

1 **Significance of the presence of antibiotics on the a microbial consortium in**  
2 **wastewater - the case of nitrofurantoin and furazolidone**

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22 **Highlights**

- 23 • Nitrofurantoin and furazolidone had no adverse effect on the microbial  
24 consortium.
- 25 • The consortium behavior and degradation of antibiotics differed between the  
26 drugs.
- 27 • NFT reduced organic carbon consumption and increased EPS and VOC  
28 production.
- 29 • FZD decomposition exceed 60% with a small reduction of organic carbon  
30 assimilation.
- 31 • The consortium behaviour was better represented by the logistic mathematical  
32 model.

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48 **Abstract**

49 Antibiotics presence in wastewater leads to migration of pollutants and disrupts natural  
50 processes of mineralization of organic matter. In order to understand the mechanism of  
51 this, research was undertaken on the influence of nitrofurantoin (NFT) and furazolidone  
52 (FZD), on the behaviour of a consortium of microorganisms present in a model  
53 wastewater in a bioreactor. Our study confirmed biodegradation of the antibiotics by the  
54 microbial consortium, with the degradation efficiency within 10 days of 65% for FZD,  
55 but only 20% for NFT. The kinetic study proved that the presence of analysed antibiotics  
56 had no adverse effect on the microbes, but the consortium behaviour differ significantly  
57 with the NFT reducing the consumption of organic carbon in wastewater and increasing  
58 the production of extracellular biopolymeric and volatile organic compounds, and the  
59 FZD reducing assimilation of other carbon sources to a less extent, at the expense of  
60 cellular focus on biodegradation of this antibiotic.

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62 **Keywords:** nitrofurantoin, furazolidone, kinetics, microbial consortium, wastewater

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64 **1. Introduction**

65 At the time of growing concern about antimicrobial resistance, the antibiotic nitrofurans  
66 remain their effectiveness even against antibiotic-resistant bacteria. This might be due to  
67 their multiple mechanisms of action such as inhibition of DNA and RNA synthesis, a  
68 metabolic enzyme, carbohydrate metabolism, and reactive oxygen species production  
69 (Gallardo-Garrido et al., 2020). In 2019, the World Health Organization (WHO) listed  
70 nitrofurans as important antimicrobials for human medicine (WHO, 2019). Currently,

71 nitrofurans are widely used in the treatment of urinary tract infections (Vass et al., 2008),  
72 and are tested as antitubercular agents (Elsaman et al., 2019).

73 Although the administration of nitrofurans has been effective, their high elimination from  
74 human and animals may raise concerns. For nitrofurantoin (NFT), 90% of the total dose  
75 is eliminated in the urine, with up to 50% excreted in unchanged form in humans  
76 (DrugBank Online, 2021). Similarly, 55–65% of the primary dose of furazolidone (FZD)  
77 is eliminated by renal excretion in humans and animals (White, 1989). These antibiotics  
78 residues can enter the wastewaters and soil, where sorption onto organic particles occurs  
79 (Cycoń et al., 2019; Nguyen et al., 2019a; 2020). Recent studies revealed, that NFT  
80 sorption in sediments and soils depends on the ionic strength and pH of the sorbent and  
81 can range from 3.967 to 5.121 mL g<sup>-1</sup>, and from 3.634 up to 43.06 mL g<sup>-1</sup> in sediments  
82 and soils, respectively (Tolić et al., 2019). Therefore, both nitrofurans drugs, NFT and  
83 FZD, may cause serious environmental problems due to their high ecotoxicity  
84 (Lewkowski et al., 2019). Mutagenicity and ecotoxicity of nitrofurans derivatives to  
85 environmental organisms have been investigated since 80s. Macri' and Sbardella (1984)  
86 found that the highest toxicity to *Selenastrum capricornutum* and *Daphnia magna* possess  
87 nitrofurazone, followed by furaltadone tertrate and furaltadone chlorohydrate. Significant  
88 toxicity of FZD on *Culex pipiens* and *Daphnia magna* was also reported later by Macri  
89 et al. (1988). More recent studies described the negative effect of furaltadone on *Ulva*  
90 *lactuca* (microalgae) (Leston et al., 2011) and the significant toxicity of FZD and NFT  
91 on *Aliivibrio fischeri* (bacteria) and *Heterocypris incongruens* (crustaceans) (Lewkowski  
92 et al., 2019).

93 For both, environmental sorption (Tolic, et al. 2019) and biodegradation (Pacholak, et al.  
94 2019) only limited studies on the process kinetics are available. However, literature

95 reports describing the fate of antibiotics in the environment, including municipal  
96 wastewater, focus on extremely different studies that are difficult to compare (Chaturvedi  
97 et al., 2021; Ruan et al., 2020; Wu et al., 2011). The first group consists of studies on  
98 biodegradation pathways and effects of pharmaceuticals on single and defined strains of  
99 microorganisms (e.g. Pan et al., 2018). The second group of studies includes those  
100 conducted with real wastewater and a complex consortium of microorganisms (e.g. Tang  
101 et al., 2017; Tolić et al., 2019). There are relatively few works of an indirect character,  
102 being a simplification of the real system, however verifying the phenomena on the macro  
103 level (e.g. Azimi et al., 2017; Peng et al., 2018; Miran et al., 2018). To fill in these  
104 knowledge gaps, we decided to analyse a close-to-real system based on the results of the  
105 strictly molecular and simplified systems based on our previous experience Pacholak et  
106 al. (2019). In previous study the NFT degradation was conducted and the kinetics of the  
107 process performed by single strains in simplified culture, with NFT as the only carbon  
108 source (Pacholak et al., 2019).

109 To mimic better the environmental conditions, and analyse the processes from a broader  
110 perspective we choose a system containing synthetic wastewater with a consortium of  
111 microorganisms of known composition, and the aim of this study was to understand the  
112 effect of the presence of NFT and FZD on the behaviour of the microorganisms.  
113 Therefore, analyses of growth kinetics, organic matter assimilation and antibiotic  
114 biodegradation were undertaken, with appropriate mathematical models proposed for the  
115 results obtained for greater precision. Nevertheless, an important aspect of the research  
116 was the analysis of the chemical composition of biomasses, as well as the profile of  
117 volatile organic compounds, as valuable markers of changes in the cell metabolism of the  
118 bacteria forming the consortium.

