27}N_3O_4$	362.2074	✓	✓	✓	✓
B19	7.70	Monohydroxylation (butyl)	$C_{19}H_{26}FN_3O_4$	380.1980		✓	✓	
B20	7.90	Ester hydrolysis, dehydrogenation	$C_{18}H_{22}FN_3O_3$	348.1718	✓	✓	✓	✓
B21	8.10	Monohydroxylation (butyl)	$C_{19}H_{26}FN_3O_4$	380.1980		✓	✓	
B22	8.30	Ester hydrolysis	$C_{18}H_{24}FN_3O_3$	350.1874	✓	✓	✓	✓
B23	8.50	Monohydroxylation (indazole)	$C_{19}H_{26}FN_3O_4$	380.1980			✓	
B24	9.00	Carbonylation	$C_{19}H_{24}FN_3O_4$	378.1824		✓		
B25	9.60	Monohydroxylation (indazole)	$C_{19}H_{26}FN_3O_4$	380.1980			✓	
В0	12.00	4F-MDMB-BINACA	$C_{19}H_{26}FN_3O_3$	364.2031		✓	✓	✓

Mass error of the precursor ion \leq 5.00 ppm and mass error of product ions \leq 20.00 ppm *ID* identity, *RT* retention time, *U* in-vivo urine sample



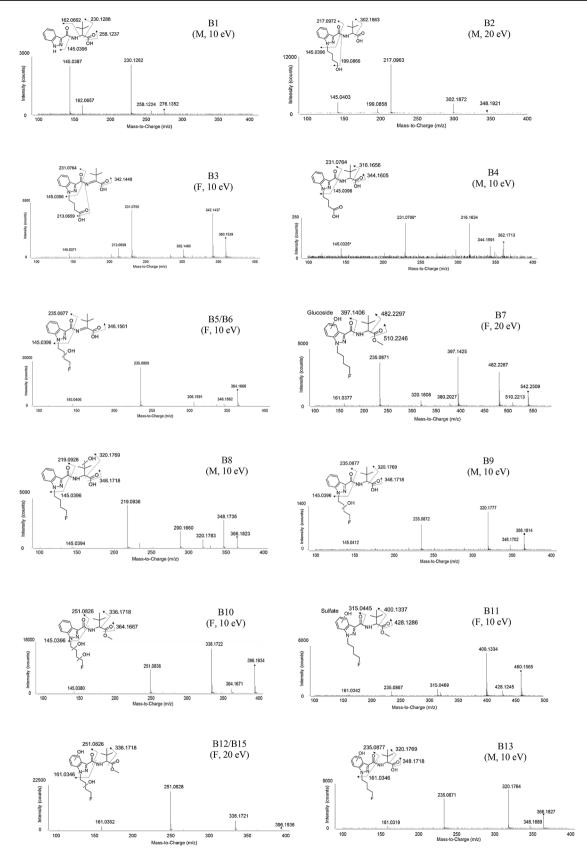


Fig. 1 Product ion spectra and tentative structures with the suggested fragmentation patterns from the three in-vitro models: *Cunninghamella elegans* (*C. elegans*)(F), human liver microsomes (HLM)(M)

and HepG2 (H). The brackets indicate product ion spectra obtained from one of the in-vitro model with collision energy 10 or 20 eV. The exact position of the hydroxylation was not determined