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## 120 **2. Materials and methods**

### 121 *2.1. Chemicals and bacterial strains*

122 Nitrofurantoin (NFT), furazolidone (FZD), and Luria-Bertani broth (LB broth) were  
123 purchased from Sigma-Aldrich (Poland). Organic solvents and chemicals were analytical  
124 grade. Synthetic wastewater was prepared according to OECD procedure (160 mg  
125 peptone; 110 mg meat extract; 30 mg urea; 28 mg K<sub>2</sub>HPO<sub>4</sub>; 7 mg NaCl; 4 mg CaCl x 2  
126 H<sub>2</sub>O; 2 mg MgSO<sub>4</sub> x 7 H<sub>2</sub>O) (OECD, 2001).

127 Five bacterial strains including, *Glutamicibacter nicotianae* AsK3a (K3a, GenBank  
128 MK503993.1), *Arthrobacter* sp. AsK4c (K3c, GenBank MN960430.1), *Ochrobactrum*  
129 sp. AsP4c (P4c, GenBank MN960431.1), *Pseudomonas aeruginosa* Asp4a (P4a,  
130 GenBank MK503653.1), and *Arthrobacter* sp. strain AsP3d (P3d, GenBank  
131 MN960429.1) were selected for the study. To obtain the bacterial consortium for this  
132 study, each strain was incubated at 25 °C for approximately 48 h until the optical density  
133 at 600 nm exceeded 1. The cultures were then centrifuged and resuspended in synthetic  
134 wastewater until an optical density of OD<sub>600</sub>≈1.0 (corresponding to 1x10<sup>8</sup> CFU mL<sup>-1</sup>).  
135 Finally, 50 mL of each suspension was mixed together (in equal volumetric proportions)  
136 to make the microbial consortium, which was used further to inoculate the batch culture.

### 137 *2.2. Biodegradation experiment*

138 The microbes were enumerated in LB broth for 24 h, centrifuged and re-dissolved in  
139 synthetic wastewater A laboratory-scale bioreactor with 1.5 L (Fig. S1 in E-  
140 supplementary data of this work, which can be found in online version of the paper)  
141 working volume was used in the experiments (Biostat B plus, Sartorius). The created  
142 bacterial consortium was cultured for 24 h in a batch reactor with 1.5 L of synthetic

143 wastewater, at 25 °C and with stirring 150 rpm to enumerate bacteria. Antibiotic, NFT or  
144 FZD, was added into each bioreactor at 5 mg L<sup>-1</sup> after 24 h to begin the degradation kinetic  
145 experiment for another 10 days. During the period of 10 days, the operation conditions  
146 were maintained at 25 °C and 150 rpm. Any potential photodegradation of antibiotics in  
147 the reactor was avoided by keeping it in darkness. Sample (30 mL) was taken every day  
148 for 10 days, centrifuged and stored at -20 °C for further analysis. For the control batch,  
149 synthetic sludge without any additions was used.

### 150 *2.3. NFT and FZD concentration analysis*

151 High performance liquid chromatograph with tandem mass spectrometry detector  
152 (HPLC-MS/MS) was used to determine the residual NFT and FZD content in the  
153 biodegradable samples. The system consists of the UltiMate 3000 RSLC chromatograph  
154 from Dionex and the API 4000 QTRAP triple quadrupole mass spectrometer from AB  
155 Sciex. The samples taken during the experiment were introduced in the amount of 5 µL  
156 on the Gemini-NX C18 chromatography column (100 mm × 2.0 mm i.d.; 3 µm) from  
157 Phenomenex maintained at 35 °C. The mobile phase consisting of 5 mM ammonium  
158 acetate in water and methanol was flowing through the column at a rate of 0.3 µL min<sup>-1</sup>.  
159 The gradient elution was performed by linearly increasing the percentage of organic  
160 modifier from 75 to 80% in 2 min and then from 80 to 100% in 1 min. A pre-run time of  
161 3 min was used before the next injection. The column eluate was directed to the  
162 electrospray ionization source (the Turbo Ion Spray) of the mass spectrometer. The Turbo  
163 Ion Spray source operated in positive and negative ion mode for FZD and NFT  
164 respectively. The following settings were used for the ion source and mass spectrometer:  
165 curtain gas 10 psi, nebulizer gas 40 psi, auxiliary gas 40 psi, temperature 400 °C, ion  
166 spray voltage +/-4500 V, declustering potential +/-60 V and collision gas set to medium.

167 The dwell time for each mass transition detected in the selected reaction- monitoring  
168 mode was set to 200 ms. The quantitative transition for FZD was from 226 to 122 m/z at  
169 collision energy set to 29 eV and collision cell exit potential was set to 6 V. The  
170 confirmatory transition was from 226 to 95 m/z at collision energy set to 21 eV and  
171 collision cell exit potential was set to 4 V. The quantitative transition for NFT was from  
172 237 to 152 m/z at collision energy set to -17 eV and collision cell exit potential was set  
173 to -10 V. The confirmatory transition was from 237 to 124 m/z at collision energy set to  
174 -20 eV and collision cell exit potential was set to -10 V.

#### 175 *2.4. Bacterial biomass analysis*

176 The infrared (IR) spectroscopic analysis of lyophilized bacterial biomass was intended to  
177 highlight specific groups on biomass surface, what allows to evaluate the biomass  
178 composition changes, as well as to verify if the drugs were adsorbed on the bacteria  
179 surface (what could confirm by presence of IR signals characteristic for nitrofurans). The  
180 bacterial biomass was separated by centrifugation (4500 rpm, 20 min, 4 °C). The cells  
181 pellet elemental composition and chemometric analysis were performed, while the liquid  
182 culture was used in further experiments. The biomass was weighted and afterwards  
183 lyophilized for 48 h, -50 °C, 0.36 mbar (Alpha 1.4 LD plus, Christ). Elemental analysis  
184 of nitrogen, carbon, hydrogen and sulphur content in samples was performed for 7 mg  
185 samples on Vario EL Cube apparatus (Elementar) to trace the dynamics of the elements  
186 during decomposition process. ATR-FTIR spectroscopy was used to determine the  
187 occurrence of the chemical functional groups on the surface of lyophilized cells biomass  
188 (Vario 70, Bruker).

#### 189 *2.5. Organic matter metabolism in microbial cultures*

##### 190 *2.5.1. Organic matter assimilation*

191 Total organic carbon (TOC) measurements were conducted using TOC-L analyzer  
192 (Shimadzu), equipped with OCT-L 8-port autosampler (Shimadzu). The inorganic carbon  
193 was separated from the organic one by automatic acidification of the sample. Then sample  
194 was injected into the first of the two zones of the furnace (1000 °C), where was dried and  
195 burned, and in the second zone was oxidized to CO<sub>2</sub> in the presence of catalyst (copper  
196 (II) oxide). The carbon concentration was calculated based on the total peak value of the  
197 measured signal, after processing with the corresponding calibration curve.

#### 198 *2.5.2. Production of volatile organic compounds (VOCs)*

199 The supernatant after samples centrifugation was subjected to properties and  
200 chromatography analysis. The Volatile Organic Compounds (VOC) content was  
201 evaluated to identify the main microbial VOCs (mVOCs) components during microbial  
202 bioremoval by microextraction from the headspace (HS-SPME), followed by separation  
203 of the analytes by gas chromatography and their identification by mass spectrometry (GC-  
204 MS). The 5 mL of culture were transferred to twisted glass chromatographic vials of 20  
205 mL volume containing 1 g of sodium chloride and 1 g of citric acid each. Successively,  
206 the samples were heated for 30 min at 60 °C, and then the PEG/PDMS fiber (Merck)  
207 adsorbent was introduced into headspace of the sample, on which the analytes were  
208 adsorbed for 10 min at 60 °C. Then, desorption from the fiber was carried out at a  
209 temperature of 250 °C for 1 min. in a gas chromatograph injector (Pegasus 4D, Leco)  
210 with a BPX-5 column (28 m, 250 µm, 0.25 µm). The analysis was carried out with helium  
211 as the carrier gas (flow 1 mL min<sup>-1</sup>). The device worked in the programmed temperature  
212 changes mode: 40 °C for 2 min, increase of 20 °C min<sup>-1</sup> to 100 °C, then increase of 7 °C  
213 min<sup>-1</sup>. up to 280 °C. The mass spectrometer was operated in the positive ion analysis  
214 mode at a voltage of the ion source of 70 V. The identification of the analytes was carried

215 out on the basis of the obtained mass spectra on the basis of the ChromaTof software  
216 spectrum library (Leco Corp.). Quantification of compounds was carried out on the basis  
217 of a previously prepared calibration curve, based on the correlation between the peak area  
218 in the chromatogram and the standard compound concentration.

### 219 *2.5.3. Production of extracellular macromolecules*

220 Surface tension measurement and emulsification assay were performed to evaluate  
221 changes in extracellular compounds secretion upon antibiotics exposition. The surface  
222 tension of the supernatant after centrifugation was determined using tensiometer (K20,  
223 Kruss,) and DuNuoy ring method. The emulsifying activity was determined as the ratio  
224 of the height of the emulsified layer to the total height that was occupied by the emulsion  
225 after a defined time period. Emulsion samples of 20 mL were prepared in sterile 50 mL  
226 plastic laboratory tubes, combining equal volumes of supernatant after centrifugation and  
227 hexadecane (analytical grade, Sigma-Aldrich). The samples were first mixed (Vortex 3,  
228 IKA) at 2500 rpm for 15 s, and then homogenized (sonicator Sonoplus, Bandelin) in the  
229 following conditions: 10 min, in cycles action/break 10 s/10 s, amplitude 16%, and cooled  
230 with tap water. The emulsion stability tests were conducted after 24 h and the  
231 emulsification index (E24) was calculated.

### 232 *2.6. Kinetic calculations*

233 To describe the kinetics of biological processes different mathematical model are  
234 commonly applied. A study by Schmidt et al. (1985) contains a wide spectrum of  
235 equations and discuss two main parameters of the bioprocess, i.e. biomass growth and  
236 substrate concentration. They distinguished four classes of biomass growth models and  
237 three options of substrate concentration. In our study, the biomass growth was observed  
238 what allowed to exclude the non-growth models, that were frequently characteristic for

239 biodegradation of pharmaceuticals, what was underlined by Pan et al. (2018) in their  
240 publication. As a consequence Michaelis-Menten or Monod equations cannot be applied  
241 (Briones et al., 2018). Finally, after analysis of several models, the logistic growth model  
242 was found as the most suitable, and it was described with the following equations in  
243 standard (Eq. 1) and general differential form (Eq. 2):

$$244 \quad X(t) = \frac{X_{max}}{1+e^{-kt}} \quad (\text{Eq.1})$$

$$245 \quad \mu = \frac{d}{dt} X(t) = \frac{X_{max}ke^{-kt}}{(e^{-kt}+1)^2} \quad (\text{Eq.2})$$

246 where:

247  $X$  represents the biomass concentration during the process ( $\text{mg L}^{-1}$ ),

248  $X_{max}$  represents the final biomass concentration (theoretical) ( $\text{mg L}^{-1}$ ),

249  $t$  represents the time of the process (day),

250  $k$  represents the logistic growth rate coefficient [ $(\text{mg L}^{-1})^{-1} \text{day}^{-1}$ ],

251  $\mu$  represents the growth rate ( $\text{mg L}^{-1} \text{day}^{-1}$ ).

252 Moreover, the kinetic curves of the function  $\mu(t)$  allow calculating the maximum value of  
253 the function, i.e. maximum growth rate,  $\mu_{max}$ . In that case  $t_{max}$  representative the time,  
254 when  $\mu_{max}$  occurred. Additionally, we calculated the average growth rate,  $\mu_{avr}$  as the ratio  
255 of total biomass increase to the time of the process.

256 To evaluate the substrate biodegradation kinetic, first order, second order and Michaelis-  
257 Menten models were considered (Jia et al., 2017; Schmidt et al., 1985). To describe the  
258 relationship between the substrate concentration, mathematical models taking into  
259 account polynomial equations of higher orders were also considered.

## 260 *2.7. Statistical analysis*

261 All the analysis were made in triplicate. The statistical significance of data was analyzed  
262 using a one-way analysis of variance (ANOVA) (SigmaPlot 11.0 program). A probability

263 (P) value of  $<0.05$  was considered significant. The data are presented as mean  $\pm$  standard  
264 deviation (SD).