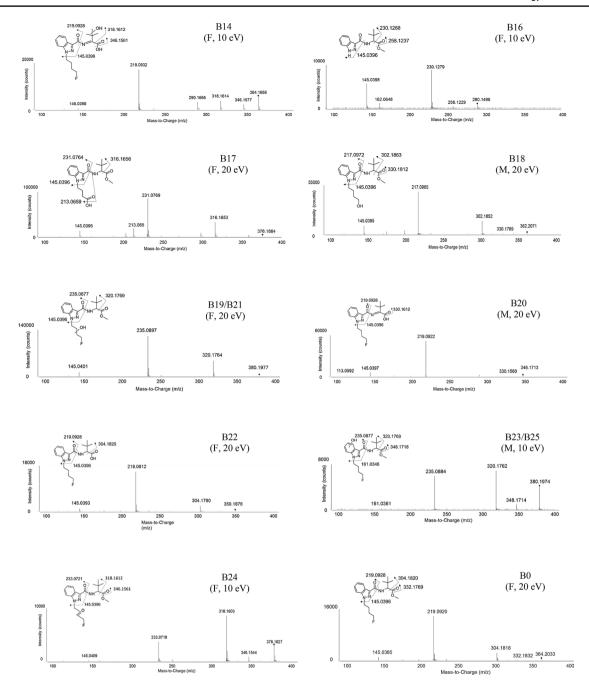


Fig. 1 (continued)

sulfation (B11) metabolites were only detected in the *C. elegans* model. The tentative structure elucidation of these various metabolites from the different models examined was constructed and shown below.

Phase I biotransformation

Ester hydrolysis, dehydrogenation, hydroxylation, oxidation and dealkylation

All the below explanations for transformations into metabolites are based on the data shown in Fig. 1 and on probable metabolic pathways shown in Fig. 2. 4F-MDMB-BINACA was hydrolysed via ester hydrolysis forming the 4F-MDMB-BINACA ester hydrolysis metabolite (B22). The product ion



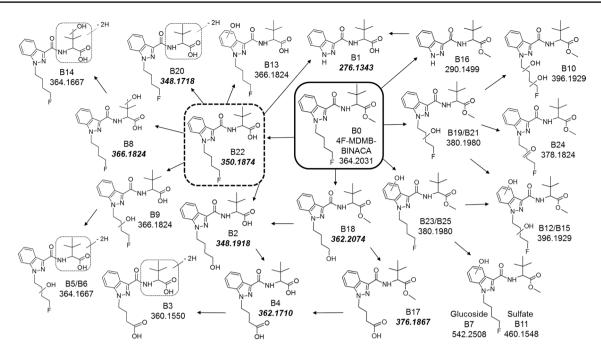


Fig. 2 Proposed metabolic pathways of 4F-MDMB-BINACA. In-vivo metabolites were in italics and bold. The exact position of the hydroxyl moiety of the metabolites was not determined

spectrum was very similar to 4F-MDMB-BINACA (m/z 304, 219, 145). The precursor ion, m/z 350 showed a loss of 14 Da explaining the hydrolysis of methyl ester from 4F-MDMB-BINACA. B20 showed a similar product ion fragmentation pattern to that of B22 (m/z, 219, 145) and additional loss of 2 Da from m/z 350, indicating ester hydrolysis and further dehydrogenation at the *tert*-leucine moiety.

Metabolites identified at m/z 366 (B8, B9, B13), which was 16 Da higher than the 4F-MDMB-BINACA ester hydrolysis metabolite (B22), confirmed monohydroxylation upon ester hydrolysis. The presence of the product ion m/z320, likely formed from a loss of carbon dioxide, indicated monohydroxylation at the tert-leucine in B8 (m/z 219), butyl side chain in B9 (m/z 145) and indazole moiety in B13 (m/z161). The precursor ion, *m/z* 364 (B14, B5/B6) had a loss of 2 Da from m/z 366 indicated further dehydrogenation of the ester hydrolysis plus monohydroxylated metabolites. Dehydrogenation occurred at the tert-leucine moieties for both the monohydroxylated metabolites (B5/B6, B14), indicated by the presence of product ions m/z 235/346 and m/z 219/346, respectively. Similarly, precursor ion identified at m/z 380 (B19/B21, B23/B25) was 16 Da higher than the 4F-MDMB-BINACA, indicating monohydroxylation at the butyl side chain (B19/B21) and indazole (B23/B25) moieties with product ions m/z 145 and 161, respectively. The metabolite (B24) identified at m/z 378 had very similar product ion spectra with B19/B21 except it was 2 Da lower (equating to the loss of 2 hydrogen atoms) while retaining the same product ion m/z 145, indicating carbonylation at the butyl side chain moiety.