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### 268 **3. Results and discussion**

#### 269 *3.1. Antibiotics degradation in bioreactor*

270 Concentration measurements of the antibiotics in bioreactor cultures showed that the  
271 removal of both antibiotics occurred, however, the efficiency of the antibiotics' decay  
272 was different (Fig. 1A). In the first 24 h, the removal of FZD was above 50%. In the next  
273 9 days, the removal increased to a maximum of 65%. On contrary, the removal of NFT  
274 was 23% after 10 days experiment, with the same initial concentration of drugs equal to  
275  $5 \text{ mg L}^{-1}$  (Fig. 1B). The results collected indicated that the biodegradation kinetics was in  
276 the form of a third-degree equation and, especially for cultures with FZD, three stages are  
277 observed - the first and the last of the accelerated substrate uptake and the middle one  
278 with a significant slowdown in degradation. Table 1 presents the conclusions of the  
279 polynomial (third-order) mathematical model, which showed the best fit to the  
280 measurement points. It can be seen from Table 1, that the final degradation rate (on the  
281 10<sup>th</sup> day) of FZD was still at the level of  $0.45 \text{ mg L}^{-1}$  and for NFT decreased down almost  
282 to  $0 \text{ mg L}^{-1}$  (Fig. 1C), whereas the average substrate degradation rate ( $S_{avr}$ ) calculated  
283 based on the polynomial equation was significantly lower for NFT and reached  $0.12 \text{ mg}$   
284  $\text{L}^{-1} \text{ day}^{-1}$  (Table 1). The maximum biodegradation ( $S_{max}$ ) of this compound occurred on  
285 day 1 and was equal to  $0.25 \text{ mg L}^{-1} \text{ day}^{-1}$  based on the model applied. On the contrary, the  
286 FZD reached the maximum of biodegradation on day 8 with the  $S_{max}$  of  $1.9 \text{ mg L}^{-1} \text{ day}$

287 <sup>1</sup>and  $S_{avr}$  0.39 mg L<sup>-1</sup> day<sup>-1</sup>, which confirms that the cooperation of the strains in a  
288 consortium led to the higher biodegradation efficiency of this antibiotic. Nguyen et al.  
289 (2019b) stated that tiamulin antibiotic was biodegraded by the bacterial consortium  
290 (containing *Achromobacter*, *Delftia*, *Flavobacterium*, *Pseudomonas*, and  
291 *Stenotrophomonas* strains) according to logistic model, similarly as we observed in our  
292 study. However, an example of biodegradation following the first-order equation model  
293 was described by Li and Zhang (2010) for cefalexin, sulfamethoxazole, sulfadiazine, and  
294 three fluoroquinolones: norfloxacin, ofloxacin, and ciprofloxacin, which were degraded  
295 by wastewater microorganisms.

### 296 3.2. Organic matter balance

297 Measurements of changes in total organic matter concentration shed additional light on  
298 the results obtained and the kinetic study of the results was performed to evaluate biomass  
299 growth parameters, changes in total organic carbon (TOC) in the samples, as well as to  
300 fit the degradation curves to the existing models. At first, the biomass growth was  
301 evaluated for all, cultures with antibiotics and the reference culture (control). For all tested  
302 systems, it was observed that it corresponds to the logistic growth model, which was  
303 confirmed by the comparison of various mathematical models to measurement points.  
304 Both the graphs (Fig. 2A-D) and the kinetic parameters presented in Table 2 show that  
305 the presence of antibiotics significantly influences the growth of microorganisms. What  
306 is interesting, the presence of xenobiotics stimulated biomass growth (Fig. 2A). The  
307 biomass growth rate profiles for batches differ significantly, indicating that the dynamics  
308 of growth were the lowest for control samples (Fig. 2B-D). While in the control sample  
309 the average biomass growth rate ( $\mu_{avr}$ ) was 4.1 mg L<sup>-1</sup> day<sup>-1</sup>, in the cultures with NFT and  
310 FZD it was 11.0 mg L<sup>-1</sup> day<sup>-1</sup> and 8.4 mg L<sup>-1</sup> day<sup>-1</sup>, respectively (Table 2). At the same

311 time, the maximum growth rate ( $\mu_{max}$ ) was the highest in the control culture and was equal  
312 to 80 mg L<sup>-1</sup> day<sup>-1</sup>, which was at least twice as high as those found for the other two  
313 cultures. This suggests that in the presence of antibiotics, growth was more even, rather  
314 than abrupt, as in the control sample. Additionally, the stage of the most intensive growth  
315 ( $t_{max}$ ) shortens from 6.0 days for the control culture to 2.8 days and 4.8 days in cultures  
316 with NFT and FZD, respectively (Table 2). That is an interesting outcome taking into  
317 account their multiactivity. For most of the antibiotics, the growth of the biomass is rather  
318 suspended, than improved, as was presented e. g. by Kim et al. (2020) who found that  
319 another antibiotic, ciprofloxacin in the concentration of 500 µg L<sup>-1</sup> reduced by more than  
320 tenfold the relative abundance of *Rhodobacteraceae* and *Nakamurellaceae* in the  
321 analyzed samples. In another study, it has been confirmed that multi-hierarchical  
322 antibiotic selection creates an imbalance in the microbiota composition, that is hard to  
323 restore even several months after treatment (Martínez, 2017). However, the maximum  
324 biomass concentration ( $X_{max}$ ), which can be calculated theoretically, was relatively similar  
325 in all cases, ranging from 303 to 348 mg L<sup>-1</sup>, and was nearly the same for both cultures  
326 with the analyzed antibiotics (Table 2).

### 327 3.3. Analysis of mass transfer in biodegradation process

328 Analysis of the results discussed above may lead to interesting conclusions. The tested  
329 antibiotics were degraded by a consortium of microorganisms, although the reduction in  
330 NFT was relatively small. At the same time, a more significant increase in biomass was  
331 observed in the presence of these compounds than in the control sample. However, in the  
332 case of this parameter, a greater increase was observed in cultures with NFT. Moreover,  
333 both antibiotics caused significantly lower assimilation of dissolved organic matter,

334 which can be seen by observing changes in TOC. Based on these observations, the  
335 following course of bioreactor processes can be proposed.

336 • In the presence of NFT, microbial cells manifest response to stress conditions,  
337 which includes a decrease in metabolism, and a relative increase in biomass might  
338 not necessarily mean an increase in cell number, but the production of protective  
339 extracellular macromolecular compounds that coat the cells from the outside  
340 (Mathivanan et al., 2021).

341 • In contrast, the response to FZD is different, despite the high similarity in  
342 chemical structure between NFT and FZD. Cells focus on biodegrading FZDs  
343 instead of assimilating the remaining dissolved carbon source present in  
344 wastewater. Assimilation of FZD (and partially other organic compounds) allows  
345 for an increase in biomass, but presumably a lot of energy is used to switch the  
346 metabolic apparatus to degrading the antibiotic rather than the organic matter  
347 present in the synthetic effluent (Zavala et al., 2019).

348 These results show the complex competition that can exist between metabolizing a  
349 relatively readily available carbon, present for example in peptone, and a rather complex  
350 antibiotic molecule. Moreover, Liu et al., (2017) reported the importance of media  
351 concentration in degradation processes, showing that 1/5 dilution of BEP medium  
352 allowed the most efficient degradation of gentamicin. However, it is important to note  
353 that our study used strains isolated from wastewater through selective cultures  
354 supplemented with nitrofurantoin derivatives (Pacholak et al., 2019). This may explain the  
355 relative ease of adaptation to degrade these compounds.