The precursor ion m/z 396 (B10, B12/B15) was 32 Da higher than the parent drug, 4F-MDMB-BINACA, suggesting the addition of two hydroxy groups. The product ion m/z 336 (loss of methyl ester moiety) further confirmed the presence of dihydroxylated metabolites. The presence of product ion m/z 251 at B10 and B12/B15 was explained by noting that hydroxylation occurred at the indazole/butyl side chain moieties and not at the *tert*-leucine moiety. B10 showed product ion m/z 145 indicating intact indazole moiety and suggested that dihydroxylation occurred at the butyl side chain moiety. The presence of product ion m/z 161 in B12/B15, indicated that hydroxylation occurred at the indazole moiety, whilst the second hydroxylation site must be assigned at the butyl side chain moiety.

The precursor ion m/z 348 and product ion detected at m/z 217 (B2) identified was 2 Da less than the 4F-MDMB-BINACA ester hydrolysis metabolite (B22), indicating oxidative defluorination (loss of fluorine with addition of hydroxy group). Product ions detected at m/z 302, 217, and 145 (B2) confirmed that tert-leucine and indazole moieties remained unchanged, leading to the structure elucidation of a hydroxy-functional group at the 4-position of the butyl side chain by oxidative defluorination. Oxidation of 4'-hydroxybutyl moiety in B2 led to formation of 4'-carboxybutyl metabolite (B4) having a precursor ion of m/z 362, which was 14 Da higher than the m/z for B2 (loss of two hydrogen atoms with addition of a carbonyl group).



The precursor ion m/z 360 and product ion detected at m/z 342 (B3) identified was 2 Da less than precursor ion at m/z 362 (B4) indicated further dehydrogenation at the *tert*-leucine moiety.

The precursor ion m/z 276 (B1) detected, which was 74 Da lower than that for the 4F-MDMB-BINACA ester hydrolysis metabolite (B22), indicated N-dealkylation of B22. The product ion of B1 detected at m/z 230 was assigned as a neutral loss of HCOOH.

N-Dealkylation

The precursor ion m/z 290 (B16) which was 74 Da lower than the corresponding m/z for 4F-MDMB-BINACA, indicated N-dealkylation of 4F-MDMB-BINACA. Product ions m/z 258, 230 and 145 showed loss of a single methoxy group, methyl ester and indazole carbonyl fragment, respectively. Taken together these data further confirmed the structure elucidation of B16.

Oxidative defluorination and oxidation

Metabolite B18 with precursor ion at m/z 362 and product ions at m/z 330, 302, 217 were consecutively 2 Da less than that of the precursor and product ions of 4F-MDMB-BINACA, being indicative of oxidative defluorination. Further oxidation of 4'-hydroxybutyl moiety of B18 resulted in 4'-carboxybutyl, B17 (m/z 376) having product ions m/z 316, 231 and 213. The product ions at m/z 316 and 231 in B17 were 14 Da higher than that of product ions 302 and 217, respectively, in B18, consistent with the formation of a butanoic acid metabolite.

Phase II biotransformation

Glucosidation

Precursor ion, m/z 542 (B7) resulted from glucosidation of B23/B25. Both have product ions m/z 320, 235 and 161. The product ion detected at m/z 380 was produced by neutral loss of 162 Da (glucoside).

Sulfation

Precursor ion, m/z 460 (B11) detected was a result of sulfation of intermediate metabolite similar to B23/B25. The product ion detected at m/z 235, indicating loss of sulfate, confirmed the identity of the sulfation metabolite.