356 *3.4. Antibiotics impact on bacteria outer surface*

357 According to the FTIR results, the tested compounds did not reveal a significant impact  
358 on the bacterial surface (Fig. S2A in E-supplementary data of this work, which can be  
359 found in online version of the paper). The infrared spectra of the untreated biomass show  
360 signals at 3286  $\text{cm}^{-1}$  (stretching vibrations of amine groups), 2932  $\text{cm}^{-1}$  (stretching  
361 vibrations of  $-\text{CH}$ ,  $-\text{CH}_2$  and  $-\text{CH}_3$ ), 1075  $\text{cm}^{-1}$  (stretching vibrations of C–O bonds) and  
362 526  $\text{cm}^{-1}$  (bending vibrations of C–C bonds). Further, peaks with maxima at 1642, 1535  
363 and 1240  $\text{cm}^{-1}$ , are characteristic for vibrations of amide I (mainly C=O stretching), amide  
364 II (mainly C–N–H bending) and amide III (mainly C–N=O bending), and confirm the  
365 presence of the surface functional proteins. Bacteria cultured with NFT and FZD did not  
366 change significantly in comparison to untreated biomass. Moreover, comparing the  
367 results to the spectra of pure NFT and FZD (Fig. S2B and S2C in E-supplementary data  
368 of this work, which can be found in online version of the paper), characteristic peaks are  
369 not observed in the biomass samples. By contrast, previous research reporting NFT or  
370 FZD attachment to different materials confirmed effective linkage of antibiotics by the  
371 presence of characteristic peaks identified on the FTIR spectra of the samples after the  
372 process. Teoh et al. (2019) observed the presence of N–O, C–O and C–N stretching  
373 vibrations, along with the fingerprint region of 2,5-disubstituted furan ring on co-crystal  
374 NFT/citric acid spectra. Gurav et al. (2020) noted shifts of C=C and C–O bands due to  
375 the interaction with N–O group from FZD. These changes were not observed for biomass  
376 analysed in this study, confirming that the drugs were removed rather by degradation,  
377 than sorption. This is also consistent with Zhang et al. (2013) who found almost none  
378 FZD in the bacterial cells lysates after growth in FZD-containing media, proving that this  
379 compound was also not accumulated inside the cells.

380 Moreover, the performed elemental analysis indicated a significant increase in nitrate and  
381 carbon content in the cells biomass after cultivation with NFT or FZD. The elevated levels  
382 of nitrogen and carbon, to 11.70% and 57.06% for NFT, and 10.95% and 56.51% for  
383 FZD, respectively (Table S1) might be associated with bacterial growth and biomass  
384 increase (Li et al., 2021). What is interesting, in all analysed experiments the C:N ratio  
385 was comparable and reached the value around 5:2. This is also consistent with the biomass  
386 growth kinetic measurements, where the maximum biomass concentration after 10 days  
387 of experiments was ranging from 303 to 348 (mg L<sup>-1</sup>) (Table 2). The impact of growth  
388 conditions on bacterial C, N and P content was investigated, among others by Vrede et  
389 al. (2002). They found that C limitation resulted in the lowest C:N ratio and the elemental  
390 composition of the biomass changed substantially among isolates and under different  
391 growth conditions (Vrede et al., 2002). This was not observed for NFT and FZD growth  
392 bacterial consortia, suggesting that these substances did not act as limiting factors on  
393 bacterial growth. Furthermore, measurements of total organic carbon in the samples  
394 revealed doubling of the amount of the organic carbon in the samples with the TOC of  
395 28.35 and 27.73 mg L<sup>-1</sup> for NFT and FZD, respectively, comparing to 15.85 mg L<sup>-1</sup> for  
396 the control (Table S1).

### 397 *3.5. Changes in microbial VOCs profile*

398 The degradation of complex molecules may result in the formation of microbial volatile  
399 organic compounds (mVOC), a small molecule compounds formed in every living cell.  
400 However, their production is mainly the result of basic metabolism processes. The  
401 emission rates and profiles of mVOCs secretion are dependable on many factors such as  
402 taxonomic relationship, time of growth, temperature and type of substrate, forming  
403 characteristic for the strain and compound fingerprint of the mVOCs (Insam and Seewald,

2010). Intense metabolism under simultaneously unfavorable conditions (e.g., lack of adequate oxygen concentration as an electron acceptor) can lead to their increased production when the cell does not fully mineralize carbon sources (Hidalgo et al., 2019), what we observed in presented study, which documents the differences in profiles of emission of mVOCs during growth on media containing NFT or FZD. To distinguish mVOCs originating from NFT and FZD degradation processes the collected results were compared to control samples, where synthetic wastewater was used as a sole carbon and energy source. Analyses revealed the presence of the methyl ketones, alcohols, esters, carboxylic acids and sulfur-containing compounds in mVOC profiles of the tested cultures (Fig. 3). Moreover, distinct mVOC profiles were observed for cultures grown with NFT and FZD. Although the compounds found in the samples were mostly the same, the amount of mVOC created by NFT and FZD samples, as well as the proportions of particular composites differs significantly (Fig. 3A). The most abundant was 1-Hexanol,2-ethyl- which proportion in treated samples increased to over 70% in the mVOC profile of the samples (Fig. 3B). Also, the presence in the samples of 2-Heptanone raised, from 0.47% (control) to 4.03% and 9.97% for NFT and FZD, respectively which was followed by a decrease in dimethyl trisulfide (DMTS) content in the samples. Interestingly, only three compounds were differentiating between NFT and FZD samples (5-Methyl-2-hexanone; Nonanal, and 2-Butyltetrahydro-furan). In all three cases, NFT-grown cells reduced the amount of these compounds in their mVOC profile, while FZD-grown cells increased their content in the mVOC profile (Fig. 3B). According to Stahl and Parkin (1996), microbial production of volatile organic compounds of bacteria treated with streptomycin varied significantly in respect of the total amount of VOC and their profile in different periods. The authors observed increased production of mVOCs

428 in samples treated with the antibiotic, comparing to other treatment (Stahl and Parkin,  
429 1996), which is consistent with our results. Presented in our study mVOCs were also  
430 detected in decomposition soil samples by Perrault et al. (2014). Compounds reported as  
431 the most abundant in our research, mainly ketones and alcohols are a common result of  
432 macromolecules breakdown. Interestingly, compound abundant in higher concentration  
433 in FZD samples (where the degradation was more pronounced) were detected only in  
434 decomposition soil samples during active decay of carrion (Nonanol and 2-  
435 Butyltetrahydro-furan) or in dry remains (5-Methyl-2-hexanone). 1-Hexanol,2-ethyl- was  
436 identified only during the fresh tissue breakdown, while 2-Heptanone was present during  
437 active and advanced decay, as well as in dry remains (Perrault et al., 2015). Moreover,  
438 the authors point out that sulphur-containing compounds, like DMTS in our study, are  
439 one of the most abundant compounds in decomposition profiles. They might originate  
440 from the degradation of sulphur-containing amino acids or metabolic processes during  
441 anaerobic decomposition. It is also worth notice, that the mVOCs analysis undertaken in  
442 our study did not detect the presence of aniline and 2-fluoroaniline, the volatile markers  
443 of nitroreductase activity in the wide range of cells (Thompson et al., 2020).