Comparison among the in-vitro models

The major in-vitro phase I metabolites (> 10% peak area ratio) tentatively identified were monohydroxylation at the butyl side chain and oxidative defluorination to butanoic acid for the *C. elegans* model. HLM model produced four major metabolites: ester hydrolysis with/without dehydrogenation, monohydroxylation at indazole moiety and oxidative defluorination. HepG2 model only produced one major metabolite (B22) which was the ester hydrolysis of 4F-MDMB-BINACA (Table 2). Although the metabolite ranking may differ in all the three models, four common metabolites amongst the seven most abundant metabolites were identified (Fig. 3 a–d): ester hydrolysis (B22, *m/z* 350), ester hydrolysis followed by dehydrogenation (B20, *m/z* 348), oxidative defluorination (B18, *m/z* 362) and further oxidation to butanoic acid (B17, *m/z* 376).

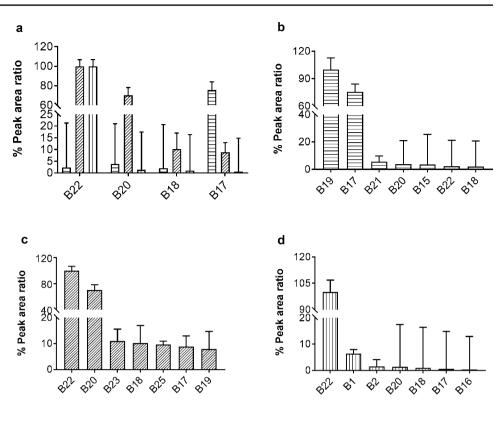
Table 2 Comparison of the in-vitro metabolites from the three models (n=3) with the in-vivo metabolites obtained from authentic human urine samples (n=20)

ID	Biotransformation	In-vivo metabolites % peak area ratio (sample size)	In-vitro metabolites % peak area ratio			
			C. elegans	HLM	HepG2	
B22	Ester hydrolysis	100 (20/20)	2.3	100	100	
B20	Ester hydrolysis, dehydrogenation	1.7–9.4 (18/20)	3.8	70	1.4	
B4	Ester hydrolysis, oxidative defluorination to butanoic acid	1.5–27 (16/20)	0.016	0.25	ND	
B8	Ester hydrolysis, monohydroxylation (tert-leucine)	3.7–14 (16/20)	0.067	1.9	0.14	
B1	Ester hydrolysis, N-dealkylation	0.65-29 (15/20)	0.003	0.73	6.4	
B18	Oxidative defluorination	1.4–5.4 (13/20)	2.1	10	1.0	
B2	Ester hydrolysis, oxidative defluorination	2.0-11 (12/20)	0.005	0.67	1.5	
B17	Oxidative defluorination to butanoic acid	0.26-1.4 (7/20)	<i>76</i>	8.9	0.57	

Major in-vitro metabolites (not less than 10% peak area ratio) are in italic and bold ND not detected



Fig. 3 a In-vitro metabolites observed in common among respective seven most abundant metabolites in \mathbf{b} *C. elegans*, \mathbf{c} HLM and \mathbf{d} HepG2 models. *C. elegans*, HLM and HepG2 are represented by horizontal, diagonal and vertical bar lines, respectively. Error bars reflect the relative abundance variation of a specific metabolite within the analysed group (n = 3)



ID	Biotransformation
B1	Ester hydrolysis, N-dealkylation
B2	Ester hydrolysis, oxidative defluorination
B15	Dihydroxylation (indazole, butyl)
B16	N-dealkylation
B17	Oxidative defluorination to butanoic acid
B18	Oxidative defluorination
B19	Monohydroxylation (butyl)
B20	Ester hydrolysis, dehydrogenation
B21	Monohydroxylation (butyl)
B22	Ester hydrolysis
B23	Monohydroxylation (indazole)
B25	Monohydroxylation (indazole)

In-vivo metabolism

A total of eight metabolites were detected in the analysed human urine samples (Tables 2, 3). The average % peak area ratio of the urine metabolites from twenty urine samples were tabulated and compared to that of the in-vitro metabolites (Table 2).

Ester hydrolysis biotransformation constituted the dominant metabolic pathway for the in-vivo metabolism in human urine samples. The prevalence of the tentatively identified metabolites was as follows: 4F-MDMB-BINACA ester hydrolysis metabolite (B22, *m/z* 350, 20/20 samples) followed by further dehydrogenation (B20, *m/z* 348, 18/20 samples). Both ester hydrolysis followed by oxidative defluorination to butanoic acid (B4, *m/z* 362) and monohydroxylation at *tert*-leucine moiety (B8, *m/z* 366) metabolites were found

in 16/20 urine samples (Table 2). Ester hydrolysis followed with *N*-dealkylation (B1, *m/z* 276) metabolite was found in 15/20 urine samples. Oxidative defluorination (B18, *m/z* 362) and ester hydrolysis of oxidative defluorination (B2, *m/z* 348) metabolites were detected in 13/20 and 12/20 urine samples, respectively. Oxidative defluorination to butanoic acid metabolite (B17, *m/z* 376) was detected in the least number of urine samples (7/20 samples). Only 3/20 urine samples were identified with the parent drug, 4F-MDMB-BINACA, highlighting the risk of analysing only parent ions and emphasizing the importance of metabolite identification in the forensic and clinical setting. The metabolite ranking was different for different individual urine samples.

The analysed urine samples (17/20 samples) were also detected with other SCBs such as methyl 2-([1-(5-fluoropentyl)-1*H*-indole-3-carbonyl]



Table 3 In-vivo urinary profile of individual urine samples that contain only 4F-MDMB-BINACA metabolites

ID	Biotransformation	% Peak area ratio (ranking)			
		Sample 1	Sample 2	Sample 3	
B22	Ester hydrolysis	100 (1)	100 (1)	100 (1)	
B20	Ester hydrolysis, dehydrogenation	4.9 (4)	ND	4.2 (4)	
B4	Ester hydrolysis, oxidative defluorination to butanoic acid	12 (2)	ND	6.8 (2)	
B8	Ester hydrolysis, monohydroxylation (<i>tert</i> -leucine)	7.4 (3)	ND	5.2 (3)	
B1	Ester hydrolysis, <i>N</i> -dealkylation	ND	ND	0.81 (7)	
B18	Oxidative defluorination	2.2 (5)	ND	2.3 (6)	
B2	Ester hydrolysis, oxidative defluorination	ND	ND	2.7 (5)	
B17	Oxidative defluorination to butanoic acid	0.55 (6)	ND	ND	
B0	4F-MDMB-BINACA	0.020	ND	ND	

amino)-3,3-dimethylbutanoate (5F-MDMB-PICA), methyl 3,3-dimethyl-2-[1-(pent-4-en-1-yl)-1H-indazole-3-carboxamido] butanoate (MDMB-4en-PINACA), methyl [1-(4-fluorobenzyl)-1*H*-indazole-3-carbonyl]valinate (MMB-FUBINACA), 5F-MDMB-PINACA and other drugs such as methamphetamine, N-methyl-1-(3,4-methylenedioxyphenyl)propan-2-amine (MDMA) and ketamine (data not shown). Three out of twenty urine samples were found to contain only metabolites of 4F-MDMB-BINACA (Table 3). From these three samples, sample 2 contained only an ester hydrolysis metabolite (m/z 350). Both samples 1 and 3 have similar first four metabolites in terms of % peak area ratio ranking: (1) ester hydrolysis (ion detected at m/z 350, B22), (2) ester hydrolysis, oxidative defluorination to butanoic acid (ion detected at m/z 362, B4), (3) ester hydrolysis, monohydroxylation (ion detected at m/z 366, B8) and (4) ester hydrolysis plus dehydrogenation (ion detected at m/z 348, B20).