#### 444 *3.6. Extracellular compounds production*

445 Moreover, the performed analysis of surface tension of the cell culture supernatant and  
446 emulsification analysis revealed some modifications which may lead to increased  
447 bioavailability of low-soluble drugs molecules. The surface tension of the liquid culture  
448 after 10 days decreased slightly, from 65.02 (control) to 61.84 mN m<sup>-1</sup> and 62.22 mN m<sup>-1</sup>  
449 for NFT and FZD, respectively (Table S1). Furthermore, the emulsification indexes  
450 (EI24) of the cultures grown on NFT and FZD have increased in comparison to control  
451 cells, reaching the highest value of 7.82% when NFT was added to the batch. The decline

452 in surface tension and increase in emulsifying properties of bacterial cultures is usually  
453 associated with the production of some surface-active molecules. In this case, the  
454 observed slight changes in surface tension and an increase in emulsification index value  
455 indicates that the molecules excreted by the cells are rather bioemulsifiers, e.g.  
456 exopolysaccharide (EPS), than biosurfactants. Thanks to the high number of reactive  
457 groups on the EPS surface they stabilize the emulsion, which increases the solubility of  
458 poorly-soluble substrates and increase their bioavailability (Uzoigwe et al., 2015).

459

#### 460 **4. Conclusions**

461 The presented study thoroughly examined the bioremoval of NFT and FZD by the  
462 microbial consortium in wastewater model system. The two antibiotics affected  
463 consortium behaviour and overall organic matter metabolism in very different ways. The  
464 differences in degradation efficiency highlighted the importance of primary and  
465 alternative carbon sources, and the behaviour of a mix of the microorganisms can be better  
466 represented by the logistic mathematical model. Moreover, the overproduction of  
467 mVOCs, especially 2-ethylhexan-1-ol in the analyzed cases, might be an important  
468 marker of stress conditions. This study is a valuable contribution to the current knowledge  
469 on biodegradation of nitrofurans derivatives.

470

#### 471 **Acknowledgements**

472 The authors are grateful to Dr Agnieszka Kołodziejczak-Radzimska from Poznan  
473 University of Technology who contributed to the elemental analysis.

#### 474 **Funding**

475 This work was supported by the National Science Centre, Poland [grant No.  
476 2017/27/B/NZ9/01603].

#### 477 **CRedit authorship contribution statement**

478 **Agata Zdarta**: Conceptualization, Data curation, Formal analysis, Investigation,  
479 Visualization, Writing – original draft; **Wojciech Smulek**: Conceptualization, Data  
480 curation, Formal analysis, Investigation, Methodology, Visualization, Writing – original  
481 draft; **Zuzanna Bielan**: Formal analysis, Investigation, Visualization; **Jakub Zdarta**:  
482 Data curation, Formal analysis, Investigation, Writing – original draft; **Luong N.**  
483 **Nguyen**: Investigation, Visualization, Writing – original draft; **Agnieszka Zgoła-**  
484 **Grześkowiak**: Formal analysis, Methodology; **Long D. Nghiem**: Conceptualization,  
485 Supervision, Writing - review & editing; **Teofil Jesionowski**: Project administration,  
486 Supervision, Writing - review & editing; **Ewa Kaczorek**: Conceptualization, Funding  
487 acquisition, Project administration, Resources, ; Supervision, Writing - review & editing

#### 488 **Declaration of Competing Interest**

489 The authors declare that they have no known competing financial interests or personal  
490 relationships that could have appeared to influence the work reported in this paper.

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- 629

630 **Figure captions**

631 Fig. 1. Nitrofurantoin (NFT) and furazolidone (FZD): (A) Biological removal rates; (B)  
632 changes of drugs concentration, and (C) daily degradation rate during 10-days  
633 experiment. Samples markers: NFT supplemented (◆), and FZD supplemented (●)  
634 batches.

635 Fig. 2. Changes in: (A) biomass concentration and (E) TOC concentration of the tested  
636 systems; (B-D) kinetic study of biomass growth rate changes and (F-H) relative TOC  
637 concentration changes; samples markers: control (▲), NFT supplemented (◆), and FZD  
638 supplemented (●) batches.

639 Fig. 3. Liquid culture volatile organic compounds (A) proportions, and (B) relative VOCs  
640 content after 10 days experiment;

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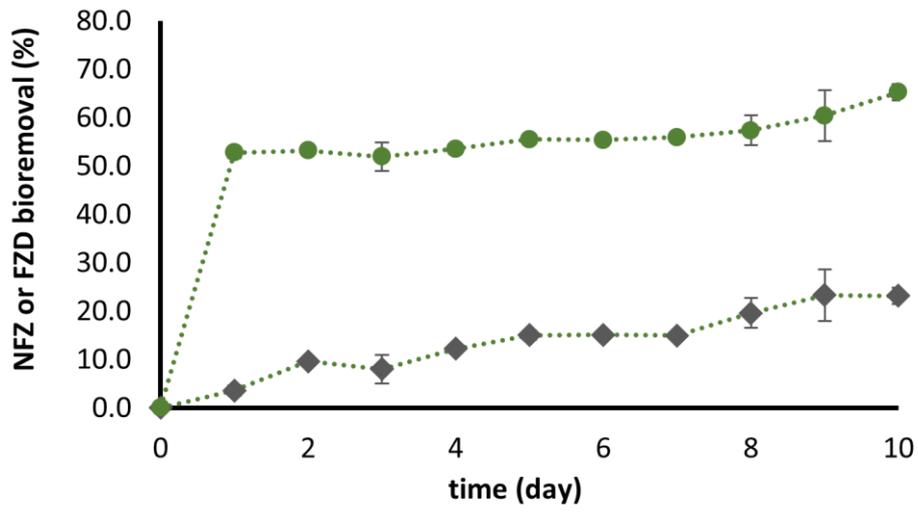
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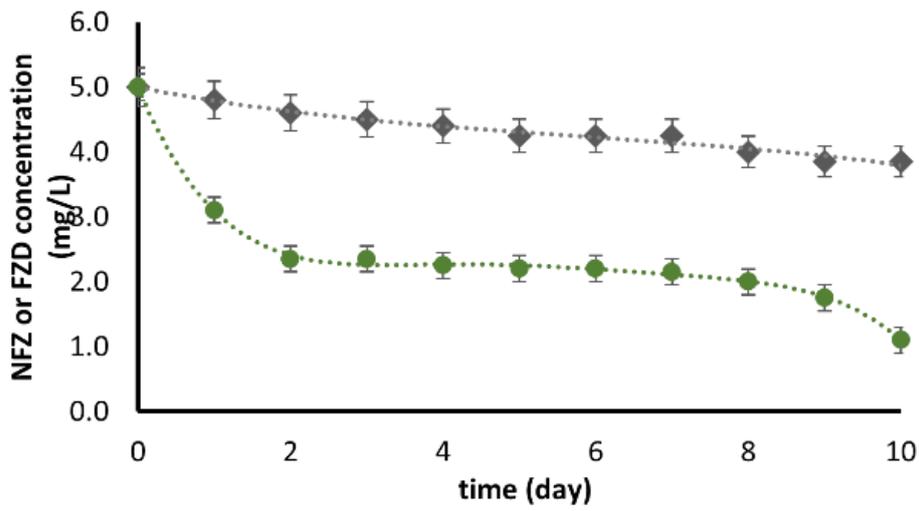
649 Fig. 1

A.

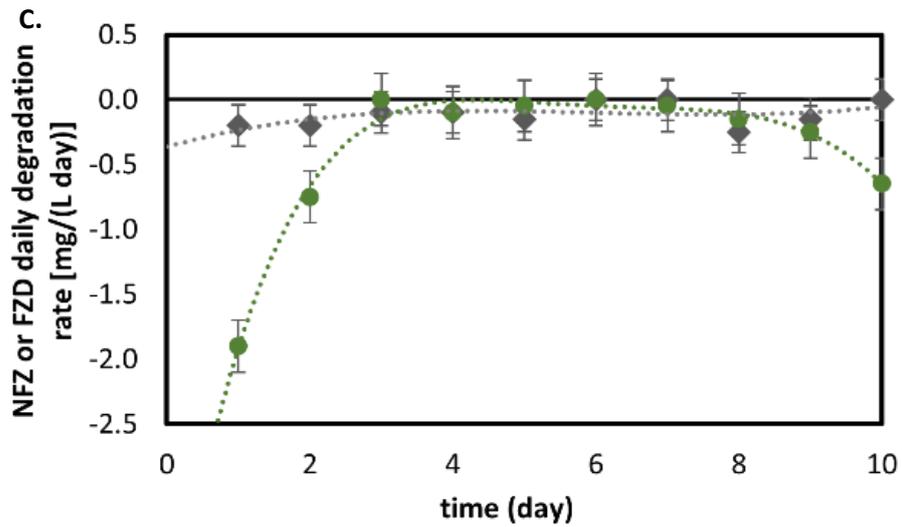


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B.