Discussion

Glucosides (B7) and sulfate (B11) phase II metabolites of 4F-MDMB-BINACA were tentatively identified in the *C. elegans* model. Glucosides and sulfate metabolites have been reported with other SCBs where *C. elegans* was employed as the in-vitro model [24]. Sulfate and glucuronide phase II metabolites of 4F-MDMB-BINACA was identified by Wagmann et al. [29]. However, glucuronides of 4F-MDMB-BINACA were not observed in the *C. elegans* model, which is consistent with findings from previous studies [23, 24]. Rydevik et al. [33] described an easy method to generate endogenous glucuronide conjugates from glucosides using free radical tetramethylpiperidinyl-1-oxy, hence proving

C. elegans' relevance in mammalian phase II metabolism. Nevertheless, phase I metabolites are usually better marker metabolites compared to phase II metabolites because of its higher mass spectrometry responses and stability over time [34].

The most prevalent and abundant in-vivo metabolite was that derived from the hydrolysis of metabolically labile terminal methyl ester moiety via a two-step reaction mechanism by the carboxylesterases (CES) [35]. The finding is similar to that obtained by Krotulski et al. [5], Haschimi et al. [30] and Wagmann et al. [29]. This outcome was anticipated since CES-mediated hydrolysis is commonly reported as the major metabolic pathway among the SCBs impacting the terminal ester group [36]. Similar to the in-vivo findings, 4F-MDMB-BINACA ester hydrolysis (B22) was the major metabolite for both HepG2 and HLM models, consistent with the known hydrolytic activity of CES reported [37]. Wagmann et al. [29] established that ester hydrolysis of 4F-MDMB-BINACA is catalysed by the CES1 isoform, in agreement with the preference of CES1 in metabolizing ester substrates that contain a small alcohol group and a bulky acyl group [35]. As CES constitutes the major metabolic pathways for the metabolism of 4F-MDMB-BINACA, interindividual CES1 variability due to genetic polymorphism cannot be overlooked [38].

Ester hydrolysis plus dehydrogenation metabolite (B20) was the second most abundant major metabolite in the HLM model, similar to the in-vivo findings where it is the second most prevalent metabolite detected in 18/20 urine samples (Table 2). Ester hydrolysis with dehydrogenation formed in-vivo in this study was also reported among other indazole carboxamide type SCBs with *tert*-leucine methyl ester moieties such as 5F-MDMB-PINACA [15] and MDMB-4en-PINACA [39, 40]. This biotransformation was reported by



both Krotulski et al. [5], and Haschimi et al. [30]. However, it was proposed by Wagmann et al. [29] that this metabolite was more likely a lactone product formed after hydroxylation of the tert-leucine moiety, with or without former ester hydrolysis of 4F-MDMB-BINACA, based upon the presence of fragments with m/z 320 ($C_{17}H_{23}FN_3O_2^+$) and m/z 113 (C₆H₉O₂⁺), a 4,4-dimethyl-2-oxotetrahydrofuran-3-ylium ion in positive ionisation mode, which appears to be more easily formed from lactone, and the absence of these lactone metabolites in negative ionisation mode due to its non-ionizable nature in contrast to ionizable carboxylic acids. Although this proposal sounds plausible, it is not without question. The presence of m/z 330 ($C_{18}H_{21}FN_3O_2^+$) in their spectrum of "lactone" indicates loss of H₂O from the precursor ion and similarly the presence of m/z 332 $(C_{18}H_{23}FN_3O_2^+)$ due to loss of H₂O was observed for ester hydrolysis metabolite, and yet such a H₂O loss fragment ion was not observed for the parent molecule retaining a methyl ester moiety. These results suggest that this compound is more likely to have a carboxylic acid than an ester, which contradicts the aforementioned reasoning by Wagmann et al. [29]. The same authors also proposed lactone formation for the B3 and B14 metabolites. Careful examination of results in this study showed the presence of the ion at m/z 113 in very low signal intensity for B20 but not found for both B3 and B14 product ion spectra. Moreover, B3 and B14 were detected in the negative ESI mode but not for B20 (Table 4). Based upon these findings, it appeared that both lactone and dehydrogenation metabolites were present, but since mass spectral results can be inconclusive and indeed seemed to be contradictory in this case, further comprehensive structural analysis by nuclear magnetic resonance (NMR) spectroscopy is needed for confirmation.