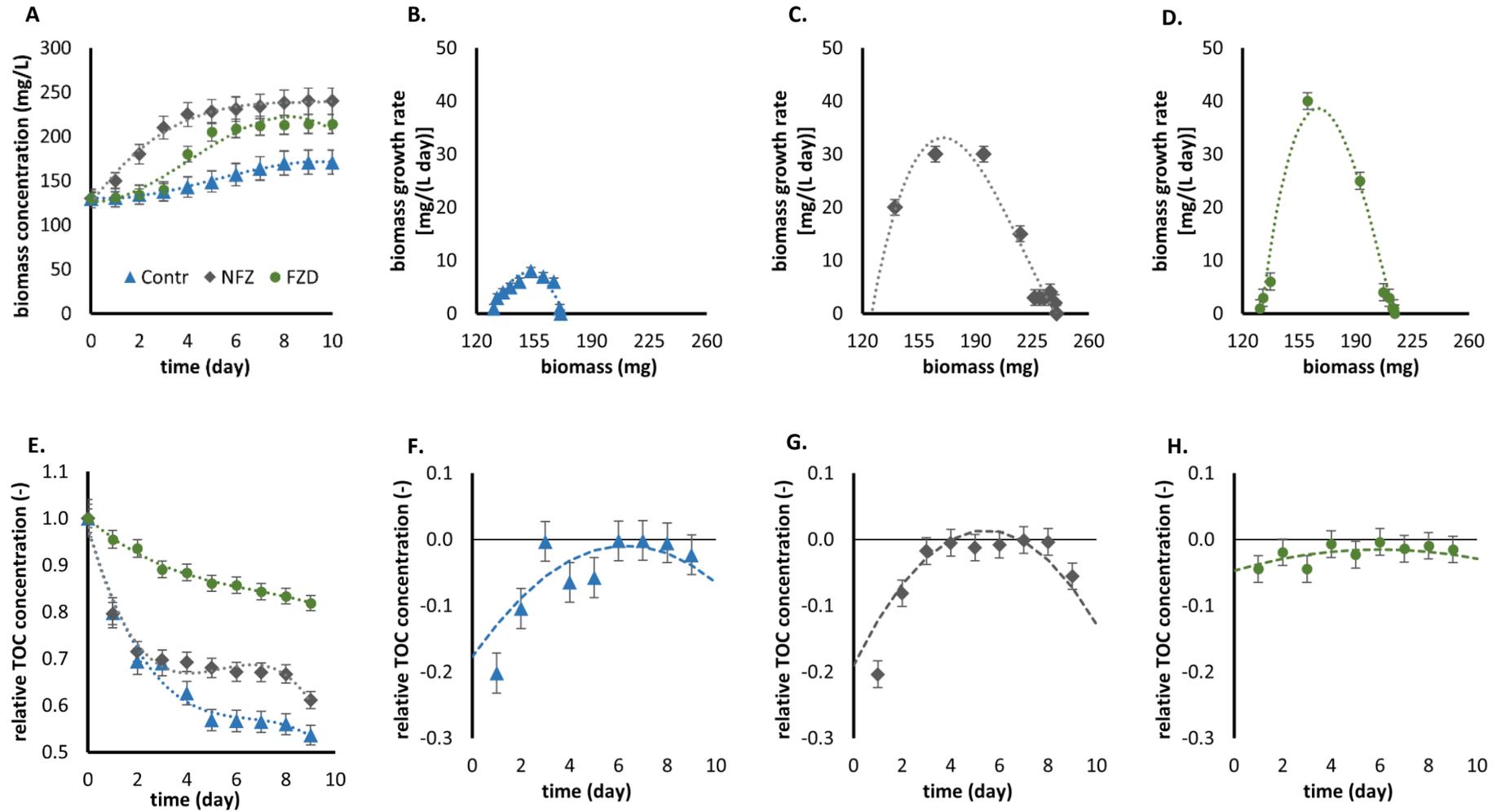


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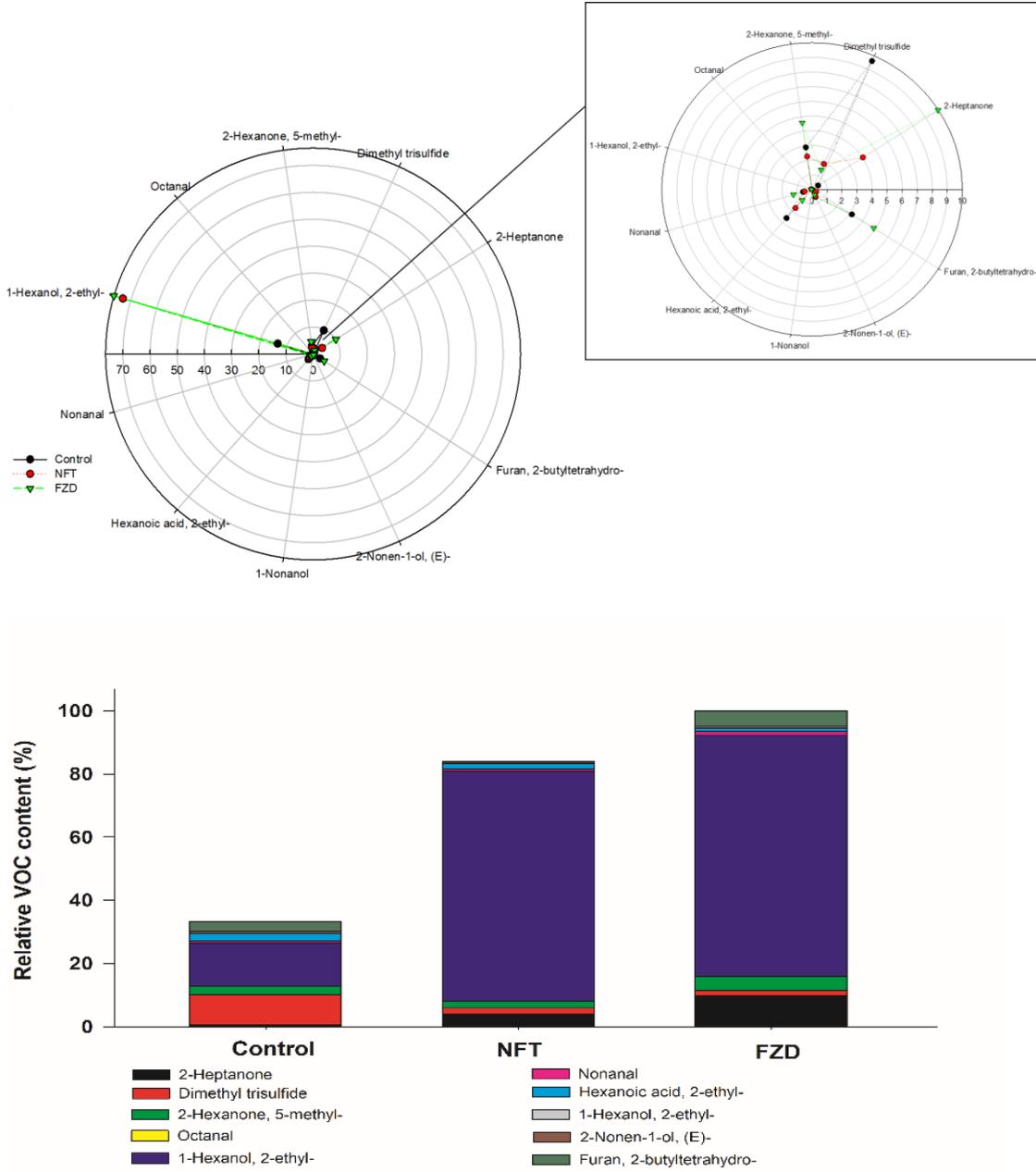
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653 **Fig. 2**



654

655 **Fig. 3**



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659 **Table 1. Substrate biodegradation models parameters**

<b>Parameter</b>	<b>NFT</b>	<b>FZD</b>
$S_{avr}$ (mg L <sup>-1</sup> day <sup>-1</sup> )	0.12	0.39
$S_{max}$ (mg L <sup>-1</sup> day <sup>-1</sup> )	0.25	1.90
$t_{max}$ (day)	1	8

660  $S_{avr}$  – average substrate degradation rate;  $S_{max}$  – maximum substrate degradation rate;

661  $t_{max}$  – the day of maximum substrate degradation rate

662

663 **Table 2. Biomass growth parameters**

	<b>Control</b>	<b>NFT</b>	<b>FZD</b>
<b>k [(mg L<sup>-1</sup>)<sup>-1</sup> day<sup>-1</sup>]</b>	4.31	2.66	7.87
<b>X<sub>max</sub> (mg L<sup>-1</sup>)</b>	303	348	342
<b>μ<sub>max</sub> (mg L<sup>-1</sup> day<sup>-1</sup>)</b>	80	30	40
<b>t<sub>max</sub> (day)</b>	6.0	2.8	4.8
<b>μ<sub>avr</sub> (mg L<sup>-1</sup> day<sup>-1</sup>)</b>	4.1	11.0	8.4
<b>μ<sub>max</sub>/ μ<sub>avr</sub> (-)</b>	19.5	2.7	4.8

664 **k** – the logistic growth rate coefficient; **X<sub>max</sub>** – the maximum biomass concentration; **μ<sub>max</sub>**  
665 – the maximum growth rate; **t<sub>max</sub>** – the day of maximum biomass growth rate; **μ<sub>avr</sub>** – the  
666 average biomass growth rate; **μ<sub>max</sub> / μ<sub>avr</sub>** – the maximum growth rate/ the average biomass  
667 growth rate ratio