Oxidative defluorination metabolite (B18), the second major metabolite (% peak area ratio > 10) in HLM model was detected as the third least prevalent metabolite in the

in-vivo samples. Oxidative defluorination with subsequent butanoic acid formation (B17) metabolite, the second major metabolite after monohydroxylation in the C. elegans model was the least prevalent metabolite found among the 20 urine samples. Fluoro-substituted SCBs at the 5-pentyl position of pentylindole/pentylindazole moiety was found to give rise to predominantly 5-hydroxypentyl and 5-pentanoic acid metabolites [41]. However, oxidative defluorination with (B17) or without further butanoic acid (B18) metabolite was found only in 7 and 13 urine samples, respectively. The B18 metabolite was ranked third by Haschimi et al. [30], whilst B17 was ranked second by Krotulski et al. [5] in terms of relative peak area abundance. One possible explanation for the differences was that both Haschimi et al. [30] and Krotulski et al. [5] used relative peak area abundance for the ranking, but for this study, the metabolite prevalence in the urine samples was employed instead. The major invitro metabolites discussed above: B22, B20, B18 and B17 (Table 2) were also the four common metabolites identified among the seven most abundant metabolites for each model (Fig. 3 a-d), demonstrating complementary in-vitro findings that may assist in further streamlining the selection of suitable urinary marker (s).

As most of the urine samples obtained in this study were found to contain other SCBs' metabolites and other drugs, the urinary profile of the three individual urine samples that encompass 4F-MDMB-BINACA metabolites exclusively (Table 3) is worth considering when selecting a suitable urinary marker. The four most abundant metabolites identified (B22, B4, B8 and B20) among the three urine samples almost coincided with the top four most prevalent metabolites found in the 20 urine samples. This indicated that the phase I metabolism of 4F-MDMB-BINACA are unlikely to be affected significantly by polydrug intake. Wagmann et al. [29] identified CYP3A5, CYP3A4, CYP1A2, CYP2C8 and CYP2C19 involvement in the phase I metabolism of

Table 4 Product ion spectra of B3 and B14 metabolites obtained in the negative electrospray ionization mode

ID	RT (min)	Biotransformation	Precursor ion exact mass [M-H] ⁻	CE (eV)	Formula	Product ions exact mass [M ⁻]	Product ions mass error (ppm)
В3	5.42	Ester hydrolysis, oxidative defluorination to butanoic	358.1408	10	$C_{14}H_{14}N_3O_3$ $C_7H_8NO_3$	272.1041 154.0510	3.3 2.6
		acid, dehydrogenation			$C_7H_5N_2$	117.0458	4.3
B14	6.75	Ester hydrolysis, monohy-	362.1522	20	$C_{17}H_{21}FN_3O_2$	318.1623	12
		droxylation (tert-leucine),			$C_{12}H_{13}FN_3O$	234.1048	4.3
		11 12 2	191.0990	-11			
					$C_{11}H_{11}N$	171.0928	-2.3

CE collision energy

Mass error of the precursor ion \leq 5.00 ppm



4F-MDMB-BINACA and also excluded the possibility of a single CYP enzyme that may impact the in-vivo biotransformation of 4F-MDMB-BINACA.

It was difficult to assign a ranking to the in-vivo metabolites based upon % peak area ratio because wide inter-individual variability was observed among the metabolites. The % peak area abundance ratio of metabolites detected in the urine samples are often affected by numerous factors such as drug intake behaviour (intake route, amount of drug and intake frequency), time from last drug intake [16] and metabolic stability. It should be noted that % peak area abundance ratios do not necessarily reflect absolute concentrations due to differences in ionization capacity and matrix effects bias for each metabolite. The exact ranking of metabolites with respect to their peak area abundance would require further validated quantification of the metabolites. Moreover, genetic makeup, physiological conditions (age, gender and ethnicity), environmental influences (diet) and pathological factors (liver diseases, diabetes, and obesity) would further complicate the metabolism of drugs. Thus, identification of the relevant urinary markers was based primarily upon the prevalence of the in-vivo metabolites instead of the metabolites ranking that was based upon % peak area abundance ratio. The urinary marker selected would also be preferably found in all three in-vitro models especially in the C. elegans model where metabolite production could be scaled up to allow isolation and characterization of the metabolites.

Based upon the two criteria and the above findings suggested relevant in-vivo urinary markers were ester hydrolysis, m/z 350 (B22, in 20/20 samples) and ester hydrolysis followed by dehydrogenation, m/z 348 (B20, in 18/20 samples) metabolites. Ester hydrolysis with subsequent oxidative defluorination to butanoic acid metabolite (m/z 362, B4) detected in 16/20 urine samples seemed to be a choice urinary marker as it appeared to have a higher % peak area ratio when compared to ester hydrolysis plus dehydrogenation metabolite (B20). However, this metabolite was only detected in trace amounts in both the C. elegans and HLM models. Hence, structural elucidation could not be confirmed unless a reference standard is made available. Despite the relatively high % peak area ratio and detection in 16/20 urine samples, ester hydrolysis followed by monohydroxylation at the *tert*-leucine moiety (*m/z* 366, B8) metabolite was not reported among the 17 urine samples from the forensic psychiatric ward and prison in Haschimi et al.'s study [30]. Hence, the use of this metabolite as the urinary marker was not selected.

Finally, a comparison of the results obtained from incubations of 4F-MDMB-BINACA with three different models showed that all of them were in agreement with the tentatively identified urinary metabolites. *C. elegans* and HLM models detected all of the in-vivo metabolites (100%), whilst HepG2 cells detected 7 out of the 8 in-vivo metabolites (87.5%).

HepG2 model detected the major ester hydrolysis metabolite of 4F-MDMB-BINACA in abundance but the rest of the metabolites were found in a small amount. This might be due to the low activity of numerous metabolizing enzymes resulting in lower drug biotransformation [42]. Comparison of C. elegans with in-vivo metabolites showed that while C. elegans has limitations such as a low abundance of ester hydrolysis metabolite, in general, the metabolic profiles were in good agreement. The findings also supported that C. elegans possessed enzymatic activity that mimics the human metabolism of 4F-MDMB-BINACA. HLM possessed both the CYP and carboxylesterase enzymes accounted for biotransformation that matches with the in-vivo findings. It should be noted that the chemical structures of the tentative elucidated metabolites have not been unequivocally confirmed. With C. elegans model, scaling up of metabolites production for NMR spectroscopy could provide unequivocal structural elucidation of the metabolites. C. elegans has also been reported to form and trap active metabolites using trapping agents such as glutathione [28]. Scaling up these potential active metabolites conjugates allows the identification of potential metabolite toxicities thus giving an enhanced understanding of the toxicity mechanism of 4F-MDMB-BINACA. Therefore, C. elegans is a useful complementary metabolic model for SCBs, especially when a larger quantity of metabolites is required.

Conclusions

The present study showed that HLM proved to be a suitable predictive in-vitro – in-vivo extrapolation model for 4F-MDMB-BINACA. C. elegans demonstrated the ability to form all of the in-vivo metabolites and has the potential to be used as a complementary model to predict and characterize human metabolites, as well as identifying possible drug toxicities for emerging SCBs. For the identification of 4F-MDMB-BINACA intake, ester hydrolysis (B22) and ester hydrolysis followed by dehydrogenation (B20) metabolites were identified as specific and relevant urinary markers. The difficulty in unequivocally identifying the structures of dehydrogenation metabolites including B20 by mass spectrometry demonstrated the need to utilise NMR spectroscopy. Future work will include isolation and confirmation of scaled-up reactions by C. elegans model to confirm the proposed chemical structure assignments to the metabolites identified here.

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Compliance with ethical standards

Conflict of interest The authors do not have financial or other relations that could lead to a conflict of interest.

Ethical approval This article does not contain any studies with human participants or animals performed by any of the authors.

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